



Boletín de Geología

ISSN: 0120-0283

ISSN: 2145-8553

Universidad Industrial de Santander

Portilla-Mendoza, Karina Andrea; Pinzón-Núñez, Diego Armando; Moreno-González, Leonardo; Mier-Umaña, Ricardo; Ríos-Reyes, Carlos Alberto; Henao-Martínez, Jose Antonio  
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Boletín de Geología, vol. 41, no. 2, 2019, May-August, pp. 123-136  
Universidad Industrial de Santander

DOI: 10.18273/revbol.v41n2-2019007

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## Mineralogical characterization of pre-hispanic pottery at the Mesa de Los Santos region, Colombia

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DOI: <http://dx.doi.org/10.18273/revbol.v41n2-2019007>



**Forma de citar:** Portilla-Mendoza, K.A., Pinzón-Núñez, D.A., Moreno-González, L., Mier-Umaña, R., Ríos-Reyes, C.A., and Henao-Martínez, J.A. (2019). Mineralogical characterization of pre-hispanic pottery at the Mesa de Los Santos region, Colombia. *Boletín de Geología*, 41(2), 123-136. DOI: 10.18273/revbol.v41n2-2019007

### ABSTRACT

The application of mineralogy in archeology has been growing in interest in particular in relation to establishing the origin of the clays used in the elaboration of ancient ceramic artifacts. The mineralogy did not show a clear relationship with the colors expressed by the pottery system defined by the archaeologists; that is, independent of the color of the slip, each group has the same mineralogical components. Fourteen Pre-Hispanic pottery sherds from the Mesa de Los Santos region (Colombia) were selected for mineralogical characterization by X-ray diffraction, scanning electron microscopy and Fourier transform infrared spectroscopy. The objective of the work was to contribute to the knowledge of the production technology and elucidate a possible regional origin of these ceramics. The observed mineralogical phases were plagioclase, quartz, potassium feldspar, clay minerals, micas, carbonate minerals, and iron oxides. Firing temperatures are in the ranges 600-800°C for almost all the analyzed samples.

**Keywords:** Pottery; sherds; mineralogy; firing temperatures; Mesa de Los Santos region.

### Caracterización mineralógica de la cerámica prehispánica de la región de La Mesa de Los Santos, Colombia

### RESUMEN

La aplicación de la mineralogía en arqueología ha venido creciendo en interés en particular con relación a establecer la procedencia de las arcillas utilizadas en la elaboración de artefactos cerámicos antiguos. La mineralogía no mostró una relación clara con los colores expresados por el sistema cerámico definidos por los arqueólogos; es decir, independiente del color del engobe, cada grupo tiene los mismos componentes mineralógicos. Se seleccionaron catorce piezas de alfarería prehispánica de la región de la Mesa de Los Santos (Colombia) para la caracterización mineralógica por difracción de rayos X, microscopía electrónica de barrido y espectroscopia infrarroja de transformada de Fourier. El objetivo del trabajo fue contribuir al conocimiento de la tecnología de producción y dilucidar una posible procedencia regional de estas cerámicas. Las fases mineralógicas observadas fueron plagioclasa, cuarzo, feldespato potásico, minerales de arcilla, micas, minerales de carbonato y óxidos de hierro. Las temperaturas de cocción están en los rangos 600-800°C para casi todas las muestras analizadas.

**Palabras clave:** Piezas; cerámicas; mineralogía; temperaturas de cocción; región de la Mesa de Los Santos.

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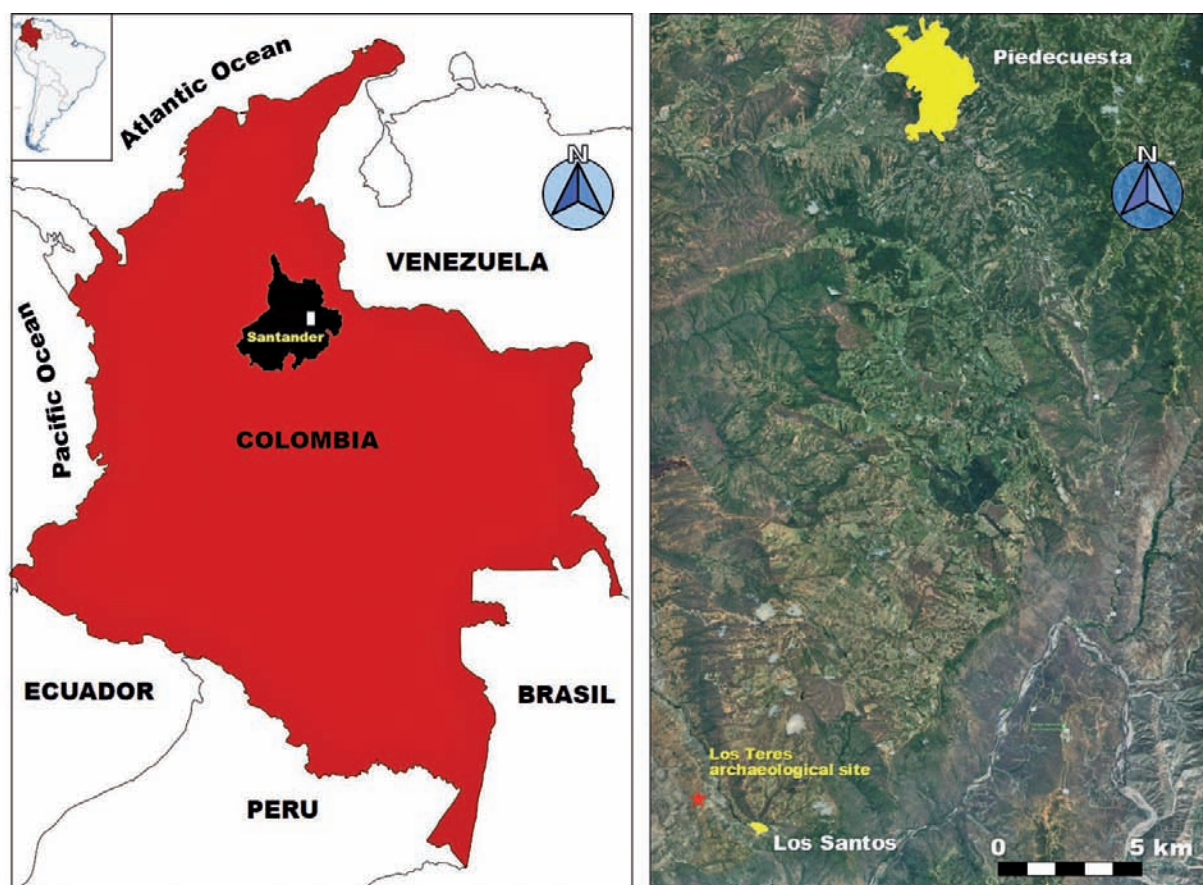
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## INTRODUCTION

Several analytical techniques, such as X-ray fluorescence spectroscopy, X-ray diffraction, scanning electron microscopy, electron microprobe analysis, infrared and Raman spectroscopy, high-resolution X-ray microtomography and gas chromatography–mass spectroscopy have been used in the mineralogical characterization of pottery sherds (*e.g.*, Bong *et al.*, 2010; Kahl and Ramminger, 2012; Grifa *et al.*, 2013; Damjanović *et al.*, 2014; Tarquini *et al.*, 2014; Lettieri, 2015; Noll and Heimann, 2016). However, the selection of the suitable analysis methods should be based on the type, form and amount of samples to be analyzed (Lettieri, 2015). Archaeological analysis of pre-Hispanic ceramic technology and its socio-economic, cultural and historical scope, contrasted with data from geosciences, in addition to its documentation value (*e.g.*, Belfiore *et al.*, 2007; Barrios-Neira *et al.*, 2009; Iordanidis *et al.*, 2009) allows qualifying archaeological inferences based on the understanding

of the attributes of the raw materials used for the manufacture of ceramic artifacts (*e.g.*, Kramar *et al.*, 2012; Moroni and Conti, 2006; Ravisankar *et al.*, 2014) and the technology involved in the manufacturing process (*e.g.*, Schwedt and Mommsen, 2004; Mangone *et al.*, 2009; Belfiore *et al.*, 2010), which may reveal significant information about the provenance of the raw materials with which the ceramic artifacts were manufactured (*e.g.*, Sherriff *et al.*, 2002; Barone *et al.*, 2005). The presence or absence of certain mineral phases provides information regarding the firing conditions in terms of both temperature and atmosphere used during pottery production (Iordanidis *et al.*, 2009). The aim of this work was to carry out the mineralogical characterization of archaeological pre-Hispanic pottery sherds from the Mesa de Los Santos region, Colombia (FIGURE 1), representing the beginning of a long-term research developing provenance studies based on the archeometry of ceramic pottery, which will provide new interesting and useful information on pre-Hispanic societies.



**FIGURE 1.** Geographical location of Los Teres archaeological site indicated by the red star. Adapted and modified from Google Earth.

## ARCHAEOLOGICAL ASPECTS

The Guane pre-Hispanic society, between the eleventh century and the arrival of the Spanish conquerors (sixteenth century) occupied the mountainous area of the Department of Santander, Eastern Cordillera of the Colombian Andes. Its social development, reached the level of chiefdom society, characterized by social complexity and stratified social organization, its responses to irrigation ditches and channels for agricultural production, its fluid trade with other ethnic groups such as Muisca, Chitaneos and Yareguies and with its form of autonomous government in each zone and, only a centralized government in case of external aggression, according to the sources of sixteenth-century chroniclers. The social organization and form of government of this ancient culture had contents of social differentiation and in the political leadership it was expressed in the autonomy of each territory governed by a local cacique, which allowed an integration and control of the territories (small regional autonomies), under the influence of a local commercial network for the supply of basic consumer goods and interregional trade to obtain sumptuous goods for the groups of power, as recorded in the sources of sixteenth-century chroniclers. The material expressions of the Guane society are varied: small towns, cemeteries in hills and caves that were formed by the calcareous geology of the region and that in the big canyons left holes where the most prestigious people were buried and, that in some cases several of them were subjected to mummification and ritual complexes, whose relatives crossed the difficult topography, the access by the cornice of deep precipices and the rich funeral trousseau that accompanies the deceased, including the ceramic vessels and whose contained foods, symbolically express the step to another life. Among the evidence of Guane material culture, the pottery stands out for the presence of muscovite mica, the diversity of forms (globular pots with handles, bowls, glasses, large bottles etc.) and, for the use of painting as a decorative technique (polychromy) and the decoration with complex geometric designs (especially in glasses) and lines painted on the body in most of the ceramic vessels. This type of ceramic and decorative variety has been found in the most superficial layers of archaeological excavations; and, on the contrary, in the deeper horizons its abundance is smaller, at the same time, that new forms appear (plates, bowls with bases and in some cases with handles) and decorative techniques (incised, pressed or modeled) in the strata deeper. It is precisely this variety in the pottery, what has generated this study of

ceramics within the pre-Guane - Guane archaeological regional context, considering that it can be about the technological continuity of the same human group or that it can correspond to two agricultural societies, different potters.

## SAMPLES AND ANALYTICAL METHODS

Fourteen samples (pottery sherds) from the excavation site were selected for this study, taking into account that sampling represented the diversity observed macroscopically among pottery fabric, surface and decoration and is proportional to the number of pottery sherds identified in each category of the ceramic system of classification proposed by Moreno-González (2012). Samples were taken from original objects and then prepared for analysis, taking care not to include pottery sherds of the painted surfaces because this approach was not focused on the characterization of pigments. Pottery sherds were characterized through conventional petrography and X-ray powder diffraction (XRPD) and scanning electron microscopy (SEM). X-ray diffractograms were recorded in a XRPD using a BRUKER D8 ADVANCE X-ray diffractometer equipped with Da Vinci geometry and equipped, operating at 40 kV and 30 mA with Cu-K $\alpha$ 1 radiation:  $\lambda = 1.5406 \text{ \AA}$  in the  $2\theta$  range of  $3.5\text{-}70^\circ$ . Microtexture and chemical composition of pottery sherds were examined using the back-scattered electron (BSE) imaging on a FEI QUANTA 650 FEG-ESEM, under the following analytical conditions: magnification = 100-20000x, WD = 9.0-11.0 mm, HV = 20 kV, signal = BSE in ZCONT mode, detector = BSED, EDS Detector EDAX APOLO X with resolution of 126.1 eV (in. Mn K $\alpha$ ). Structural characterization from the functional groups by FTIR-ATR, using a computer model THERMO SCIENTIFIC IS50, with diamond crystal in the spectral range  $400\text{-}4000 \text{ cm}^{-1}$ .

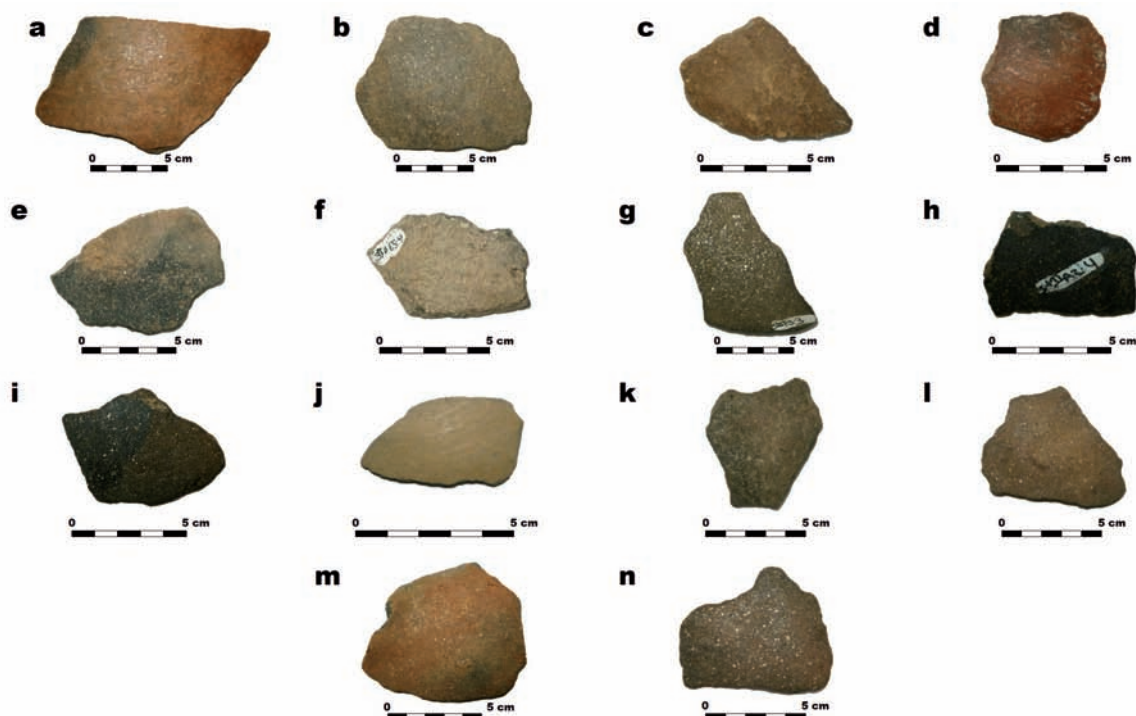
## RESULTS AND DISCUSSION

### Petrography

FIGURE 2 illustrates the analyzed pottery sherds from the pre-Hispanic Los Teres archaeological site, which were selected for this study basing on the principle of representativeness in which an attempt was made to cover the diversity of colors and pottery fabrics observed in the entire ceramic system. The observation of pottery sherds allowed identifying 5 main ceramic groups: red, gray, black, brown and redish brown groups (TABLE 1). A preliminary visual inspection

of the pottery surfaces was performed through a stereomicroscopy to distinguish the residues from the pottery material and optimize the sampling of the as-received pottery sherds. The observation of pottery sherds under the stereomicroscope, allowed identifying 5 main ceramic groups: red, gray, black, brown and reddish brown groups. Red group includes the A2-1 (orange-AC), F2-2 (intense-AC), A3-4 (matte-AC), D2-11 (orange-AC without mica), and B2-2 (intense-AC without mica) subgroups. Gray group includes the D4-4 (dark gray-AC), B5-6 (dark gray-CE without mica), F3-3 (light gray-AC), and E5-4 (light gray-AC without mica) subgroups. Black group includes the B4-6 (black-AC), and A2-4 (black-AC without mica) subgroups. Brown group includes the C3-3 (brown-AC) subgroup. Reddish brown group includes

the D2-6 (reddish brown-AC) subgroup. All subgroups with its subtypes with external, internal color or both. The classification of these ceramic groups was based on visual characteristics such as the external and internal color, surface treatment such as engobe or the absence of this, polishing, content, size and density in the structure of the paste of the degreaser and finding of a high amount of visible muscovite in the pottery sherd. Nonetheless, the mineralogy and composition of the paste weren't considered. To approach the compositional and mineralogy differences in the pottery sherds, the set of samples were taken from each one of the ceramic groups being each sample representative from each group and each sample were taken from original objects and then prepared for analysis, taking care not to include fragments of the painted surfaces.



**FIGURE 2.** Samples of pottery sherds from the pre-Hispanic Los Teres archaeological site: (A)-(D) red, (F)-(G) gray, (H)-(I) black, (J)-(M) brown and (N) reddish brown groups.

The use of petrographic analysis has been increasing in the study of archaeological ceramics. FIGURE 3 illustrates the main petrographic features of the analyzed pottery sherds. In general, they show an opaque and pseudo-isotropic microcrystalline and heterogeneous clayed matrix with frequent to common voids, which tend to show a subparallel orientation. It is brown to reddish-brown in PPL and black to dark

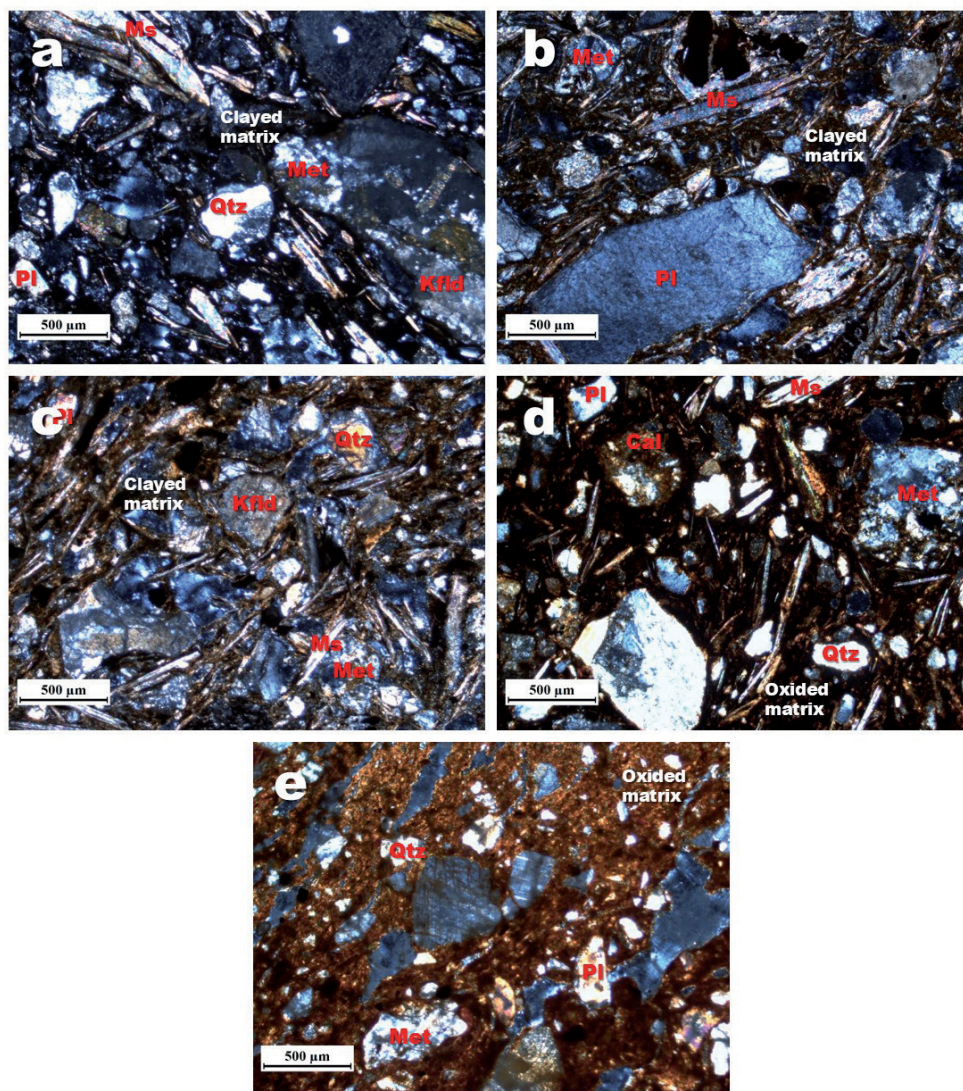
brown in XPL. Inclusions reveal that grains are poorly sorted. The coarse fraction is mainly composed of monocrystalline or polycrystalline quartz, plagioclase, micas (biotite and muscovite), amphibole, pyroxene, calcite, and opaque minerals lithoclasts of metamorphic rocks (quartzite, micaschist, biotite-amphibole or amphibolitic gneiss), with rare potassium feldspar. The fine fraction shows predominant to frequent opaque

minerals, monocrystalline quartz, common micas (biotite and muscovite), scarce plagioclase, potassium feldspar and amphibole. In general, one of the main characteristics for these pottery sherds is the presence of high amount of muscovite flakes. The textural pattern can be either preferentially oriented or showing a random orientation of muscovite. It must be specified that the presence of muscovite is a natural product, taking into account that in some pottery artifact it was intentionally added and, therefore, archaeologists who have worked in this region consider that it is an addition for decorative purposes. There is not evidence of the presence of high temperature anorthite, and, therefore, it was probably hidden in the amorphous phases. Most of the amorphous phases observed in the matrix were created after firing at a high temperature enough to dehydrate and dehydroxylate the clay minerals present in the raw clayed paste with which the ceramic artifacts were manufactured. Albite can occur, which

is a reflection of the stability of the mineral under changing firing temperatures, which may indicate that the raw clayed paste with which the ceramic artifacts were manufactured were exposed to high temperatures. Alkaline feldspars present in the pottery sherds probably proceed from the raw clayed paste with which the ceramic artifacts were manufactured. However, they also can react to firing conditions, with the formation of polymorphs of high temperature. Carbonate minerals occur mainly in samples of the red, black, brown and reddish brown groups are absent in samples of the gray group. There is an evident transformation of the raw clayed paste with which the ceramic artifacts were manufactured, which is accompanied by the dehydration and dehydroxylation of clay minerals and the formation of non-crystalline amorphous phases. Similarly, after cooking of the raw clayed paste for the manufacture of ceramic artifacts, some clay minerals can be rehydroxylized and reconstructed.

**TABLE 1.** Results of quantitative analysis by XRD, showing the percentage of pottery's crystalline and amorphous phases. NC represents that the crystalline phase was identified; however, due to its low percentage it is not possible to do its quantification.

| Code          | Group            | Subgroup                      | Unit/<br>serial            | Qtz   | Ms    | Ab    | Mic  | Or   | Cal   | Clc  | Kln  | Mag | Ant  | Mhbl | Cha  | Crystalline<br>phases (%) | Amorphous<br>phases (%) |
|---------------|------------------|-------------------------------|----------------------------|-------|-------|-------|------|------|-------|------|------|-----|------|------|------|---------------------------|-------------------------|
| A2-1<br>(1)   | Red              | Orange                        | Orange<br>AC/1             | 33.90 | 26.90 | 8.10  | NC   | NC   | NC    | -    | -    | NC  | -    | -    | -    | 68.90                     | 31.10                   |
| F2-2<br>(9)   |                  | Intense                       | Intense<br>AC/9            | 21.10 | 16.90 | 9.60  |      | 8.80 | NC    | -    | -    | -   | -    | -    | -    | 56.40                     | 43.60                   |
| A3-4<br>(17)  |                  | Matte                         | Matte<br>AC/17             |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| D3-11<br>(25) |                  | Orange<br>without<br>mica     | Orange<br>AC/25            | 27.50 | 5.00  | 10.40 | 8.90 | 2.80 | -     | -    | 2.30 | -   | -    | 3.50 | 0.50 | 60.90                     | 39.10                   |
| B2-2<br>(33)  |                  | Intense<br>without<br>mica    | Intense<br>AC/33           |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| D4-4<br>(49)  | Gray             | Dark gray                     | Dark gray<br>AC/49         | 26.20 | 10.30 | 14.10 | 4.60 | 6.00 | -     | -    | -    | -   | -    | -    | -    | 61.20                     | 38.80                   |
| B5-6<br>(57)  |                  | Dark gray<br>without<br>mica  | Dark gray<br>CE/57         | 26.40 | 5.10  | 25.10 | 6.20 | -    | -     | -    | -    | -   | -    | 3.80 | 3.10 | 69.70                     | 30.30                   |
| F3-3<br>(65)  |                  | Light gray                    | Light gray<br>AC/65        |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| E5-4<br>(73)  |                  | Light gray<br>without<br>mica | Light gray<br>AC/73        |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| B4-6<br>(81)  | Black            | Black                         | Black<br>AC/81             |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| A2-4<br>(89)  |                  | Black<br>without<br>mica      | Black<br>AC/89             |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| C3-3<br>(97)  | Brown            | Brown                         | Brown<br>AC/97             |       |       |       |      |      |       |      |      |     |      |      |      |                           |                         |
| A4-4<br>(105) |                  | Brown<br>without<br>mica      | Brown<br>AC/105            | 18.00 | 6.10  | -     |      | -    | 41.20 | -    | -    | -   | 0.20 | -    | -    | 65.50                     | 34.50                   |
| D2-6<br>(113) | Reddish<br>brown | Reddish<br>brown              | Reddish<br>brown<br>AC/113 | 18.60 | 25.50 | 3.90  |      | 4.50 | -     | 4.80 | -    | -   | -    | -    | -    | 57.30                     | 42.70                   |



**FIGURE 3.** Microphotographs of thin sections of the analyzed pottery sherds of the **A.** red, **B.** gray, **C.** black, **D.** brown and **E.** reddish brown groups (polarized light, crossed polarizers, scale bar = 500 µm. Met, clast of metamorphic rock).

### X-ray powder diffraction

XRPD analysis reveal that the observed mineralogical phases were albite, quartz, microcline, orthoclase, sanidine and plagioclase, clay minerals, micas, carbonate minerals, and iron oxides. Mineral abbreviations are after Whitney and Evans (2010). Traces of hematite can cause a vivid orange colour of the pottery sherds, which implies an oxidizing firing atmosphere, whereas samples do not contain hematite suggest a low firing temperature (Kramar *et al.*, 2012). The presence of magnetite reveals that it was formed under reducing conditions (Sciau, 2016). TABLE 1 shows the mineral

composition of the pottery sherds analyzed by XRPD. FIGURE 4 illustrates the corresponding XRPD patterns. Results indicate that the pottery sherds are mainly composed of quartz, muscovite, albite and orthoclase/microcline (except sample A4-4 (105), which contains abundant calcite), with minor Mg-hornblende, diopside, gehlenite, chabazite, anatase, clinocllore and kaolinite, and traces of magnetite/hematite. Data reveals that: 1) the crystalline phases range from 56.40 to 69.70% and the amorphous phases range from 30.30 to 43.60% and 2) the scarce occurrence of clay minerals (kaolinite) in the analyzed pottery sherds.

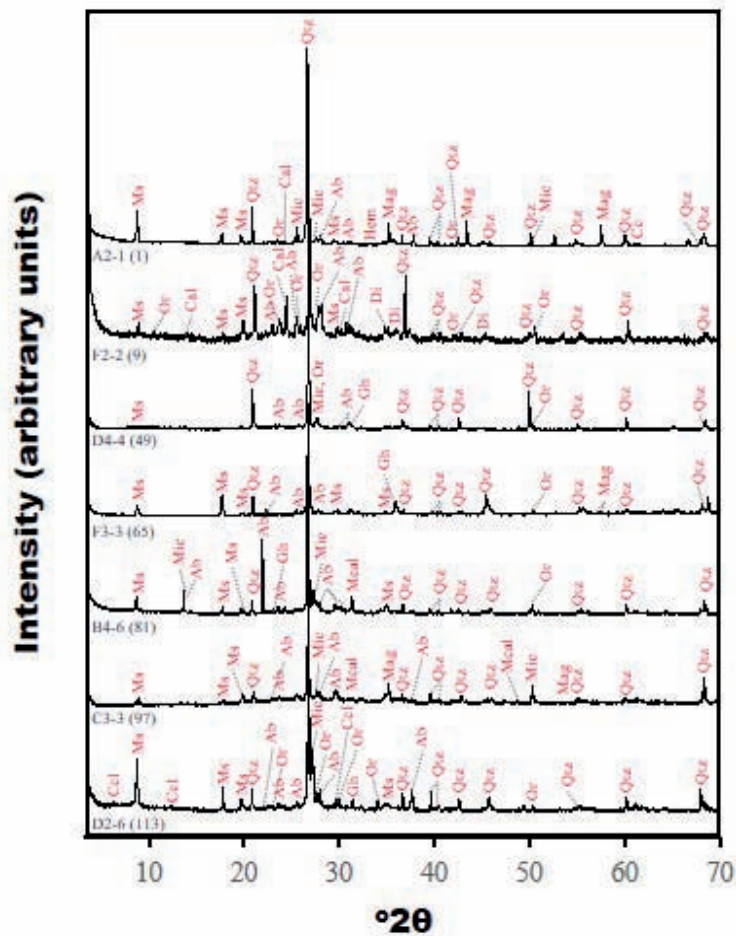


FIGURE 4. XRPD patterns of pottery sherds obtained by quantitative analyses.

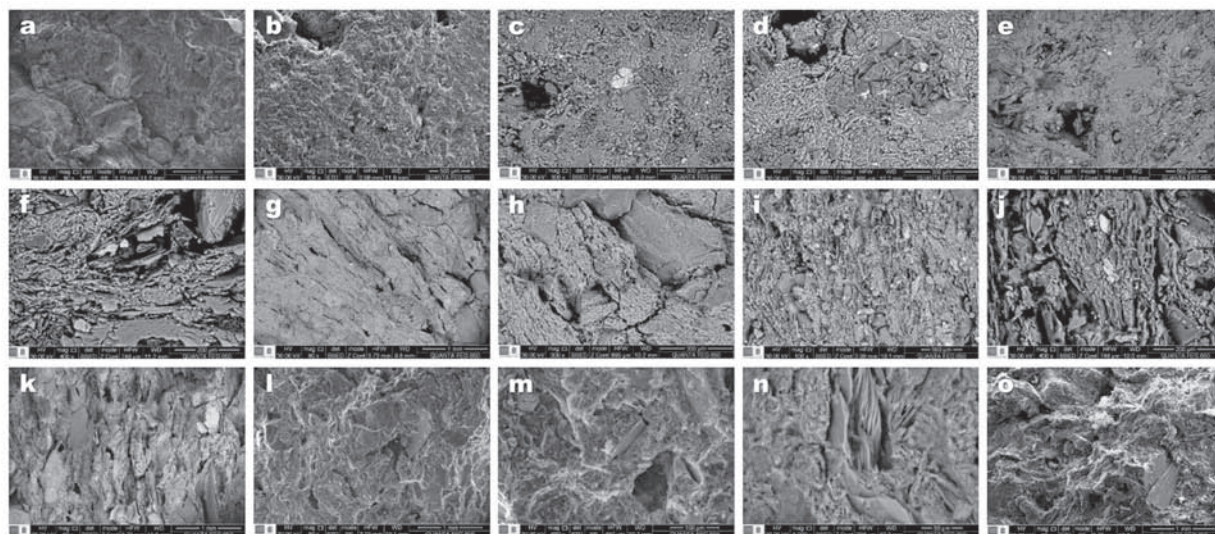
### Scanning electron microscopy

SEM analyses of the pottery sherds provided information about their microstructure and mineral morphology (FIGURE 5) developed during the firing of the raw materials used in their manufacture, which is very useful to estimate the firing temperature of the analyzed pottery sherds based on the characterization of their morphology and degree of vitrification as suggested by Tite *et al.* (1982). The degree of particle interconnection is important because it suggests an increase in the firing temperature. Maniatis and Tite (1978, 1981) defined, based on the fusion textures observed in SEM images, four degrees of vitrification (no vitrification; initial vitrification; vitrification; total vitrification) according to the temperature reached. No vitrification was mainly evidenced as a mortar texture. There are pottery sherds that show some light buckles and occasional rounded edges of the small clay plates, which suggest that although the pottery artifact did not

reach a state of vitrification, some minerals reacted at the firing temperature. Faced with this challenge it is decided to accept the degree of initial vitrification which do not have clearly interconnected filaments, but if they have a marked structural anisotropy. The vitrification is observed in most of the analyzed pottery sherds as vitrified filaments mostly connected to each other forming networks with the same orientation despite the amount of inclusions of amorphous matter, clay minerals and scarce pores. The total vitrification was evident in some pottery sherds, especially observed with the texture “bubbly”, which is attributed to partial fusion in high temperatures (850°C), particularly in non-carbonated ceramics (Maniatis and Tite, 1981). Taking into account that pores (from 5 to 100  $\mu\text{m}$ ) are isolated (FIGURE 5), they indicate that the total vitrification did not reach its final state. The development of the glassy phase which implies high temperatures, is in agreement with the indicators

of firing temperature or neoformation of minerals, suggesting that in the manufacture of the ancient ceramic artifacts the Guane potters paid more attention to control the degree of cooking of the vessel and

thus prevent it from fracturing and preventing fulfill a previously determined function, where the temperature control played an important role to achieve the result as an archaeological product.



**FIGURE 5.** SEM photomicrographs of pottery samples of the (A)-(C) red, (D)-(F) gray, (G)-(I) black, (J)-(L) brown and (M)-(O) reddish brown groups.

### Fourier transform infrared spectroscopy

FIGURE 6 illustrates the FTIR-ATR spectra of the analyzed pottery sherds. According to the FTIR analysis of the archeological artifacts, there is not a great variation on mineralogy, which reveal that the analyzed pottery shreds can represent products manufactured from raw materials from a similar origin. The absorption bands at  $3420\text{--}3431\text{ cm}^{-1}$  can be attributed to the OH stretching vibration of  $\text{H}_2\text{O}$  adsorbed on the surface of pottery and the bending  $\text{H}\text{--}\text{O}\text{--}\text{H}$  vibration ( $\delta$ ) of water due to the exposure of the powders to atmosphere, respectively, as reported by Dubnikova *et al.* (2011). The absorption bands at  $3420\text{--}3436\text{ cm}^{-1}$  (stretching vibration of  $\text{--OH}$ ) can be attributed to quartz or calcite. The weak absorption band at  $1636\text{--}1649\text{ cm}^{-1}$  indicates OH bending modes of water or hydroxyls in the analyzed samples (Schrader, 1995). The peak at  $1636\text{ cm}^{-1}$  corresponds to the bending vibration of  $\text{--OH}$  from  $\text{Si}\text{--}\text{OH}$  silanol groups in quartz. The absorption bands at  $1636\text{--}1649\text{ cm}^{-1}$  is attributed to the  $\text{H}\text{--}\text{O}\text{--}\text{H}$  bending of water molecule existing in muscovite-rich samples as suggested by Palanivel and Velraj (2007). The main  $\text{CO}_3$  band of the calcite in the pottery is located at  $1420\text{--}1450\text{ cm}^{-1}$ . The characteristic peaks ranging from  $900$  to  $1200\text{ cm}^{-1}$

are attributed to the symmetrical stretching of  $\text{Si}\text{--}\text{O}$  bonds (*e.g.*, Chaikina and Kryukova, 2004; Zhou *et al.*, 2008) in quartz, albite and muscovite. The absence of a strong intensity peak at  $1085\text{ cm}^{-1}$  in the spectra reveals that wollastonite is not present in the analyzed pottery shreds, which has been reported in previous studies (*e.g.*, Dowty, 1987). The absorption bands at  $793\text{--}797\text{ cm}^{-1}$  and  $775\text{--}780\text{ cm}^{-1}$  are attributed to the bending vibrations of  $\text{Si}\text{--}\text{O}$  bonds (*e.g.*, Chaikina and Kryukova, 2004; Zhou *et al.*, 2008) and deformation of  $\text{Si}\text{--}\text{O}$  bonds respectively in quartz. The absorption band at  $569\text{ cm}^{-1}$  can be attributed to the  $\text{Fe}\text{--}\text{O}$  vibration of magnetite (Velraj *et al.*, 2009). However, hematite, which is characterized by absorption bands at  $530\text{--}531\text{ cm}^{-1}$  also occur. According to Palanivel and Rajesh-Kumar (2011), the formation of magnetite and hematite depend on the firing atmosphere prevalent at the time of manufacture. The presence of weak intensity peaks of hematite reveals the incipient transformation of magnetite ( $\text{Fe}_3\text{O}_4$ ) to hematite ( $\text{Fe}_2\text{O}_3$ ) during the firing process under an oxidizing condition (Velraj *et al.*, 2009). The absorption bands at  $464\text{--}469\text{ cm}^{-1}$  are assigned to the presence of quartz, albite and microcline. The  $\text{Si}\text{--}\text{O}$  bending mode of quartz appears at  $465\text{--}470\text{ cm}^{-1}$ .

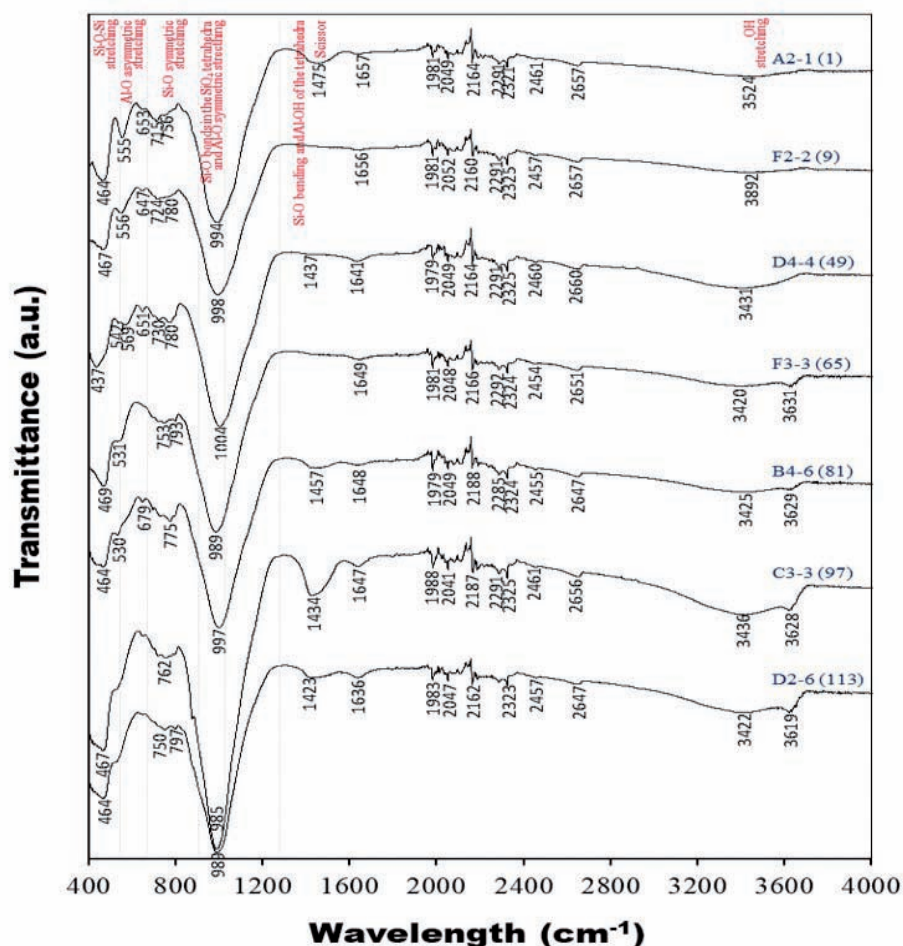


FIGURE 6. FTIR-ATR spectra of the analyzed pottery sherds.

The samples A2-1 (1) and F2-2 (9) from the red group (FIGURE 7A) show similar spectra, which indicates the presence of quartz (695, 794, 1160  $\text{cm}^{-1}$ ), calcite (711, 835, 879 and 1456  $\text{cm}^{-1}$ ), muscovite (1062 and 553  $\text{cm}^{-1}$ ), muscovite/albite (990  $\text{cm}^{-1}$ ), albite/microcline (742  $\text{cm}^{-1}$ ) and albite (649 and 1032  $\text{cm}^{-1}$ ). Muscovite, microcline and albite are more clearly inferred using the second derivative. In the brown group, sample C3-3 (97) has higher absorbance rates than sample A3-4 (17) (FIGURE 7B). The vibration bands at 694, 798 and 1160  $\text{cm}^{-1}$ , and at 875 and 1420  $\text{cm}^{-1}$ , can be attributed to quartz and calcite, respectively. The second derivative reveals the presence of muscovite at 1022  $\text{cm}^{-1}$ , albite at 530 and 1096  $\text{cm}^{-1}$ , microcline at 463, 648, 768, 1049, 1134 and 1142  $\text{cm}^{-1}$ , wollastonite at 540, 640, 930, 1026, 1057 and 1088  $\text{cm}^{-1}$ , kaolinite at 938 and 1034  $\text{cm}^{-1}$ , illite at 1030  $\text{cm}^{-1}$ , diopside at 470, 510, 864 and 965  $\text{cm}^{-1}$ , anorthite at 482, 540, 575 and 668  $\text{cm}^{-1}$ , and orthoclase at 428  $\text{cm}^{-1}$ . The highest absorption peak is 990  $\text{cm}^{-1}$ ,

which can be attributed to muscovite or albite. FIGURE 7C shows quite similar spectra for samples F3-3 (65) of the gray group and D2-6 (113) of the reddish brown group. However, the spectra present variations in the number of bands with presence of calcite, in F3-3 (65) (714, 877 and 1420  $\text{cm}^{-1}$ ) and D2-6 (113) (714  $\text{cm}^{-1}$ ). The spectra also show the presence of quartz (695, 777, 797 and 1160  $\text{cm}^{-1}$ ), muscovite/albite (982 and 990  $\text{cm}^{-1}$ ), albite (530 and 746  $\text{cm}^{-1}$ ) and microcline (650  $\text{cm}^{-1}$ ). On the other hand, the second derivative shows that the sample with the highest amount of calcite the bands are more pronounced and displays minor quartz bands and probably presents anorthite (668  $\text{cm}^{-1}$ ). The red group (FIGURE 7D) shows more similar spectra to each other although with different indices of absorbance. The spectrum reveals the presence of quartz (695, 777, 798 and 1160  $\text{cm}^{-1}$ ), orthoclase/muscovite (535  $\text{cm}^{-1}$ ), orthoclase (580  $\text{cm}^{-1}$ ), albite (530 and 645  $\text{cm}^{-1}$ ), muscovite/albite (725  $\text{cm}^{-1}$ ), muscovite (1021  $\text{cm}^{-1}$ ) and

kaolinite ( $1000\text{ cm}^{-1}$ ). Calcite at  $1420\text{ cm}^{-1}$  was observed only in sample B4-6 (81). In the second derivative, microcline, wollastonite, sanidine, kaolinite, anorthite and calcite were observed. A small amount of diopside with an absorption peak of  $470\text{ cm}^{-1}$  was also observed. FIGURE 7E illustrates the spectrum of sample B2-2 (33) (red group), which indicates the presence of quartz ( $530, 695, 776, 795$  and  $1160\text{ cm}^{-1}$ ), albite ( $1030\text{ cm}^{-1}$ ) and microcline ( $646\text{ cm}^{-1}$ ). In the second derivative, muscovite at  $1022\text{ cm}^{-1}$ , kaolinite at  $1010\text{ cm}^{-1}$ , diopside

at  $965\text{ cm}^{-1}$  and anorthite at  $575, 733$  and  $950\text{ cm}^{-1}$ , wollastonite ( $563, 930$  and  $1088\text{ cm}^{-1}$ ) and sanidine at  $1026\text{ cm}^{-1}$  can be distinguished. FIGURE 7F illustrates the spectrum of sample A2-4 (89), which indicates the presence of quartz ( $512, 694, 778, 797$  and  $1160\text{ cm}^{-1}$ ), calcite ( $712, 874$  and  $1420\text{ cm}^{-1}$ ), and microcline/orthoclase ( $1010\text{ cm}^{-1}$ ). It is also possible to identify in the second derivative minerals as albite, orthoclase, microcline, sanidine, anorthite, muscovite, wollastonite and kaolinite.

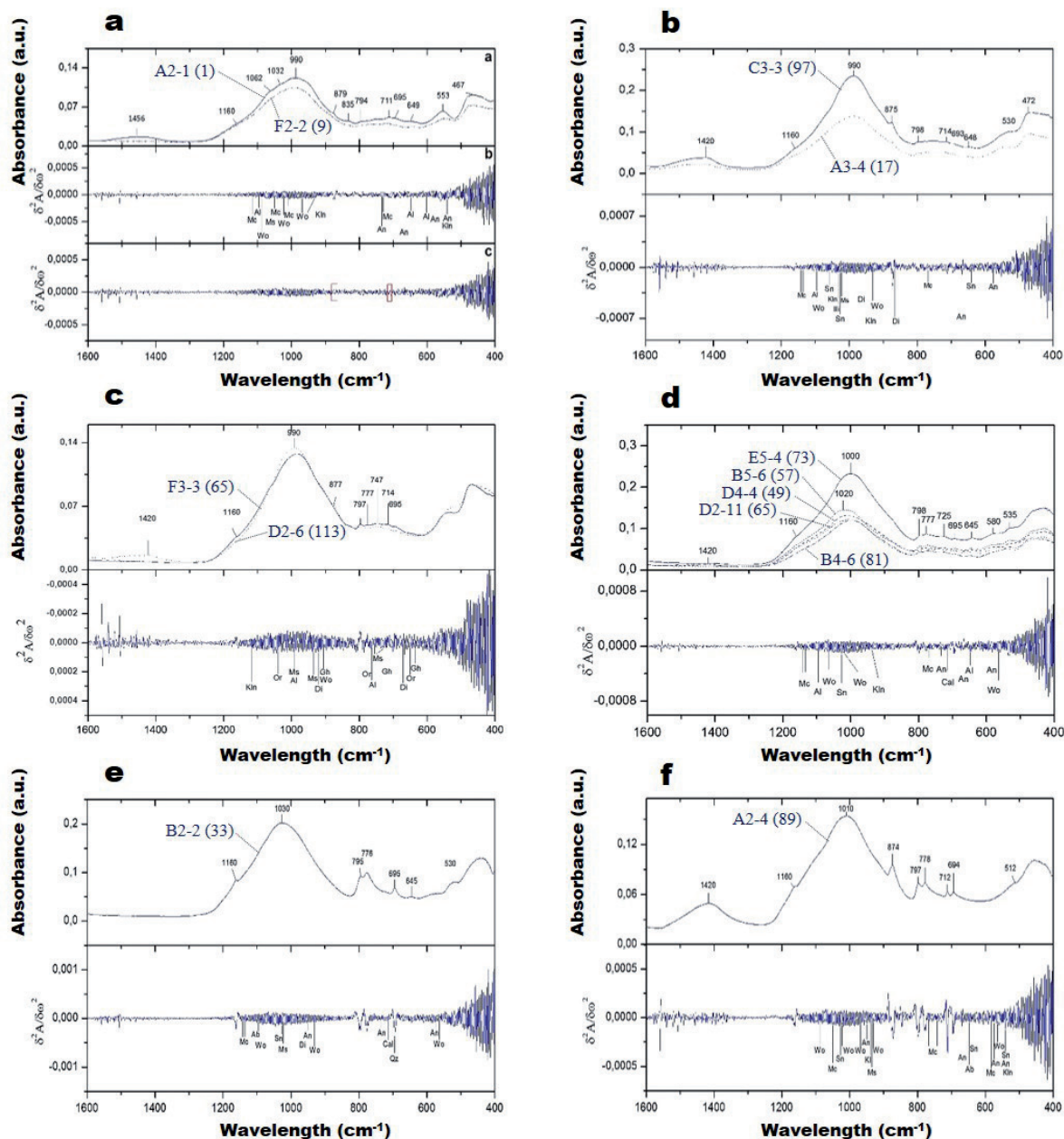


FIGURE 7. FTIR-ATR spectra and second derivative of the analyzed pottery sherds (spectra according to TABLE 1).

## Estimation of the firing temperature conditions for the manufacture of potteries

The raw material for manufacturing was undoubtedly clay, which consists of clay minerals (kaolinite) mixed with a variety of other minerals as quartz, muscovite, potassium feldspar and plagioclase. Quartz, potassium feldspar and plagioclase are stable even at high temperatures, so that they cannot be considered for the estimation of the firing temperature (*e.g.*, İssi *et al.*, 2011). According to Kakali *et al.* (2001), between ~400-650°C, the kaolinite dehydroxylation promotes its transformation to an amorphous phase (metakaolinite), which can occur in the temperature range of 450-650°C (*e.g.*, Frost *et al.*, 2003; Chakchouk *et al.*, 2009). Non-calcareous clays show minor occurrence of magnetite and/or hematite. It is probably that clay minerals disappeared between 550°C-950°C. Close to 950°C, orthoclase and/or microcline can form at expenses of the progressive disappearance of plagioclase and muscovite, reaching its maximum abundance at 1100°C. On the other hand, taking into account that muscovite disappears at 950-1000°C and plagioclase at ~1100°C, we suggest a maximum firing temperature of 950°C. Carbonate-rich clays show variable amounts of calcite and small amounts of magnesium calcite, plagioclase and alkaline feldspar. Clay minerals were the first minerals to disappear at 950°C. Carbonates can react with other components to produce mineral phases as gehlenite, diopside, and wollastonite (Trindade *et al.*, 2009). At temperatures above 950°C in the presence of sufficient SiO<sub>2</sub> and CaO, gehlenite can react to anorthite and wollastonite. The presence of carbonates suggests firing temperatures lower than 700°C. According to Grapes (2010), calcite reacts at 600-700°C, which is unlikely taking into account that it still remains in some pottery sherds. Several studies (*e.g.*, Holakoei *et al.*, 2014; Velraj *et al.*, 2015) reveal that the maximum temperature after which calcite occurs is 900°C. However, the presence of magnesium calcite is indicative of temperatures slightly lower than those indicated by calcite, taking into account that it reacts in the range 500-650°C (Grapes, 2010). Alkaline feldspar disappeared at temperatures close to 950°C and plagioclase seemed to persist up to 1100°C. Quartz is known for its low reactivity and inalterability despite high temperature changes. According to Liou (2015), anatase is a stable mineral phase, which can remain after firing of the raw materials, suggesting a firing temperature in the range of 750-950°C. Firing conditions were variable, under oxidation and reduction environments, although the presence of magnetite possibly indicates predominantly reducing conditions.

With regard to the origin of the raw materials with which the pottery sherds object of this study were prepared, they probably came from the middle member of the Los Santos Formation, which is composed of claystones, siltstones and muddy fine-grained sandstones of very local extension, developing a thin layer on top of the Mesa de Los Santos region. We consider that the local manufactured pottery is the same without changes in the time but only with variations on the surface colors, examining the possible regional provenance of the pottery sherds and elucidating broad questions regarding the raw clayed paste for manufacture choices and the sociocultural background of the pre-Hispanic cultures.

## CONCLUSIONS

This study dealt with the characterization of fourteen pottery sherds from Los Teres archaeological site located in the region of the Mesa de Los Santos region, Colombia. The pottery sherds were studied by the archaeologist to define the ceramic system, based on two determining variables present in the sample: (1) the color of the bath or slip of the surface and (2) the mineralogical composition of the paste for each group. To correlate these two variables of archaeological analysis and determine if the communities added as a degreaser to the paste, substances or local or foreign minerals, XRPD, SEM and FTIR-ATR analyses were performed in order to extract information about mineralogy and firing temperatures as a contribution to elucidate a possible regional provenance. On the other hand, from the cultural point of view, the decoration color in slips does come to be a differentiating marker of the cultural group and the use of ceramics. There are not significant changes regard firing temperature conditions and mineralogy, which suggests that there were no significant transformations in the treatment of pasta and its incidence in the ceramic body. Results reported in this work contribute to developing and progressing in the archaeometry analysis of pottery sherds in Colombia, and in the advancing in the knowledge and understanding of ancient civilizations. The results reported in this paper also show that there is a management of local mineral resources, both to make lithic tools, and to grind them and use them in the paste as a degreaser for the manufacture of pottery. Likewise, they contribute to the development and progress in the analysis of the archeometry of ceramic pottery in northeastern Colombia and in the advance in the knowledge and understanding of the pre-Hispanic societies in this region.

## ACKNOWLEDGEMENTS

We gratefully acknowledge the Vicerrectoría de Investigación y Extensión of the Universidad Industrial de Santander for the use of their research facilities. Authors thank the Microscopy and X-Rays laboratories of the Guatiguará Technology Park and their staff for the analytical service provided for data acquisition. The authors would also like to acknowledge the anonymous referees for their critical and insightful reading of the manuscript and are most grateful to the above-named people and institutions for support.

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Trabajo recibido: agosto 26 de 2018  
Trabajo aceptado: abril 09 de 2019