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Passivation process quality in reinforced concrete: effects of polarization periodicity and passivation consolidation parameters on data processing

Calidad del proceso de pasivación en concreto reforzado: efectos de la periodicidad de la polarización y los parámetros de la consolidación de la pasivación en el procesamiento de datos

Qualidade do processo de passivação em concreto armado: efeitos da periodicidade da polarização e parâmetros da consolidação da passivação no processamento de dados

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ABSTRACT:

The passivation process quality was studied considering polarization periodicity, passivation consolidation parameters, and data processing. Passivation process quality in steel reinforcement affects a structure's planned future service life. Some research has addressed this phenomenon, but its study is complicated by the limits of analog-era data, dispersion in corrosion rate data, and their interpretation. Two series of small reinforced concrete specimens were built using two water/cement ratios and two curing/storage combinations and exposed to the marine environment. Polarization periodicity did not affect passivation/depasivation during passivation but on the data processing. The curing and storage process influenced the tendency towards depasivation. Post-curing storage type affected the cumulative corrosion rate from 1 to 5 $\mu\text{A}\cdot\text{day}/\text{cm}^2$; this is equivalent to the margin of uncertainty in interpretation.

KEYWORDS: cumulative corrosion rate, w/c ratio, reinforced concrete, tropical marine environment.

RESUMEN:

La calidad del proceso de pasivación se estudió considerando la periodicidad de polarización, los parámetros de consolidación de pasivación y el tratamiento de datos. Algunas investigaciones han abordado la calidad del proceso de pasivación, pero su estudio se complica por la limitación de los datos de la era analógica, la dispersión de éstos y la interpretación de la velocidad de corrosión. Se construyeron dos series de especímenes de concreto reforzado utilizando dos relaciones agua/cemento y dos combinaciones de curado/almacenamiento y se expusieron en un medio marino. La periodicidad de la polarización no tuvo efecto sobre la pasivación/despasivación durante la pasivación, sino sobre el tratamiento de los datos. El proceso de curado y almacenamiento influyó en la tendencia a la despasivación. El tipo de almacenamiento posterior al curado afectó la velocidad de corrosión acumulada de 1 a 5 $\mu\text{A}\cdot\text{día}/\text{cm}^2$; esto es equivalente al margen de incertidumbre en la interpretación.

PALABRAS CLAVE: velocidad de corrosión acumulada, relación a/c, concreto reforzado, ambiente tropical marino.

RESUMO:

A qualidade do processo de passivação foi estudada considerando a periodicidade da polarização, os parâmetros de consolidação da passivação e o tratamento dos dados. Algumas pesquisas abordaram a qualidade do processo de passivação, mas seu estudo é complicado pela limitação dos dados da idade analógica, pela dispersão destes e pela interpretação da taxa de corrosão. Duas

séries de amostras de concreto armado foram construídas usando duas relações água / cimento e duas combinações de cura / armazenamento e expostas em um ambiente marinho. A periodicidade da polarização não teve efeito na passivação / despassivação durante a passivação, mas no tratamento dos dados. O processo de cura e armazenamento influenciou a tendência à despassivação. O tipo de armazenamento pós-cura afetou a taxa de corrosão cumulativa de 1 a 5 $\mu\text{A} \cdot \text{dia}/\text{cm}^2$; isso é equivalente à margem de incerteza na interpretação.

PALAVRAS-CHAVE: taxa de corrosão acumulada, relação a / c, concreto armado, ambiente marinho tropical.

INTRODUCTION

In natural conditions, a layer of oxides that protects reinforcing steel is consolidated over time (Poursae & Hansson, 2007). This layer is formed due to the high alkalinity of the concrete and is known as the passivation layer (Andrade et al., 1995). This layer provides a barrier against aggressive agents, on whose stability of the passivation quality depends. Therefore, when the structures are exposed to an aggressive natural environment, the time for passivation to occur and consolidate is very important, since the electrochemical behavior of the reinforcing steel will depend on it and, therefore, its corrosion process (Ahmad & Kumar, 2013), understanding as consolidation the electrochemical stability of the passive layer. There are studies on the passivation of the reinforcement, as well as the initiation and propagation stages of corrosion (Andrade et al., 1995; Huet et al., 2005; Jiang et al., 2017). However, the objectives of these studies did not necessarily took into account providing adequate times for steel to consolidate passivation (Poursae & Hansson, 2007). Furthermore, the variability in the study methods of passivation (Alonso et al., 2000; Jiang et al., 2017; Page, 2009; Poursae & Hansson, 2007) may cause a discrepancy in the interpretation of the same, causing it to have overestimation or underestimation values. Data on corrosion in reinforced concrete is commonly expressed as instantaneous corrosion rate, i_{corr} , as a function of time (Alonso et al, 2000; Andrade & Alonso, 1996; Castro-Borges et al, 2013; Castro-Borges et al, 2017; Lopez & Gonzalez, 1993; Pech-Canul & Castro, 2002; Pedrosa & Andrade, 2010; Rebolledo & Andrade, 2010). Many studies include analysis of i_{corr} behavior from the passivation zone to the moment when it begins to exhibit clear signs of depassivation. This helps in evaluating the corrosion resistance of different materials. Some works of both: before the 2000 year (Alonso et al., 2000; Andrade & Gonzalez, 1978; Lopez & Gonzalez, 1993) and after the 2000 year (Ghods et al, 2009; Pech-Canul & Castro, 2002; POURSAEE & HANSSON, 2007) sources, address the phenomena that occur before depassivation. However, the limitation of using analog data from earlier studies, and the intrinsic dispersion of i_{corr} data have delayed a deeper understanding of i_{corr} tendencies in the passivation zone. Previous studies (Andrade & Alonso, 1996; Castro-Borges et al., 2013, 2017; Melchers & Li, 2006; Pedrosa & Andrade, 2010; Rebolledo & Andrade, 2010) include passivation zone analyses and some of them found interpretations of the data trends by applying straightforward mathematical processing, which of course would need to be confirmed under different exposure conditions.

The current practice of designing reinforced concrete structures with nearly one-hundred-year service lives implies the application of durability criteria to prevent corrosion. This calls for a much deeper understanding of corrosion in the passivation zone. Attaining passivation zone stability and durability require optimum concrete quality (curing conditions, water/cement ratio), knowledge of the exposure environment (prevailing and non-prevailing winds, the regime of temperature and relative humidity) and damage prevention measures (concrete cover, surface sealing) among other factors. An inadequate combination of these at an early age can negatively affect the passivation quality and future service life. The study of the passivation quality in the steel embedded in concrete is vital since it can significantly affect later electrochemical behavior and consequently a structure's short-, medium- and long-term service life.

However, routine procedures for long-term field studies, with a variety of combined objectives and different materials, do not easily allow a detailed study of the passivation quality at an early age, and therefore the impact of this situation is not known in the service life of the element. An example of this is usually

when the measurements are performed in the field at intervals of low periodicity. This situation could mask trends that could be useful for service life models. For these reasons, doubts arose after more than ten years of experimentation (from 2007) in field specimens in which a relatively early depassivation process was expected, which did not occur. Then, it was decided to do, ten years later, another set of specimens (from 2017), in which some parameters in their composition and exposure would be modified to be able to analyze in more detail the quality of the passivation.

On the other hand, studying passivation quality requires full knowledge of the general criteria used to define a reinforcement bar as passive, in transition or active. The most accepted criterion is that passivation ends when the corrosion rate exceeds the $0.1 - 0.2 \mu\text{A}/\text{cm}$ range (Andrade & Gonzalez, 1978; Berke et al, 1996), although some widely used criteria include a range of $0.1 - 0.5 \mu\text{A}/\text{cm}$. (Andrade & Alonso, 1996; NMX-C-501-ONNCCE, 2015; Troconis et al, 1998). For practical purposes, a conservative criterion of values $< 0.1 \mu\text{A}/\text{cm}$ indicates passivity in concrete reinforcement. Considering the above, the present study objective was to analyze the passivation quality in concrete reinforcement in terms of polarization periodicity during measurement, passivation parameters and data processing.

For this paper, only the data corresponding to a concrete cover of 20 mm is analyzed, which is the most related to the typical situation of concrete in the region.

EXPERIMENTAL PROCEDURE

Specimen design

Using the ACI design method (ACI 211-91, 2002), twelve prismatic specimens (150x150x300 mm) were made using Portland Compound Cement (CPC30R) standardized by Mexican standard (NMX-C-414-ONNCCE, 2014), and limestone rock aggregate typical of southeast Mexico. Table 1 shows the basic characteristics of concrete: curing, storage condition, type of cement, and water/cement ratio, and Table 2 the dosage of concrete. Each specimen was reinforced with six, 42-grade, 3/8" diameter carbon steel bars (350 mm long), placed at depths of 15, 20 and 30 mm. Before placement in the specimens, the study area was marked on each bar, leaving a 150 mm long clear area in its center. The remaining bar was painted with epoxy paint and its end covered with isolating tape (Figure 1).

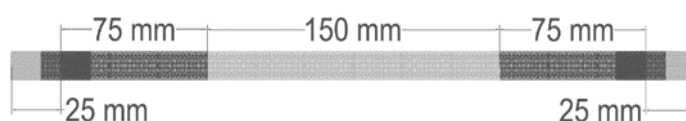


FIGURE 1

Marking of 150 mm unprotected area in the center of the bar (study zone).

The concrete was prepared using two water/cement (w/c) ratios: 0.45 (considered high-quality concrete in the study area), and 0.65 (considered low quality). Six specimens (old specimens - 2007) were cured in a moist room for 28 days and then kept in storage for 233 days. The second set of six (new specimens - 2017), was cured for seven days by direct application of tap water, common practice in the study area, and kept in storage for 27 days.

TABLE 2
Mixture proportions for concrete

Specimen	Curing process	Storage condition (without control of relative humidity and temperature)	Type of cement	Water/cement ratio
2007	Moist room for 28 days	Kept in semi-controlled storage for 233 days after curing	Portland Compound Cement (CPC30R)	0.45 (3 specimens) 0.65 (3 specimens)
2017	Direct tap water, few times/day for 7 days	Kept in semi-controlled storage for 27 days after curing		0.45 (3 specimens) 0.65 (3 specimens)

TABLE 2
Mixture proportions for concrete

Specimen	w/c ratio	Content of material (kg/m ³)				f _c - 28 days (kg/cm ²)
		cement	sand	gravel	water	
2007	0.45	408	681	800	287	N/A
	0.65	277	739	868	280	N/A
2017	0.45	456	638	757	205	231
	0.65	315	740	757	205	159

Before exposure, the inferior and superior sides of the specimens were painted with epoxy paint to limit the influence of aggressive agents to the lateral faces (Figure 2).

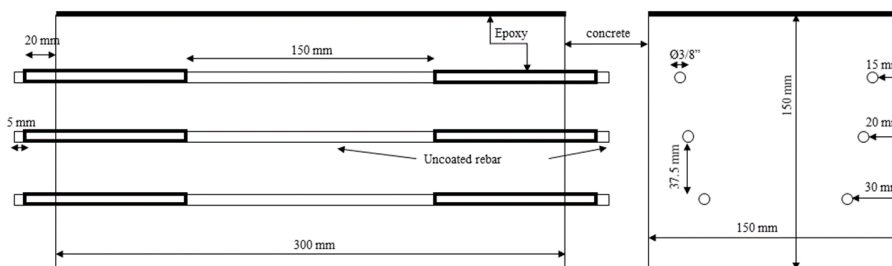


FIGURE 2
Specimen setup

Exposure environment

Specimens were exposed in a marine tropical environment on the north coast of the Yucatan Peninsula, Mexico. They were placed on the beach 50 m from the seashore and 0.95 m above the soil surface, positioning the area with most covering on the underside of the specimen. Their orientation was such that they were exposed to different regional climate conditions: prevailing winds out of the northeast (PW) and non-prevailing winds out of the southwest (NPW) (Figure 3). The first set of specimens (2007) was exposed 261 days after cast and the second set (2017) 34 days after cast.

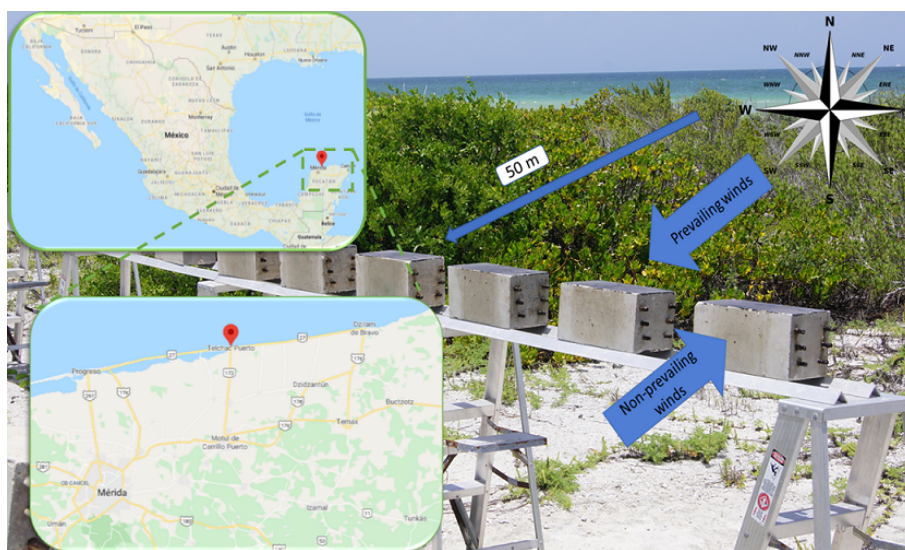


FIGURE 3
Setup of station for exposure of specimens.

Corrosion measurements

Steel reinforcement corrosion rate (i_{corr} , $\mu\text{A}/\text{cm}^2$), potential (E_{corr} , mV vs. Cu/CuSO₄), and concrete electrical resistance (R_s , k Ω) were measured with a commercial portable corrosimeter with a confinement system, which uses the linear polarization resistance technique (R_p) (Andrade & Alonso, 1996). The value for electrical resistivity (ρ , k Ω -cm) was calculated with equation (1), where D is the rebar diameter in cm.

$$\rho = 2R_s D \quad (1)$$

For the 2007 specimens, electrochemical measurements were taken approximately every 90 days for 557 days, while for the second set, 2017, they were taken every 14 days for 279 days.

RESULTS AND DISCUSSION

Passivation quality at initial ages

For the present purposes, initial age is defined as the time during which the specimens exhibited electrochemical behavior in the passivation zone. Interpretation of passivation zone data was facilitated by using a conservative limit value of $0.1 \mu\text{A}/\text{cm}^2$ with a range of $0.1 - 0.5 \mu\text{A}/\text{cm}^2$ for depassivation. Figure 4 is a typical plot used for i_{corr} recording versus time in small specimens exposed to a natural marine environment in different seasons. Even though three concrete covers (15 mm, 20 mm and 30 mm) were studied, this work discusses only data for the 20 mm concrete cover. This one is very common in the region's marine zone although applicable regulations demand thicker cover (NMX-C-530-ONNCCE, 2017). Other variables were the specimen's orientation (PW and NPW), included to identify any effect from wind, and concrete quality (w/c 0.45 and 0.65).

In the high-quality concrete (w/c 0.45) specimens, two data types were collected, one for exposure begun in 2007 (Table 1 and Figure 4a) and the other from 2017 (Table 1 and Figure 4b). The parameter of similar

quality among specimens was mainly given by the type of cement and the w/c ratio. This was due to the lack of more information in 2007. However, one way to minimize the variability of the CPC was by comparing elements from the same family (2007 or 2017), where the difference between elements was more noticeable, from the same family, due to more important factors such as curing and storage. In fact, the data for exposure since 2007 (Figure 4a) were essentially stable during the first 557 days. The first 233 days corresponded to storage in a semi-controlled environment. From day 233 to day 557, they were exposed to a tropical marine environment, but the passivation condition continued almost unchanged.

On the other hand, the data for exposure 2017 (Figure 4b), corresponding to 27 days in laboratory conditions and 211 days of natural exposure, started to show differences between PW and NPW. These figures are confirming a difference between both passivation qualities which is mainly due to the suitable curing and storage process of specimens for 233 days at the non-aggressive environment, without any aggressive agents, for specimens from 2007. Therefore, the quality of the passivation at initial ages allowed a subsequent better instantaneous electrochemical behavior for 2007 (Figure 4a) than for 2017 specimens (Figure 4b). The response of the low-quality concrete (w/c 0.65) was like that of the high-quality concrete (w/c 0.45) for both 2007 (Figure 4c) and 2017 (Figure 4d). However, instantaneous data for both w/c ratios seem to be a little bit dissimilar in 2017.

Although passivation zone behavior was stable, there were still doubts as to whether there were moments when, in response to changes in season (rainy and dry), the corrosion rate might have exceeded passivation zone values. As part of a larger project, specimens were built and exposed in 2017 for other purposes but clearly could serve to verify if this could influence the cumulative corrosion rate (ic_{um}), as reported recently (Castro-Borges et al., 2013, 2017; Melchers & Li, 2006; Pedrosa & Andrade, 2010; Rebolledo & Andrade, 2010). Unlike those built-in 2007, these new ones were exposed to the marine environment almost immediately after curing. In a comparison of the two processing over the same time scale (Figures 4a to 4d), the newer specimens exhibited behavior nearer the depassivation zone than the older specimens for all the studied variables. Curing, storage, and environmental variables affected, to a greater or lesser extent, the tendency towards depassivation. The parameters measured here helped to confirm passivation process quality at initial ages regardless of dependent variables (i.e. w/c and orientation). This will affect the planned specimen service life in terms of depassivation and propagation.

Effects of electrochemical measurement periodicity on the data processing

An obvious concern when taking frequent electrochemical measurements in passive steel reinforcement in concrete is possible damaging polarization which could affect the real ic_{orr} value. In modern times the damaging polarization is controlled, for example, by measuring the ic_{orr} through the linear polarization resistance with sweeping rates on the order of hundredths of millivolts and with cathodic polarization close to 20 mV at the most. Therefore, and within these ranges, a higher measurement periodicity in the polarizations to measure corrosion is not detrimental to the studied rebar, and especially in field tests. This general subject has been addressed extensively, but no previous studies (Andrade & Gonzalez, 1978; Andrade et al, 1986; Clément et al, 2012; Hansson et al, 2012) contain information about the possible effect of the periodicity of the measurements to interpret the meaning of the ic_{um} under natural conditions. The most obvious difference between laboratory and field studies is that under laboratory conditions measurement intervals can be minutes, whereas under natural conditions intervals are normally on the order of days. The possible effect of polarization periodicity at initial ages in cumulative corrosion could occur in a natural environment in response to peaks in temperature and humidity, which would allow interpreting more accurately the changes in observed trends. Therefore, determining if electrochemical measurement periodicity influences the interpretation of passivation/depassivation in reinforced concrete with ic_{um} under a natural environment during the initiation stage (passivation zone) is therefore quite technically and

economically important. The icorr confirm that measurement periodicity in the older or newer specimens did not affect data credibility since both specimen series exhibited the same tendencies (Figure 4). However, when using icum, electrochemical measurement periodicity could create differences among trends that, if known, could be equivalent to tolerance levels used in destructive tests such as those used to compare electrochemical losses to gravimetric losses (Andrade & Martinez, 2005). To approximate these differences the data in Figure 4 were analyzed, resulting in the data in Figure 5.

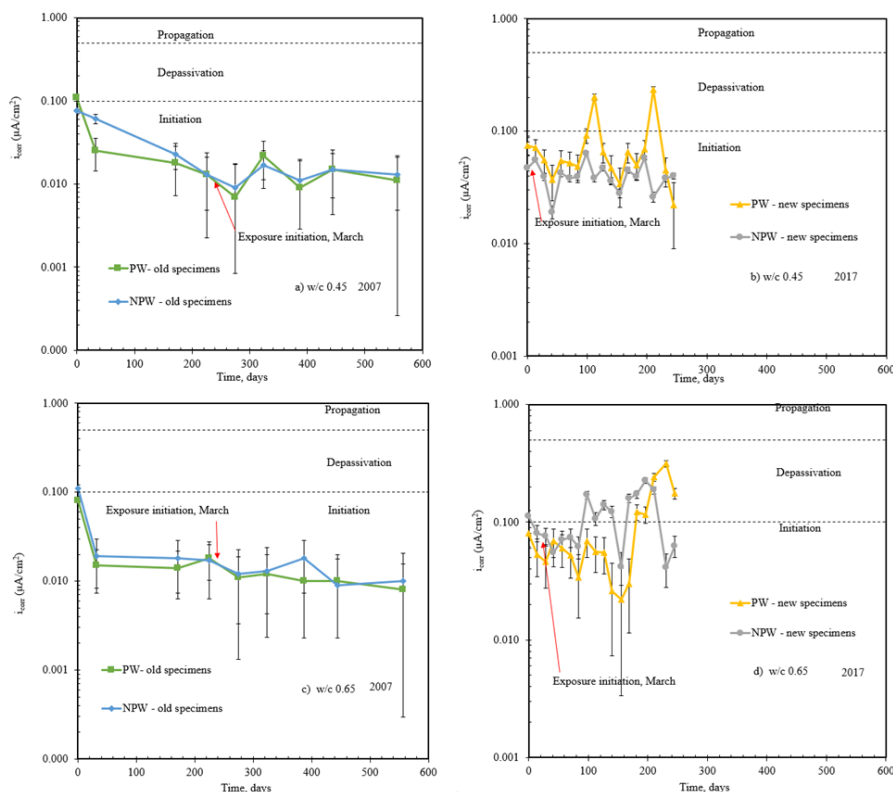


FIGURE 4

Instantaneous corrosion rate vs. time: a) w/c 0.45, march 2007; b) w/c 0.45, march 2017; c) w/c 0.65, march 2007; d) w/c 0.65, march 2017.

Figure 5 shows a relation between sixteen measurements from March 2017 and four measurements from March 2007 that correspond, in both cases, to an exposure period of 211 days just after the beginning of the exposure to the environment. This relation $16/4 = 4/1$ was taken as our reference to discuss the effect of the measurement periodicity on the interpretation of icum. The term icum can be defined as the sum of previous corrosion rates at each specific period (Castro-Borges et al., 2013) as given by equation (2):

$$i_{cum} = \frac{(t_f - t_i) * (i_{corrf} - i_{corri})}{2} \quad (2)$$

Where:

t_f = Final day of the considered period

t_i = Initial day of the considerer period

i_{corrf} = Final icorr of the considered period

i_{corri} = Initial icorr of the considered period

i_{cumi} = Previously accumulated corrosion rate

When the Figure 5 data were analyzed using a 4/1 electrochemical measurement periodicity ratio, the difference between PW and NPW was up to 100% on the i_{cum} value on specimens that did not have time to consolidate their passivation correctly (2017). Moreover, the tendencies between PW and NPW were inverted when considering the quality of concrete (w/c 0.45 or 0.65). This coincides and could be compared with tolerance due to errors that are attributed to data collection when electrochemical losses are compared to gravimetric losses (Andrade & Martinez, 2005). However, this did not happen with those (2007) that were stored for 233 days and then exposed for 211 days, and that influenced the i_{cum} between PW and NPW of up to 40%. In this case, it could mean the possibility of having a sharper tolerance if electrochemical losses were compared to gravimetric ones. The values of 100% and 40% probably mean a characteristic difference between specimens with not consolidated or consolidated passivation just before the beginning of the depassivation stage for the 4/1 ratio of measurement periodicity. The experimental and data collection limitations could slightly influence the results, so they do not necessarily apply under different conditions than those reported in this work. In equivalent form, if measurement ratios $> 4/1$ were found and give even higher numerical differences in cumulative corrosion would place data credibility in question at initial ages (passivation zone). This is supported by the fact that the values of i_{corr} (Figure 4) begin to show limit values close to $0.1 \mu A/cm^2$, the 100% of tolerance when comparing electrochemical vs gravimetric losses, and the inverted position of the PW and NPW tendencies in Figure 5.

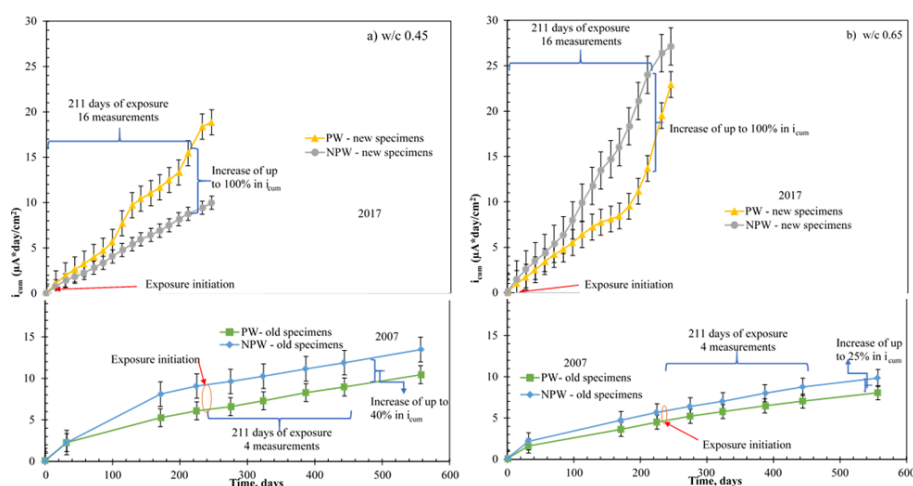


FIGURE 5
Cumulative corrosion rate vs. time: a) w/c 0.45; b) w/c 0.65.

Margin of uncertainty in cumulative corrosion

In Figure 6, the i_{cum} of the specimens made in 2007 corresponds to the moment just before exposure to the tropical marine environment. These specimens were under semi-controlled conditions (storage in a laboratory, sheltered from the rain and the sun and without control of relative humidity and temperature) for the entire period. No significant variations were noted at a measurement periodicity of approximately three months (Figure 4). Considering that they were electrochemically stable in the passivation zone, cumulative corrosion immediately before exposure to the natural environment differed between surfaces (PW and NPW), from $1 \mu A \cdot day/cm^2$ for w/c 0.65 to $5 \mu A \cdot day/cm^2$ for w/c 0.45. Ahmad (2003) mentions that the lower the cement content, the lower the plastic consistency in the mix, which causes that the passivation of the steel does not occur uniformly, resulting in the depassivation; therefore, the results found could represent the margin of uncertainty when interpreting i_{cum} at initial ages (passivation zone) and these differences (1

$\mu\text{A}\cdot\text{day}/\text{cm}^2$ for w/c 0.65 to $5 \mu\text{A}\cdot\text{day}/\text{cm}^2$ for w/c 0.45) can be deemed as characteristics of the materials used.

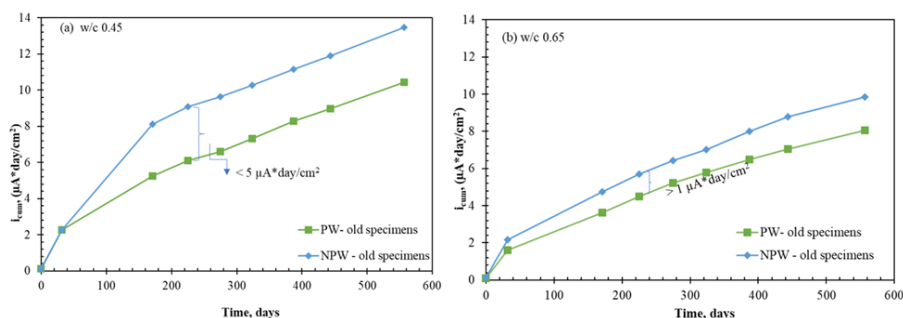


FIGURE 6.

Uncertainty in cumulative corrosion rate vs. time: a) w/c 0.45; b) w/c 0.65.

CONCLUSIONS

The reported results apply to these specific study conditions. Therefore, they can be extrapolated taking the appropriate precautions to other situations. In the present data, the parameters used in consolidating passivation quality at initial ages (i.e. curing, storage, and environmental variables) affect the tendency to depassivation regardless of the dependent variables (w/c ratio and orientation). This can affect the planned service life.

The reported results apply to these specific study conditions. Therefore, they can be extrapolated taking the appropriate precautions to other situations. In the present data, the parameters used in consolidating passivation quality at initial ages (i.e. curing, storage, and environmental variables) affect the tendency to depassivation regardless of the dependent variables (w/c ratio and orientation). This can affect the planned service life.

As and expected, the polarization produced by electrochemical measurement periodicity (days) in a tropical marine environment did not influence reinforced concrete passivation/depassivation during the initiation stage (passivation zone).

When electrochemical measurement periodicity was calculated at a 4/1 ratio, it was found that it could numerically influence the cumulative corrosion value by up to 100%. This is equivalent to the tolerance used in destructive tests comparing electrochemical losses to gravimetric losses.

Even though it is considered a largely qualitative method, i_{cum} allows the detection of changes in the electrochemical behavior of reinforced concrete in the initiation stage (passivation). From a quantitative perspective, storage type after curing and before exposure to the environment, even in the initiation stage (before depassivation), had an influence that differed from $1 \mu\text{A}\cdot\text{day}/\text{cm}^2$ for 0.65 w/c to $5 \mu\text{A}\cdot\text{day}/\text{cm}^2$ for 0.45 w/c in i_{cum} . This is equivalent to the margin of uncertainty when interpreting cumulative corrosion in the studied concrete qualities.

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ADDITIONAL INFORMATION

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