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GENETICS AND PLANT BREEDING

Chemical diversity of accessions of the *in vivo* germplasm bank of *Varronia* curassavica (Jacq.)

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ABSTRACT. : Varronia curassavica (Jacq.) is a shrub and perennial plant native to Brazil that has therapeutic, anti-inflammatory and cicatrizing properties. The objective of this work was to study the chemical diversity of the native populations of this species maintained in the *in vivo* germplasm bank from Montes Claros, Minas Gerais, Brazil. The essential oils of 10 accessions analyzed in triplicate were obtained by hydrodistillation. Fifty-five compounds were detected, 46 of which were identified by GC-MS analysis. Essential oil yield ranged from 0.1 (VC-9) to 1.22% (VC-3) among the studied accessions. Three groups were formed by cluster analysis. Group A presented lower relative abundance of the compounds α-humulene (1.4%) and β-caryophyllene (8.5%). Conversely, in Group B, the most abundant compound was α-humulene (31.6%). Group C, with the greatest number of accessions, presented values of up to 6.1 and 41.2% for α-humulene and β-caryophyllene, respectively. Only two accessions (VC-1 and VC-5) presented α-humulene content lower than that recommended for phytotherapeutic production. These results, along with consolidating breeding programs, contribute to the conservation of the species.

Keywords: essential oil, medicinal plant, chemotype, α-humulene, β-caryophyllene.

Introduction

Varronia curassavica (Jacq.) [(synonym Cordia verbenaceae DC and Cordia curassavica (Jacq.) Roem. & Schult (Miller & Gottschling, 2007; Passos et al., 2007)] belongs to the family Boraginaceae and is not endemic to Brazil; however, it is observed in the phytogeographical domains of the Amazon, Caatinga, Cerrado and Atlantic Forest, with confirmed occurrences in 20 Brazilian states (Stapf, 2015).

This shrub species presents tannins, flavonoids and essential oils in the shoots (Fernandes et al., 2007). The major compounds of the essential oils are α -pinene, β -caryophyllene, β -elemene, α -



humulene, bicyclogermacrene and δ -cadinene (Carvalho, Rodrigues, Sawaya, Marques, & Shimizu, 2004). The economic and pharmacological potential of these essential oils is due to the production of the chemical markers β -caryophyllene and α -humulene (Figueira, Risterucci, Zucchi, Cavallari, & Noyer, 2009), which have spicy and woody tastes (Hampel, Monsandl, & Wüst, 2005) and which are associated with cytotoxicity and anti-proliferative, pro-apoptotic and anti-tumor activities (Parisotto et al., 2012).

Within the same population, essential oil presents differences in chemical composition due to biological activities of the plant (Eisenman, Juliani, Struwe, & Simon, 2013). This variation within the same species has been reported by several researchers (Rao, Rajput, & Mallavarapu, 2011; Karousou, Hanlidou, & Lazari, 2012; Verma, Padalia, Chauhan, & Thul, 2013).

Genetic and chemical diversities between populations are explained by geographic isolation, polymorphisms and environmental conditions (Djabou et al., 2012; Xie, Huang, Yang, & Lei, 2012). Abiotic factors, such as climate, influence secondary metabolism, especially in sesquiterpenes and monoterpenes (Feijó, Oliveira, & Costa, 2014).

Thus, conservation of species in germplasm banks can capture a large proportion of the common allelic variation present within the species of interest (Whitlock, Hipperson, Thompson, Butlin, & Burke, 2016). Therefore, the broad genetic base allows for the development of new varieties (Marfil, Hidalgo, & Masuelli, 2015), in addition to contributing to the conservation of the species. Moreover, wild plant cultivation under human selection boosts the transition to domestication by selecting genotypes adapted to human needs (Martínez-Ainsworth & Tenaillon, 2016).

The objective of this study was to analyze the chemical diversity between accessions from native populations of an *in vivo* germplasm bank of *V. curassavica* and to select plants with higher content of α -humulene and β -caryophyllene.

Material and methods

Establishment of the germplasm bank

For the collection of *Varronia curassavica* accessions, expeditions occurred between October 2011 and January 2012 in the central, north and high Jequitinhonha regions of the state of Minas Gerais and in the south region of the state of Bahia, Brazil (Figure 1). Plant material (10 accessions) was collected and propagated by cutting and remained in the greenhouse until seedling formation. Seedlings were transplanted into a planting pit (30 x 30 x 30 cm), spaced 1 x 0.5 m apart, with applications of natural phosphate and bovine manure. The collection was maintained *in vivo* under drip irrigation, and the experimental area (16° 40′ 58.5″ S, 43° 50′ 25.6″ W, at 626 m a.s.l., in Montes Claros, Minas Gerais, Brazil) was subsequently micro-sprayed.



Exsiccates of *V. curassavica* are maintained in the herbarium of ICA/UFMG. In addition, exsiccates were sent to the Institute of Biosciences, Languages, and Exact Sciences of the Sao Paulo State University (Ibilce/Unesp) for taxonomic identification by Prof. Dr. Neusa Taroda Ranga.

Extraction and analysis of essential oil

Leaves of 10 *Varronia curassavica* accessions (40 g each) were collected in triplicate and at standardized times (18 hours), since essential oil content is not influenced by collection time (Queiroz, Mendes, Silva, Fonseca, & Martins, 2016). Samples were kept in the refrigerator (8°C) until the extraction of essential oil by hydrodistillation in the Clevenger apparatus (2 hours) in a round bottom flask (1,000 mL) containing 500 mL of distilled water. At the end of the extraction process, the essential oil was collected, weighed and dried with anhydrous sodium sulfate, and it was stored in a freezer (-20°C) in amber flasks.

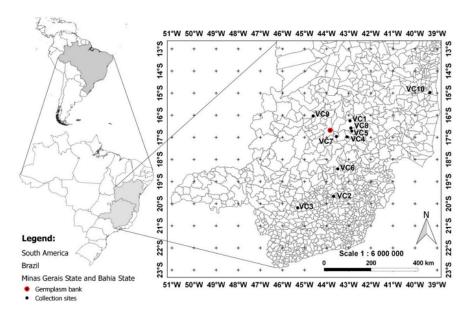


Figure 1

Location of the *Varronia curassavica in vivo* germplasm bank of the Institute of Agricultural Sciences of Federal University of Minas Gerais (ICA/UFMG). Identification of collection sites: Minas Gerais State = VC 1 - Grão Mogol (Americana Settlement), VC 2 - Taquaraçu de Minas, VC 3 - Santo Antônio do Monte, VC 4 - Itacambira, VC 5 - Cristália, VC 6 - São Gonçalo do Rio das Pedras, VC 7 - Juramento, VC 8 - Grão Mogol, VC 9 - São Francisco and Bahia State, VC 10 - São José da Vitória.

For the analysis of the chemical composition of the essential oil, samples were diluted in dichloromethane (1 mg mL $^{-1}$), transferred to vials (2 mL) and analyzed individually by gas chromatography coupled to mass spectrometry (GC-MS). The Agilent Technologies system (7890A) was coupled to a mass spectrometer (MS 5975C) with an HP-5 MS capillary column (30 m x 0.25 mm x 0.25 µm) and helium (1 mL min. $^{-1}$) as the drag gas. The injector temperature was 220°C, with a split ratio of 1:5. The oven temperature was programmed to 60 - 240°C (3°C min. $^{-1}$) for 10 min. The



interface temperature was 240°C. The MS data (total ion chromatogram, TIC) were acquired in full scan mode (m z⁻¹ of 40-350) at a scan rate of 0.3 scan/s using electron ionization (EI) with an electron energy of 70 eV. The retention index was calculated from the retention time of a mixture of n-alkanes (C7-C40, Sigma USA) at 20 ppm, with a flow rate of 1:100.

Data were analyzed using the MSD ChemStation software together with the NIST library (National Institute of Standards and Technology [NIST], 2011). The relative abundance (%) of the compounds was calculated from the peak area of the chromatogram and organized according to the order of elution. The percentage of each compound was obtained from the normalized area of each chromatogram, with mean values for each compound. The identification of β -caryophyllene and α -humulene was performed by co-injection of a synthetic standard (Sigma Aldrich), and other compounds were identified by comparison of the mass spectra with that of the NIST library. The relative retention index (RI) of each compound was calculated according to Van Den Dool and Kratz (1963) and compared with information in the literature (Adams, 2007).

The chromatograms obtained in the experiment were evaluated individually to verify normal deviations, such as contaminants, excessively broad peaks and low abundances. Afterward, chromatograms were subjected to automatic integration in the MSD ChemStation software (E.02.02.1431, Agilent Technologies), which considered all peaks with abundance greater than 0.1%. The compounds were then adjusted based on retention time, and relative abundance of each chromatographic peak was used in the chemometric analyses after calculating the logarithm and centering on the mean. The logarithms of the absolute areas were used in the construction of the statistical models in order to reduce the overvaluation of the major compounds.

Statistical analysis

All statistical analyses were carried out using the free software R Project (R Core Team, 2014). Data correspond to the mean area of total ion chromatogram obtained by the GC-MS (in triplicate) of the volatile compounds extracted by hydrodistillation. These data were subjected to Pearson's correlation analyses using the 'color' function of the 'stats' package. For the multivariate analysis, the Euclidean distance was estimated by means of the 'dist' function; the hierarchical cluster analysis (HCA) was performed by the UPGMA method (Unweighted Pair Group with Arithmetic Mean) by means of the 'hclust' function; and the principal component analysis was performed by the 'princomp' function. All of the functions belong to the 'stats' package.



Results and discussion

The yields of the essential oil obtained from the leaves of *V. curassavica* varied from 0.1 (CV-9) to 1.22% (CV-3) among the accessions. This variation in the essential oil content of plants of different origins is related to their geographical origins, which in turn is also associated with their chemical compositions (Mechergui et al., 2010).

According to the chemical analyses, 55 compounds were detected by GC-MS in the essential oil samples from the 10 V. curassavica accessions, of which 46 compounds were identified. The compounds α -pinene (2), β -caryophyllene (17), α -humulene (20), caryophyllene oxide (42), and humulene epoxide II (45) were common to all the accessions analyzed in the experiment. Table 1 shows the variation in the chemical composition of the essential oil of the accessions from V. curassavica germplasm bank, highlighting the chemical diversity within the same species.

The analysis of the chemical composition of the essential oils presented β -caryophyllene and α -humulene in all the accessions, but with variations between them. β -Caryophyllene in the accessions VC-3 (26.2 \pm 1.5%), VC-4 (27.3 \pm 2.3%), VC-6 (22.9 \pm 3.1), VC-7 (27.8 \pm 0.1), and VC-9 (41.2 \pm 0.2) presented higher area values than in the other accessions. VC-8 differed from the other accessions in relation to α -humulene (31.6 \pm 0.1%). In addition to the two major chemical compound characteristics of the species (β -caryophyllene and α -humulene), other compounds such as bicyclogermacrene (VC-1, 39.6 \pm 1.3%), bisabolol (VC-2, 33.1 \pm 28,4%), α -pinene (VC-5, 14.5 \pm 3.2%), and caryophyllene oxide (CV-10, 26.8 \pm 0.2%) also were evident in the analysis.



Table 1
Chemical composition of essential oil extracted from the leaves of *Varronia curassavica* from the germplasm bank of the Institute of Agricultural Sciences of UFMG (ICA-UFMG).

No.	N°	Compounds	RT RI ¹ RI ² % area (TIC) of the accession in the germplasm bank
September S. 5.99 \$2	IN	Compounds	VC-1 VC-2 VC-3 VC-4 VC-5 VC-6 VC-7 VC-8 VC-9 VC-10 TI
Sabinene 6.9 75 972	01	α-thuje ne	
Sabinene 6.9 75 972	02	α-pinene	5.5 939 932 - 0.1 ± 0.0 5.0 $\pm 0.67.4 \pm 1.5$ $\frac{14.5}{3.2}$ 6.2 $\pm 1.11.1 \pm 0.02.3 \pm 0.00.5 \pm 0.00.4 \pm 0.0$ a,b
50 β-myrcene 70 99 99 8.8 = 6.03 = 10.00 = 0.27 = 0.41 = 0.0 0.35 = 0.0	03	sabinene	
Prycymene 2,100s1004 0.21-0.51 2.0.5 0.82-1.4 0. 0.35-0.0.3	04	_	6.7 979 978 - 0.2 ± 0.1 - 3.4 ± 0.64.7 ± 1.0 5.9 ± 2.83.2 ± 0.0 1.2 ± 0.0 1.3 ± 0.0 0.2 ± 0.0 a,b
			10 11 10 10 10 10 10 10 10 10 10 10 10 1
Sequest Compared Montergener			
Section Sec	07	Ilmonene	
Sepainterprints	08	eucalyptol	
S-elemene	09	γ-terpinene	9.3 10621056 0.2 ± 0.2 - 7.0 ± 2.1 a,b
12 α-cubebene 21215511545 - 0.11-0.11 0.01-0.01 0.1 - 0.02-0.01 0.3			
12			•
15 β-bourbonen 227;180719/14-01 19-12 13-12			11.6+
β-cubebene 22,91301384 0.09=11.01=0009=0.103=0001=0.1 2.22=0.0 at		•	22.4137/13/22.6=0.0 1.7=0.9 0.9=0.91.0=0.13.4=0.21.8=0.50.6=0.0 - 0.5=0.5 0.3 a,b
Selemene 25 1911 186			
16			481+
P-caryophyllene		•	23 13911386 0.4 1.5 ± 0.8 1.5 ± 0.83.7 ± 0.42.4 ± 0.11.8 ± 0.01.1 ± 0.13.2 ± 0.12.1 ± 0.01.5 ± 0.0 a,b
17	16	α-cedrene	
P-famesene	17	β-caryophylle ne	24 2141014160 5 + 0.0
25 25 25 25 25 25 25 25			24.6143214250.2 ± 0.0 0.5 ± 0.4 0.4 ± 0.3 0.4 ± 0.10.3 ± 0.1 0.4 ± 0.10.3 ± 0.0 0.8 ± 0.0 0.2 ± 0.0 1.5 ± 0.0 a,b
21 alloaromadendrene 25.8146014552.7 = 00 1.8 = 2.9 0.2 = 0.10.8 = 0.01.2 = 0.2.5 = 5.3		•	•
25			0.1
25			
24			26.5147714711.1 ± 0.0 0.1 ± 0.0 0.6 ± 0.30.1 ± 0.02.0 ± 0.20.5 ± 0.20.7 ± 0.00.1 ± 0.03.0 ± 0.00.9 ± 0.0 a,b
S-selinene		·	0.5 4.6
26			
Calasken 27.514981491 - 48 = 0.7 5.9 = 6.5 1.1 4.6 = 1.4 6.1 = 0.224 = 0.05.4 = 0.01.5 = 0.0 = 0.			
Bicyclogermacrene		•	16.2 ±
Sesquicine 17415001494 13			1.1
Sesquicineol			27.415001494 0.9 ± 0.9 0.4 ± 0.20.2 ± 0.00.4 ± 0.00.5 ± 0.01.0 ± 0.0
Cubebol 28.1151815110.6 = 0.0 - 0.5 = 0.40.4 = 0.0 4.7 = 0.3 0.6 = 0.22.5 = 0.00.9 = 0.1.1 = 0.01.9 = 0.0 1.3		•	
Secadine			•
33			
1,4,7, -cyclododecatriene,1,5,9,9 tetramethyl 29,115671556 1,9 = 0.6 0.2 = 0.2 - 0 - 0.2 = 0.0 - 0.6 = 0.0 a)			·
Sesquisabine ne hydrate 29.1156/1536 19.9=0.6.02 = 0.2 - 0.02 = 0.0 - 0.6 = 0.0 a.]	34		$28.7153515280.1\pm0.0 -\qquad -\qquad -\qquad -\qquad -\qquad 0.8\pm0.1\qquad -\qquad -\qquad -\qquad a,b$
Sequisabinene hydrate 29.215441541 12 = 0.4.0.1 = 0.2 - 0.5 = 0.0 - 0.2 = 0.0 a]	35		29.115671536 - 1.9 ± 0.6 0.2 ± 0.2 0.2 ± 0.0 0.6 ± 0.0 a,b
36		tetrametnyi	Oxygenated Sesauiterpenes
38	36	sesquisabinene hydrate	
39	37	unknown	$29.4 \ \ - \ \ 1545 \ \ \ - \ \ \ 0.1 \pm 0.1 \ 0.2 \pm 0.1 \ 0.1 \pm 0.0 \ - \ 0.1 \pm 0.00.6 \pm 0.0 \ - \ 0.1 \pm 0.11.4 \pm 0.0 \ - \ 0.1 \pm 0.00.6 \pm 0.0 \ - \ 0.1 \pm 0.00.4 \pm 0.0 \ - \ 0.0 \pm 0.00.4 \pm 0.0 \ - \ 0.0 \pm 0.00.4 \pm 0.00.4 \pm 0.00.4 \ - \ 0.0 \pm 0.00.4 \pm 0.00.4 \ - \ 0.0 \pm 0.00.4 \ - \ 0.0 \pm 0.00.4 \ - \ 0.0 \pm 0.00.4 \ - \ 0.0 \pm $
40 unknown 50.3 1567 - 0.1±0.1 5.5±3.2 - 2.5±0.2 1.5±0.50.2±0.0 - 0.1±0.0 0.2±0.0 - 41 spathule nol 50.415781571 12.5± 0.9±0.4 1.8±2.1 3.4±1.3 2.6±0.8 2.5±0.85.5±0.1 5.8±0.1 2.6±0.0 3.0±0.1 a.] 42 caryophylle ne oxide 50.6158515770.4±0.1 2.0±1.6 5.4±1.1 1.9±0.7 1.9±0.5 2.7±1.1 10.2± 7.0±0.1 9.8±0.1 26.8± a.] 43 unknown 50.8 15802.9±0.1 0.5±0.1 0.9±0.51.7±0.20.7±0.1 0.6±0.51.5±0.00.5±0.0 - 1.0±0.0 ±0.0 4.4 epiglobulol 50.915851585 - 0.2±0.1 0.5±0.1 0.9±0.5 2.7±1.1 10.2± 7.0±0.0 ±0.0 1.0±0.0 ±0.0 1.0±0.0 ±0.0 1.0±0.0 5.0±0.0 ±0.0 ±0.0 1.0±0.0 5.0±0.0 ±0.0 ±0.0 ±0.0 ±0.0 ±0.0			
Spathule nol 30.415781571 12.5 ± 0.9 ± 0.4 1.8 ± 2.13.4 ± 1.32.6 ± 0.8 2.5 ± 0.85.3 ± 0.15.8 ± 0.12.6 ± 0.0 5.0 ± 01 al al 42 caryophylle ne oxide 30.6158315770.4 ± 0.1 2.0 ± 1.6 5.4 ± 1.11.9 ± 0.71.9 ± 0.5 2.7 ± 1.1 10.2 ± 7.0 ± 0.1 ± 0.8 ± 0.1 26.8 ± al 2.4 ± 0.0 ±			
caryophyllene oxide 30.6158315770.4 ± 0.1 20 ± 1.6 5.4 ± 1.11.9 ± 0.7 1.9 ± 0.5 2.7 ± 1.1 10.2 ± 7.0 ± 0.1 9.8 ± 0.1 26.8 ± a.] unknown 30.8 15802.9 ± 0.1 0.3 ± 0.1 0.9 ± 0.61.7 ± 0.2 0.7 ± 0.1 0.6 ± 0.51.3 ± 0.0 0.5 ± 0.0 − 1.0 ± 0.0 − 4.4 ± piglobulol 30.915851585 − 0.2 ± 0.1 0.3 ± 0.1 − 0.2 ± 0.1 0.5 ± 0.1 1.0 ± 0.00.5 ± 0.0 − 1.0 ± 0.0 − 4.4 ± 0.2 a.] humulene epoxide II 51.7160916030.5 ± 0.0 0.7 ± 0.0 0.5 ± 0.1 0.6 ± 0.20.4 ± 0.1 0.5 ± 0.11.0 ± 0.00.9 ± 0.0 1.0 ± 0.05.6 ± 0.1 a.] humulene -1,6-dien-3-ol 32.016191612 − 0.3 ± 0.1 0.3 ± 0.1 0.5 ± 0.1 1.5 ± 0.1 3.1 ± 1.20.5 ± 0.00.9 ± 0.0 0.5 ± 0.0 ± 0.1 a.] 48 unknown 32.2 − 1617 − − 0.1 ± 0.10.2 ± 0.00.2 ± 0.3 1.9 ± 0.80.1 ± 0.00.3 ± 0.0 0.5 ± 0.0 1.a.] 49 unknown 32.3 − 16211.1 ± 0.0 0.5 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 1.5 ± 1.5 0.3 ± 0.10.7 ± 0.00.4 ± 0.00.3 ± 0.00.4 ± 0.0			12.5+
42 caryophylene oxide 30.8 15802.9 ±0.1 0.3 ±0.1 0.9 ±0.61.7 ±0.20.7 ±0.1 0.6 ±0.51.5 ±0.00.5 ±0.0		•	30.413/813/1 1.1 0.9=0.4 1.8 = 2.13.4 = 1.32.6 = 0.82.3 = 0.83.3 = 0.13.8 = 0.12.6 = 0.0 3.0 = 01 a,0
## epiglobulol 30,915851583 0.2 ± 0.1 0.3 ± 0.1 - 0.2 ± 0.1 0.5 ± 0.1 0.1 ± 0.03.1 ± 0.1 - 0.4 ± 0.2 a.] ### humulene epoxide II 31,7160916030.5 ± 0.0 0.7 ± 0.2 0.5 ± 0.1 0.6 ± 0.2 0.4 ± 0.1 0.5 ± 0.11.0 ± 0.00.9 ± 0.0 1.0 ± 0.05.6 ± 0.1 a.] ### selina-6-en-4-ol 32.2 1617 - 0.1 ± 0.1 0.3 ± 0.1 0.3 ± 0.1 0.3 ± 0.1 0.3 ± 0.1 0.3 ± 0.0 0.5 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 0.2 ± 0.0 1.9 ± 0.0 0.3 ± 0.0 0.5 ± 0.0 1.3 ± 0.0 0.2 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 0.2 ± 0.0 1.5 ± 0.0 0.2 ± 0.0 1.9 ± 0.0 0.3 ± 0.0 0.3 ± 0.0 0.5 ± 0.0 1.3 ± 0.0 0.2 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 1.5 ± 1.5 0.2 ± 0.0 1.5 ± 0.0 0.2 ± 0.0 1.5 ± 0.0 0.2 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.5 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.8 ± 0.0 1.2 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.8 ± 0.0 0.5 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.8 ± 0.0 0.5 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.8 ± 0.0 0.5 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.8 ± 0.0 0.5 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.4 ± 0.0 0.3 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.3 ± 0.0 1.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0.2 ± 0.0 0.5 ± 0.0 0			30.6158515770.4±0.1 2.0±1.6 3.4±1.11.9±0.71.9±0.5 2.7±1.1 0.0 7.0±0.19.8±0.1 0.2 a,0
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		Total	100 100 100 100 100 100 100 100 100 100



IT: identification type; (-): compound not detected; a: identification based on the retention index, b: identification based on the mass spectrum (70 eV), c: substance identified by the co-injection of the commercial synthetic standard; RT: retention time in minutes; RI1: retention index of the literature, RI2: retention index calculated. Mean values of the area of the total ion chromatogram (TIC), followed by the respective standard deviation (±).

Such behavior can be explained by changes in secondary metabolism, which favors the biosynthesis of volatile compounds due to the induction of specific enzymes (Venditti et al., 2015). Although essential oil biosynthesis is determined by genetics, the content and proportion of chemical constituents of essential oils are intensely affected by geographical origin (Santos, Pinto, Santos, Cruz, & Arrigoni-Blank, 2016).

Five native populations of V. curassavica from the northeast of Brazil present differences in the relative abundances of chemical compounds of the essential oil (Farhat, Jordán, Chaouch-Hamada, Landoulsi, & Sotomayor, 2016). Differences in β -caryophyllene (1.67 to 22.25%) and α -humulene (0.00 to 10.32%) were found in this research and are the result of the influence of genetic factors and environmental conditions. These findings occur due to polymorphisms and geographic locations, which are responsible for the chemical diversity of the essential oil, since it is associated with adaptations as a survival strategy.

Although climate, soil orders (Rahimmalek, Heidari, Ehtemam, & Mohammadi, 2017) and water availability (Maatallah, Nasri, Hajlaoui, Albuch, & Elaissi, 2016) alter the secondary metabolism of plants, the effects of genetic factors on essential oil composition are greater than the effects of environmental factors (Moghaddam & Farhadi, 2015). In this study, the chemical composition of the essential oil of leaves of different *V. curassavica* genotypes subjected to the same culture conditions was analyzed. Researchers have evaluated the content and chemical composition of twenty-one accessions of *Acorus calamus* from different geographic origins grown under the same environmental conditions and have also noted a variation in the number of compounds present in the essential oil (Kumar et al., 2016).

Figure 2 shows the correlation for the 15 most abundant compounds in all accessions; green and red lines are positive and negative correlations, respectively, and their correlation intensity is expressed by the thickness of the line that connects each compound. Thus, copaene (C10), bicyclogermacrene (C11), δ -cadinene (C04) and α -humulene (C09) are strongly associated, which may be a consequence of a common metabolic pathway.

The dendrogram of Figure 3 classifies the essential oils of 10 *Varronia curassavica* populations based on the Euclidian distance, presenting the groups similar to each other in relation to the chemical compositions of the essential oil. The hierarchical cluster analysis revealed three principal groups, classified as A, B, and C. Group A was formed by accession VC-1; group B was formed by the accession VC-8; and group C was formed by the other accessions (VC-2, VC-3, VC-4, VC-5, VC-6, VC-7, VC-9, VC-10).

The identification of several chemotypes in a single species is valuable for the commercial propagation and cultivation for domestication and



breeding programs, since it allows for identification of divergent groups by the chemical composition (Fattahi, Nazeri, Kalantari, Bonfil, & Fattahi, 2016). For this species, five groups were characterized with the following major compounds: turmerone (8.96-30.15%) and βcaryophyllene (4.6-20.17%) for group 1; tricyclene (22.20-35.95%) and camphene (16.62-27.38%) for group 2; α-azingiberene (24.81-35.83%) and β-sesquiphellandrene for group 3; E-caryophyllene (3.90-31.06%) and 7-cyclodecen-1-ona,7-methyl-3-methylene-10-(1-propyl) for group 4; and finally 7-cyclodecen-1-ona,7-methyl-3-methylene-10-(1-propil) (24.68-50.20%) as the major compound for group 5 (Nízio et al., 2015). These differences are possibly related to survival strategies for the different cultivation conditions, since the studied genotypes are in constant evolution. The state of Sergipe is similar to the north of the state of Minas Gerais; however, the latter is a complex region, due to regional topography and maximum altitudes between 700 and 800 m asl (Alvares, Sentelhas, Gonçalves, & Sparovek, 2013).

The principal component analysis notes the most important characteristics in the data set. This technique revealed that four principal components explain 82.58% of the variation in the chemical composition of the essential oil from V. curassavica (Table 2). The first principal component (PC1) explained 28.71% of the variation. The compounds of this group presented significant positive correlation for α -humulene (0.79) and negative correlation for copaene (-0.92) and bicyclogermacrene (-0.95). The second principal component (PC2) explained 24.45% of the variation among the studied accessions, with positive correlation for spathulenol (0.77) and negative correlation for α -pinene (-0.66), terpinene (-0.66), cymene (-0.8), and α -alaskene (-0.63).

The distribution of the 10 *V. curassavica* accessions analyzed is presented in Figure 4. PCA confirmed the relationships between the accessions; the interrelationship between the chemical compounds grouped those that presented similarities. This figure shows the same tendency of dendrogram grouping, since VC-1 and VC-8 accessions formed groups isolated from the other groups. In addition, VC-3 and VC-10 are almost overlapping, indicating that they are quite similar. The arrows indicate the compounds with the highest correlation with PC1 and PC2, which are the most important components in the study.

Table 3 shows the similarity between the three groups formed by the dendrogram. *V. curassavica* accessions were highlighted in Group A for 6 chemical compounds: muurolene (C01), α-pinene (C02), β-caryophyllene (C03), δ-cadinene (C04), cymene (C06), and alloaromadendrene (C07). Group B presented α-pinene (C02), caryophyllene (C03) and δ-cadinene (C04), with absence of bicyclogermacrene (C11). Group C had α-pinene (C02), β-caryophyllene (C03), and δ-cadinene (C04), while α-humulene (C9) and myrcene (C14) were absent.

Monoterpenes, diterpenes and sesquiterpenes are produced by terpene synthases (TPs) that use geranyl diphosphate (GDP) farnesyl diphosphate (FDP) and geranylgeranyl diphosphate (GGDP),



respectively (Cheng et al., 2007). The chemical diversity among these compounds may be related to the ability of a single TP protein to form several terpenes from a single substrate. Furthermore, these authors note that the formation of several products could be a consequence of the greater conformational flexibility in the enzyme's active site, allowing the formation of reaction intermediates (Degenhardt, Jöllner, & Gershenzon, 2009).

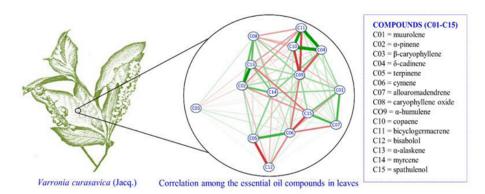


Figure 2
Diagram of Pearson's correlation coefficients matrix among the fifteen major essential oil compounds (C01-C15) of the analyzed *Varronia curassavica* populations.

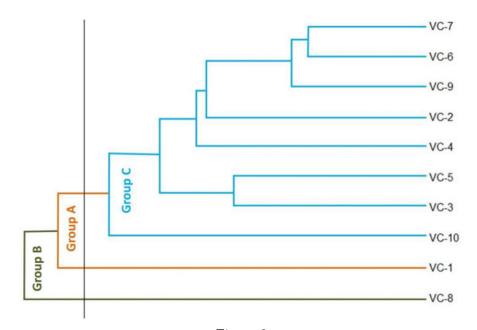


Figure 3 Similarity b

Dendrogram representing the similarity between 10 *Varronia curassavica* accessions based on the chemical composition of the essential oils obtained by hydrodistillation and analyzed by gas chromatography coupled to mass spectrometry.



Table 2

Correlation between component loadings for the first four principal components and 15 major essential oil compounds (C01-C15) of the analyzed *Varronia curassavica* populations.

Compound	N°	Principal component			
Compound		PCA1	PCA2	PCA3	PCA4
muuro le ne	C01	-0.22ns	0.45 ^{ns}	0.70*	0.13 ^{ns}
α-pinene	C02	0.47ns	-0.66*	0.55ns	-0.13ns
β-caryophylle ne	C03	-0.01 ^{ns}	-0.03 ^{ns}	-0.15 ^{ns}	0.42ns
δ-cadine ne	C04	-0.91*	-0.06ns	0.33ns	-0.18ns
terpine ne	C05	-0.27 ^{ns}	-0.66*	0.21 ^{ns}	0.62ns
cymene	C06	-0.32ns	-0.80*	-0.29ns	0.21ns
alloaro made ndre ne	C07	-0.21ns	0.59ns	0.56ns	0.16 ^{ns}
caryophylle ne oxide	C08	0.59ns	-0.26 ^{ns}	0.33ns	-0.24ns
α -humule ne	C09	0.79*	0.33ns	0.39ns	0.14 ^{ns}
copaene	C10	-0.92*	-0.17ns	0.19ns	-0.3ns
bicyclogermacre ne	C11	-0.95*	0.06 ^{ns}	0.15 ^{ns}	-0.23 ^{ns}
bisabolol	C12	0.19ns	0.55ns	-0.13ns	-0.62ns
α-alasken	C13	0.48ns	-0.63*	0.49ns	-0.29ns
myrcene	C14	-0.13 ^{ns}	-0.32ns	0.88*	-0.03 ^{ns}
spathule no l	C15	-0.01 ^{ns}	0.77*	0.35 ^{ns}	0.45 ^{ns}
Eigenvalue		4.31	3.67	2.83	1.58
% of Variance		28.71	24.45	18.86	10.56

Significance is indicated as follows: *significance at 5% nominal level; **significance at 1% nominal level; nsnot significant by the t-test.

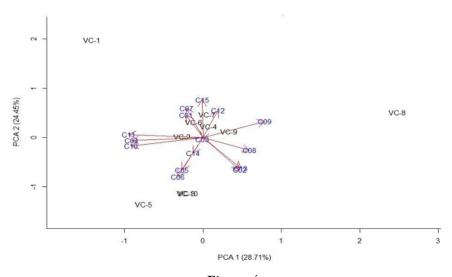


Figure 4

Distribution of the chemical components of the essential oil of the *V. curassavica* germplasm in relation to the two principal components by the principal component analysis (PCA).



Table 3
Content of the 15 principal components of essential oil (C01-C15) for the three groups (A, B, C) found in 10 *Varronia curassavica* accessions.

	N° -	Content [%]			
Compound	IN -	Α	В	С	
muurole ne	C01	8.74 ± 1.69	8.63 ± 0.69	8.44 ± 0.35	
α-pinene	C02	8.16 ± 3.43	9.36 ± 0.83	9.32 ± 0.39	
β-caryophylle ne	C03	8.16 ± 3.43	9.36 ± 0.83	9.32 ± 0.39	
δ-cadine ne	C04	8.16 ± 3.43	9.36 ± 0.83	9.32 ± 0.39	
terpine ne	C05	6.68 ± 3.2	8.12 ± 0.69	4.74 ± 4.11	
cymeno	C06	8.06 ± 1.3	7.61 ± 1.44	7.17 ± 0.3	
alloaro made ndre ne	C07	8.06 ± 1.3	7.61 ± 1.44	7.17 ± 0.3	
caryophylle ne oxide	C08	7.1 ± 3.47	8.68 ± 0.52	8.16 ± 0.34	
α-humule ne	C09	2.72 ± 3.59	6.07 ± 1.92	0 ± 0.00	
copaene	C10	7.1 ± 3.47	8.68 ± 0.52	8.16 ± 0.34	
bicyc logermacre ne	C11	4.57 ± 3.64	0 ± 0.00	7.93 ± 0.33	
bisabolol	C12	5.89 ± 3.27	8.98 ± 0.9	7.09 ± 0.41	
α-alasken	C13	7.1 ± 3.47	8.68 ± 0.52	8.16 ± 0.34	
myrceno	C14	2.72 ± 3.59	6.07 ± 1.92	0 ± 0.00	
spathule no l	C15	5.89 ± 3.27	8.98 ± 0.9	7.09 ± 0.41	

Contents of the main essential oil components are represented as the means \pm standard deviations.

The biosynthesis of isoprenoids (terpenoids) is important for the production of precursors, such as triose phosphate and pyruvate, which may have favored methylerritol phosphate (MEP) to assume functions of both the photosynthetic tissues and secondary metabolism under stress conditions (Hemmerlin, Harwood, & Bach, 2012). Little information is found in genomic databases regarding the specific metabolic pathways of the synthesis of these compounds in trichomes and the genes involved in their developmental stages (Tiwari, 2016).

In general, metabolic pathways control the production of compounds synthesized for cell growth and survival, as well as interactions with the environment, including responses to biotic and abiotic factors. The reduction or absence of the major compounds of interest (β -caryophyllene/ α -humulene) is an example of such a situation. The significant decline in β -caryophyllene expression and α -humulene synthesis activity is significantly influenced by heat stress conditions (Pazouki et al., 2016).

Conclusion

The present study reveals chemical diversity in the composition of the essential oil obtained from leaves of the *in vivo* germplasm bank of *V. curassavica*. These differences enabled the separation of the accessions into three groups, where the biosynthesis of compounds influenced the relative abundance of the essential oil compounds. Knowledge of the chemical diversity of *V. curassavica* plants in Minas Gerais State and neighboring regions allow for breeding programs enhancement and generation of pharmaceutical products, with consequent improvement in the raw material production chain.



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