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Short report

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Keywords: *Avocado, Antioxidant, Total phenols, FRAP, DPPH, ABTS.*

Palabras clave: *Palta, Antioxidante, Fenoles totales, FRAP, DPPH, ABTS.*

ABSTRACT

The high content of bioactive compounds in *Persea americana* seed has generated great interest worldwide due to its various potentialities, in which its antioxidant capacity stands out. The objective of the study was to evaluate the antioxidant activity of the aqueous and methanolic extracts of *Persea americana* Mill seeds, variety Hass, sold in supermarkets in the city of Maracay, Venezuela. The total phenol content of the extract was determined through the Folin-Ciocalteu method to then evaluate the antioxidant activity by three chemical methods (DPPH, FRAP and ABTS). A higher concentration of total phenolic compounds was obtained in the methanolic extract, with a statistical difference with respect to the aqueous extract. The antioxidant activity was higher in the methanolic extract in the three methods used ($p < 0.05$). These results show the high antioxidant power of the avocado seed, considered as a waste product, which allows proposing strategies that allow its use for medicinal purposes.

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RESUMEN



El alto contenido de compuestos bioactivos en la semilla de *Persea americana*, ha originado un gran interés a nivel mundial por sus diversas potencialidades, en la que destaca su capacidad antioxidante. El estudio tuvo como objetivo evaluar la actividad antioxidante de los extractos acuoso y metanólico de semillas de *Persea americana* Mill, variedad Hass, comercializadas en supermercados de la ciudad de Maracay, Venezuela. Se determinó el contenido de fenoles totales del extracto a través del método Folin-Ciocalteu para luego evaluar la actividad antioxidante por tres métodos químicos (DPPH, FRAP y ABTS). Se obtuvo una mayor concentración de compuestos fenólicos totales en el extracto metanólico, con diferencia estadística respecto al extracto acuoso. La actividad antioxidante fue superior en el extracto metanólico en los tres métodos empleados ($p < 0.05$). Estos resultados muestran el alto poder antioxidante de la semilla de aguacate, considerada como producto de desecho, lo que permite proponer estrategias que permitan su uso con fines medicinales.

INTRODUCCIÓN

Avocado is an evergreen tree belonging to the order Ranales, family Lauraceae and genus *Persea*, native to Central America and Mexico that can now be found in most tropical and subtropical regions of the world, such as in Colombia, Perú and Venezuela [1,2]. Among the species is the *Persea americana* classified by Miller, which developed several subspecies due to its geographical isolation, which originated different botanical types [3,4]. Avocado consumption has increased because it is increasingly valued by consumers, not only for its unique taste and texture but also for its health benefits [5,6].

The separation, isolation and characterization of phenolic compounds in foods, specifically in avocado waste, not only increases the possibility of increasing the number of phytochemicals that can improve the quality of life of people, but also the alternative of using waste by-products (epicarp and seed) at industrial level and thus achieve a comprehensive use of this fruit [7]. Seeds have been found to possess insecticidal, fungicidal and antimicrobial activities with a high content of phenolic compounds, which play an important role in potential health effects [8]. Therefore, the purpose of the present study was to evaluate the antioxidant capacity of the aqueous and methanolic extracts of avocado seed (*Persea americana* Mill, variety Hass) by three chemical methods.

EXPERIMENTAL

Origin of plant material (PM)

The seeds were obtained from avocados sold in a super market in the city of Maracay, Aragua state, Venezuela.

Sample preparation for extraction

To obtain the alcoholic extract, 10 g of plant material were weighed and previously crushed, placing them in a 400 mL beaker with 300 mL of 70% v/v methanol/water. It was macerated and left to rest for 24 h, and then it was subjected to microwave radiation for 15 s with a power of 125 MHz. The extracts were filtered, using Whatman filter paper number 4. The extract was left to macerate for 24 h, filtering with Whatman filter paper number 4[9].

Determination of total phenolics

For the determination of total phenolics, 50 μ L of the alcoholic extract were mixed with 250 μ L of the Folin-Ciocalteu 1 N reagent (Analytical grade, Merck). It was left to stand for 8 minutes and then 750 μ L of 20% Na_2CO_3 and 950 μ L of distilled water were added. Was incubated for 30 min at room temperature and the absorbance was read on a Genesis 20 UV/VIS spectrophotometer (Thermo Scientific, Waltham, Massachusetts, USA). A calibration curve for Gallic Acid (Sigma-Aldrich, Germany) was prepared with concentrations of 50, 100, 200, 300, 400, 500 and 1000 ppm. The results were expressed in mg of Gallic Acid Equivalents (GAE) / g of PM [10].

Ferric-Reducing Antioxidant Power (FRAP) Assay



The FRAP test was used to determine the reducing capacity of infusions [11]: 100 μL of infusion were mixed with 3 mL of FRAP reagent consisting of a mixture of 300 mM sodium acetate and acetic acid buffer, 10 mM TPTZ solution (2,4,6-tri(2-pyridyl)-s-triazine), and 20 mM FeCl_3 solution, in a volume ratio of 10:1:1.



Figure 1. Avocado seeds

The reaction was carried out at room temperature for 4 min, and absorbance was recorded at 593 nm using a Genesis 20 UV/VIS spectrophotometer (Thermo Scientific, Waltham, Massachusetts, USA). FeSO_4 was used as standard, and the results were expressed as $\mu\text{mol Fe}^{2+}/\text{g PM}$.

Free radical 2,2-diphenyl-1-picrylhydrazyl (DPPH) Assay

100 μL of sample and 2.9 mL DPPH (100 mM solution of DPPH in 80 % methanol) (Sigma Aldrich) were placed together in a quartz cell. Absorbance (Genesis 20 UV/VIS spectrophotometer) was monitored every 5 min for 30 min at a wavelength of 515 nm. The reference absorbance (A_0) was obtained by substituting the sample volume for 80 % methanol. The percentage of DPPH reduction was obtained with the equation $\text{DPPH (\%)} = (A_0 - A_n) 100 / A_0$, where A_0 and A_n were the reference and sample absorbance, respectively [12]. The data were used to determine the IC_{50} parameter, which represents the concentration ($\mu\text{g.mL}^{-1}$) of phenolic compounds required to reduce the DPPH free radicals by 50 % [5].

2,2'-Azino-bis (3-ethylbenzothiazoline-6-sulfonic acid Assay (ABTS)

This test is based on the ability of antioxidants to sequester the cation radical $\text{ABTS}^{*\cdot}$ (7 mM with $\text{K}_2\text{S}_2\text{O}_8$ at 2.45 mM)[13]. The reaction mixture was kept in the dark at room temperature for 14 h. After time, the $\text{ABTS}^{*\cdot}$ solution was diluted with ethanol to obtain an absorbance of 0.70 ± 0.02 at 734 nm, using a Genesis 20 UV/VIS spectrophotometer (Thermo Scientific, Waltham, Massachusetts, USA). Subsequently, 250 μL of $\text{ABTS}^{*\cdot}$ solution diluted with methanol was added to 10 μL of sample. Radical reduction was monitored for 5 min at 734 nm, using ascorbic acid (Vitamin C) as standard. The results were expressed in mg/mL^{-1} .

Statistical analysis

All determinations were made fivefold and the values were expressed as the means $\pm$ the standard deviation. Statistical differences were determined by analysis of variance (ANOVA) by using the Statistic 9.0 program for Windows.



RESULTS AND DISCUSSION

Total Phenolics of the extracts

Prior to the evaluation of the antioxidant activity of the extracts, the concentration of total Phenolics was determined in triplicate, obtaining a concentration of 26.32 ± 2.45 mg GAE/g PM in the aqueous extract and $57, 32 \pm 2.45$ mg GAE/g PM for the methanolic extract, observing a statistical difference ($p=0.035$). The difference observed is similar to that reported in several studies carried out with *Persea americana* Mill, which indicate that the extraction of phenolic compounds is influenced by the type of solvent used [14,15]

Antioxidant activity

The antioxidant capacity was determinate as recommended by several studies, which indicate that at least two methods should be used to evaluate the antioxidant activity of an extract [16-18]. The antioxidant effect by the DPPH and FRAP method is shown in Table 1. The antioxidant capacity observed by the DPPH and FRAP methods shows that the extracts possess an important source of antioxidant compounds, a conclusion specially valid when compared to other studies worldwide, which indicate that avocado seeds can be used as a functional food ingredient or as an antioxidant additive [19,20].

Table 1. Antioxidant activity of extracts.

Extracts	Methanolic	Aqueous	<i>p</i>
FRAP (Fe +2 / g MP)	118.2 ± 1.13	95.11 ± 0.24	0.041
DPPH ($\mu\text{g.mL}^{-1}$)	46.2 ± 0.17	64.80 ± 0.13	0.039

For each method, means $\pm$ SD. Significantly different ($p < 0.05$).

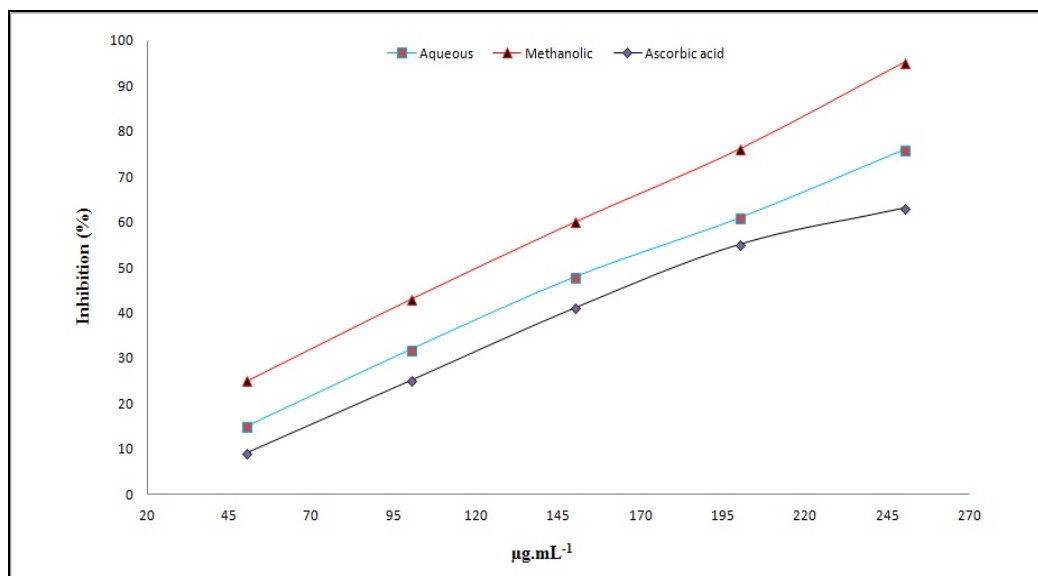


Fig 2. Evaluation of antioxidant activity by ABTS method

Among the methods used to determine the capacity of an antioxidant to capture free radicals, the ABTS radical is one of the most applied, as it is considered highly sensitive, practical, fast and very stable [21]. The ABTS assay showed



higher inhibition in the methanolic extract with statistical difference from the aqueous extract (Figure 2). These results coincide with what was reported in Bogota Colombia, whose study consisted in determining the phenolic profile of avocado byproducts, among them the seed, indicating that the extracts showed an antioxidant capacity (ABTS method) related to their content of bioactive compounds, considering these byproducts for pharmaceutical and industrial use [15].

CONCLUSIONS

We may conclude that the seed of *Persea americana* Mill variety Hass, constitutes an excellent source of phenolic compounds with high antioxidant power, conditioned to the way in which these are extracted from the plant material. Finally, this study supports and extends what has been found in several countries where avocado is a daily food, with high projection of exportation from South America to the Asian continent.

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