

DYNA

ISSN: 0012-7353

Universidad Nacional de Colombia

Torres-Lozada, Patricia; Amezquita-Marroquín, Claudia Patricia; Agudelo-Martínez, Karen Daniela; Ortiz-Benítez, Natalia; Martínez-Ducuara, David Santiago Evaluation of turbidity and dissolved organic matter removal through double filtration technology with activated carbon DYNA, vol. 85, no. 205, 2018, April-June, pp. 234-239

Universidad Nacional de Colombia

DOI: https://doi.org/10.15446/dyna.v85n205.65488

Available in: https://www.redalyc.org/articulo.oa?id=49657889030



Complete issue

More information about this article

Journal's webpage in redalyc.org



Scientific Information System Redalyc

Network of Scientific Journals from Latin America and the Caribbean, Spain and Portugal

Project academic non-profit, developed under the open access initiative







Evaluation of turbidity and dissolved organic matter removal through double filtration technology with activated carbon

Patricia Torres-Lozada, Claudia Patricia Amezquita-Marroquín, Karen Daniela Agudelo-Martínez, Natalia Ortiz-Benítez & David Santiago Martínez-Ducuara

Study and Control of Environmental Pollution (ECCA) Research Group, Engineering Faculty, Universidad del Valle, Cali, Colombia. patricia.torres@correounivalle.edu.co, claudia.amezquita@correounivalle.edu.co, karen.agudelo@correounivalle.edu.co, natalia.ortiz.benitez@correounivalle.edu.co, david.martinez.ducuara@correounivalle.edu.co

Received: June 06th, 2017. Received in revised form: February 12th, 2018. Accepted: March 15th, 2018.

Abstract

The decrease of physicochemical and microbiological quality of surface supply source for human consumption, requires specific complementary treatments to ensure the supply of safe drinking water. This study evaluated the double filtration technology with two types of Granular Activated Carbon (GAC: vegetable-coconut and mineral-bituminous shells, respectively), to determine its influence in reducing turbidity and mainly dissolved organic matter (UV $_{254}$). We used filtered water from a Conventional Water Treatment Plant (WTP) and the second filtration stage were made using continuous flow gravity columns at laboratory-scale with different percentages of GAC:sand (100:0, 80:20, 50:50, 30:70, and 0:100). While the sand filter presented the best turbidity removal efficiency, the CAG filters were more efficient at UV $_{254}$ removal, being most efficient the filters with greater percentage of GAC. The results demonstrate that the use of double filtration technology with GAC can be an efficient alternative for the removal of organic matter to obtain safe drinking water.

Keywords: drinking water; granular activated carbon; double filtration; organic material; filter medium; turbidity.

Evaluación de la remoción de turbiedad y materia orgánica disuelta mediante la tecnología de doble filtración con carbón activado

Resumer

La reducción de la calidad fisicoquímica y microbiológica de las fuentes superficiales de suministro para consumo humano, requiere utilizar tratamientos complementarios para garantizar el suministro de agua segura. En este estudio se evaluó la doble filtración con carbón activado granular (CAG: vegetal- cáscara de coco y mineral-bituminoso), para evaluar la reducción de turbiedad y materia orgánica disuelta (UV₂₅₄). Se empleó agua filtrada de una Planta de Tratamiento Convencional–PTAP; la segunda filtración se realizó en columnas de laboratorio de flujo continuo a gravedad, con porcentajes CAG:arena 100:0, 80:20, 50:50, 30:70, 0:100. Mientras el filtro de arena fue más eficiente en remoción de turbiedad, las configuraciones con CAG lo fueron en la remoción de UV₂₅₄, siendo más eficientes los filtros con mayor porcentaje de CAG. Los resultados demuestran que la doble filtración empleando CAG, puede ser una alternativa eficiente para la remoción de materia orgánica y la producción de un agua potable segura.

Palabras clave: agua potable; carbón activado granular; doble filtración; materia orgánica; medio filtrante; Turbiedad.

1. Introduction

In developing countries, the majority of sources for drinking water supply systems are affected by natural phenomena and anthropogenic effects that deteriorate their quality [1]. Limitations on the efficiencies of the production processes result in drinking water that complies with established quality standards [2-4]. Various treatment technologies, which use several combinations of processes, are available for the purification of surface water, including coagulation/flocculation, sedimentation, filtration, disinfection and pH stabilization [2,5,6].

Filtration is a process that is always used in water treatment technology. It consists of the separation or removal of particles

How to cite: Torres-Lozada, P., Amezquita-Marroquín, C.P., Agudelo-Martínez K.D., Ortiz-Benítez N. and Martínez-Ducuara, D.S., Evaluation of turbidity and dissolved organic matter removal through double filtration technology with activated carbon. DYNA, 85(205), pp. 234-239, June, 2018.

in a liquid by flowing through a porous bed [2,5]. Conventional treatment technology with sand or sand and anthracite rapid filters has allowed compliance with Colombian Resolution 2115 of 2007, which established a maximum permissible turbidity in drinking water of 2 NTU [7]. However, several authors recommend that to ensure the minimum risk of drinking water, the turbidity of the filtered water should be in the range of 0.1 to 0.5 NTU to eliminate or deactivate pathogenic microorganisms [3,8,9].

Several typical filtration systems do not have the ability to effectively remove turbidity and dissolved organic matter, and it is necessary to evaluate other treatments to improve the quality of drinking water [3]. Double filtration is a treatment technology that consists of a first stage of filtration in which the water flows through a granular medium with a high capacity of solids removal. This process reduces the turbidity and is followed by a second stage of filtration for polishing [10]. In the last stage, mineral or vegetal GAC has been used as a filtration medium for water purification due to its characteristics, which allow it to remove turbidity and dissolved organic matter and increase the treatment efficiency [11-14].

Turbidity is a key parameter for evaluating the efficiency of filtration because it is simple and rapid to quantify, and it is indirectly related to the particles in the water, which in turn are associated with bacteria, protozoa and viruses [15,16]. Organic matter is associated with aesthetic impacts on water quality, with the formation of disinfection byproducts that may become carcinogenic and/or with synthetic organic compounds (e.g., pesticides, personal care products, pharmaceuticals) that are difficult to remove with conventional drinking water treatments [6,17].

GAC is one of the most widely used adsorbents in water treatment [18]. According to Wiecheteck et al. [19] and Silva et al. [20] double filtration has several favorable characteristics, such as greater solids retention and a more effective response in the removal of organic matter. Authors such as Bundy et al. [21] have used conventional sand and anthracite filtration and secondary filtration with GAC and achieved a turbidity reduction to less than 1 NTU and a removal efficiency of pharmaceutical compounds on the order of 95%. Silva et al. [20] also used GAC to eliminate cyanobacteria, color, organic matter and halogenated organic by-products. Pham et al. [22] used coconut husk GAC and obtained removal efficiencies of turbidity (97%) and organic matter (COD:68%) and a pesticide adsorption capacity greater than 50%. Thiel et al. [17] concluded that sand:GAC filters are effective at removing precursor organic matter from disinfection products and generate effluents with turbidities less than 0.3 NTU.

Based on these investigations, this study evaluates at laboratory scale, the secondary filtration with different proportions of mineral and vegetal activated carbon using filtered water from a conventional water treatment plant in the city of Cali. The second filtration was evaluated to determine its effect on the removal of turbidity and dissolved organic matter, which were measured as UV_{254} .

2. Matherials and methods

2.1. Experimental unit

Twenty-four laboratory glass filters with a nominal diameter of 25 mm, an internal diameter of 19 mm and a length

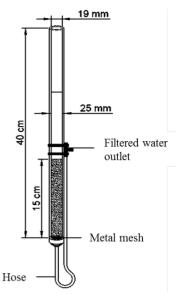


Figure 1. Experimental unit

Table 1.

Quality characteristics of the filtered water used for the study.

PM	pH	Color	Conductivity (µs/cm)	Turbidity	UV ₂₅₄
filtered	(units)	(UPC)		(NTU)	(cm- ¹)
WTP water	5.87	3	156.3	0.44	0.034

Source:

of 40 cm were used. The lower part of the filter was supported by a metal mesh to prevent clogging of the filter outlet and loss of the filter medium, which was 15 cm high [5,23,24]. To provide a hydraulic load to ensure that water was present throughout the filter medium, the effluent was collected through a silicone hose attached to the bottom of the filter. The other end of the hose was located above the upper part of the filter medium [5]. The filtration columns were placed in a support structure, which allowed proper operation of the filters (Fig. 1).

The experimental unit was fed by a multi-partition distribution system. Gravity flow was distributed to each laboratory filter through silicone hoses, which had flow control valves at their ends to allow the flow to be distributed through constant dripping to the filters. This distribution system was supplied from a temporary storage tank in which the water level was kept constant to avoid variations in the inlet flows to the filters.

2.2. Water used in the study

The laboratory-scale study used filtered water from the Puerto Mallarino WTP, which had undergone coagulation, flocculation, clarification and downstream sand and anthracite filtration [25,26]. Table 1 shows the characteristics of the filtered water.

2.3. Filter media

The characteristics of the GAC and sand filter media used in the test are given in Table 2.

Table 2. Characteristics of the filter media used in the test.

Characteristics	GAO	Sand Medium	
Raw material	Coconut shell	Bituminous coal	Silica
Amount of iodine (mg/g)	900	850	
Ash (% by weight)	5	8	
Bulk Density (g/cc)	0.48	0.58	1.5
Effective Size (mm)	0.8-1.2	0.55-0.75	0.61
Coefficient of Uniformity	2.1	1.9	1.5

Source:

Table 3.
Configurations of GACVEG, GACMIN and sand filter media used in the study.

Configuration	GACVEG	Sand	
100VEG	100% (15 cm)	0%	
80VEG	80% (12 cm)	20% (3 cm)	
50VEG	50% (7.5 cm)	50% (7.5 cm)	
30VEG	30% (3 cm)	70% (12 cm)	
Configuration	GACMIN	Sand	
100MIN	100% (15 cm)	0%	
80 MIN	80% (12 cm)	20% (3 cm)	
50 MIN	50% (7.5 cm)	50% (7.5 cm)	
30 MIN	30% (3 cm)	70% (12 cm)	
Configuration	CORE	Sand	
SAND	0%	100% (15 cm)	

Source:

Table 4. Variables and methods used to characterize water during the test.

Variables	Units	Method
pН	Units	4500-H+-B
Conductivity	μs/cm	2510-В
Turbidity	NTU	2130-В
UV_{254}	cm-1	5310-B

Source:

The configurations correspond to the height proportions of the filter media that consist of GACVEG and sand, GACMIN and sand and only sand. These configurations are presented in Table 3 and were analyzed in triplicate.

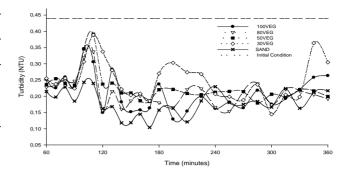
2.4. Operation of the experimental units and process monitoring

Filtered water from the WTP, was distributed to each laboratory filters at a constant filtration rate of 61 m³/m²d for a flow rate of 12 ml/min and a filtration time of six hours [5, 24]. The follow-up variables are shown in Table 4 and were measured every 5 minutes during the first half hour of filtration. Subsequently, they were measured at 10 minute intervals for the remainder of the first three hours and every 15 minutes in the last three hours of the test. The physical and chemical variables were measured according to the guidelines established in the standard method [27].

3. Results and discussion

3.1. Influence on turbidity removal

Turbidity is an indirect parameter that indicates the potential of microbiological risk in drinking water. Low values of turbidity



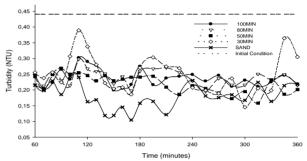


Figure 2. Behavior of the turbidity over time for the two types of activated carbon and sand with their respective configurations.

in filtered water (<0.30 NTU) indicate greater efficiencies in the removal of protozoa (Giardia and Cryptosporidium) during filtration and favor the elimination of bacteria and viruses during disinfection [15]. Fig. 2 shows the behavior of the turbidity during the test for the two types of carbon (vegetable/mineral) and sand with their respective filter media configurations.

Fig. 2 shows that all of the media configurations resulted in similar turbidity behavior as a function of time. The turbidity values are below the initial turbidity value obtained in the first filtration stage. As indicated by Wiecheteck et al. [19] and Silva et al. [20], double filtration provides greater turbidity removal than conventional filtration.

Ninety-seven percent of the data for all configurations resulted in turbidity values less than 0.3 NTU, which is the threshold value of the WHO [2] and the EPA [15] before disinfection to eliminate chlorine-resistant pathogens and ensure the effective elimination of Giardia, Cryptosporidium, and other material. All of the turbidity values measured for each configuration in this study complied with the national regulations for drinking water (<2 NTU) [7].

Graese et al. [28] founded turbidity <0.2 NTU in the effluent with GAC filtration, which is consistent with this study. The median values for the filters with the GACVEG configurations were between 0.199 and 0.238 NTU, those for the GACMIN filters were between 0.206 and 0.236 NTU, and those with sand had the lowest value of 0.181 NTU.

The median values obtained for each configuration show that the configuration with the sand filter resulted in the best reduction of turbidity; however, all configurations with GAC provided effective turbidity removal.

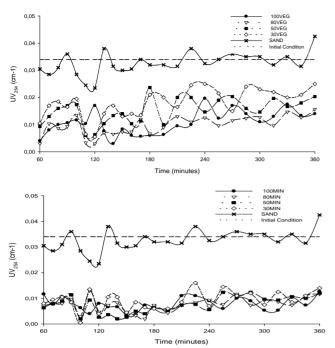


Figure 3. Behavior of UV_{254} over time for the two types of activated carbon and their respective configurations.

Although the GAC configurations showed lower turbidity removal than the sand filter, these results demonstrate that the implementation of activated carbon ensured the quality of the final effluent.

3.2. Influence on the removal of organic matter measured as UV₂₅₄

Organic matter is an important constituent of water that affects the performance of treatments in drinking water processes and the quality of drinking water. As a result, it requires the extensive use of coagulants, oxidants and disinfectants in addition to being a precursor to the formation of disinfection byproducts [29]. Several studies have used the UV₂₅₄ absorbance test as a parameter to evaluate the efficiency of the GAC filtration process for the removal of dissolved organic matter, which is capable of absorbing UV light [6, 20, 29]. Figure 3 shows the behavior of UV₂₅₄ during the tests for the two types of carbon (vegetable/mineral) and sand with their respective filter medium configurations.

Fig. 3 shows that the configurations with GACMIN and GACVEG were below the UV₂₅₄ absorbance value reported in the first filtration stage, whereas the sand filter remained very similar to the initial UV₂₅₄ value. Similar results were obtained by other studies, which concluded that sand does not effectively remove compounds associated with organic matter [30,31].

Median values of UV₂₅₄ between 0.010 and 0.019 cm⁻¹ were obtained with the GACVEG configurations, median values between 0.007 and 0.008 cm⁻¹ were obtained with the GACMIN configurations, and a median of 0.032 cm⁻¹ was obtained with the sand. The results indicate that the sand did

not significantly affect the reduction of organic matter because its values oscillated near the initial absorbance of 0.034 cm-1, and the configurations with GACMIN had a greater effect. The GAC filters clearly yielded a reduction of organic compounds as measured by UV₂₅₄. Therefore, the use of granular activated carbon as a filter media is recommended not only for reducing odor and flavor but also in the adsorption of organic compounds [9].

Silva et al. [20] demonstrated the removal of UV_{254} using GAC in double filtration, and Pham et al. [22] achieved removal efficiencies of organic matter greater than 68%. These results were similar to those found in this study; the removal percentages for the GACVEG configurations ranged from 46% to 72%. The GACMIN configurations resulted in the highest removal percentages, between 75% and 78%, whereas the sand configuration only removed 6% of the organic matter.

The results indicate that filters with GAC in double filtration technology can remove organic substances, which reduces the possible risk associated with the formation of disinfection byproducts [17,32].

Additionally, Kim and Kang [33] found that replacing sand filters with dual-medium GAC-sand filters represents an ideal choice for the removal of organic matter compared to turbidity removal. This was confirmed by the results of this study, which showed small differences in the turbidity decrease between the GAC and sand configurations, whereas the configurations of the GAC filters eliminated organic matter more effectively than the sand filter alone.

4. Conclusions

Although the sand filter achieved a higher turbidity removal efficiency, all configurations with GAC provided effective removal that complied with the limit of 0.3 NTU recommended by the WHO and EPA to mitigate microbiological risk and with the limit of <2 NTU provided by Colombian Resolution 2115 of 2007.

For organic matter, which was measured as UV_{254} , all configurations containing GAC were more efficient than the sand, which removed only 6% of the organic matter. A greater proportion of GACVEG resulted in a greater percentage of reduction of UV_{254} in the effluent (100VEG: 70%, 80VEG: 70%, 50 VEG: 53%, 30VEG: 46%), and GACMIN obtained similar and higher efficiencies than GACVEG (100MIN: 78%, 80MIN: 77%, 50MIN: 77%, 30MIN: 75%).

Although Resolution 2115 of 2007 does not provide a maximum allowable amount of organic matter in drinking water in terms of UV₂₅₄, the removal efficiencies for organic matter of the configurations with GAC were greater, which confirmed that GAC is suitable as a filter medium not only for the reduction of odor and flavor but also in the adsorption of organic compounds.

Double filtration has several favorable characteristics, including a greater retention of solids and a more effective removal of organic matter, which make it an efficient technology to ensure the quality of the final effluent, minimize risk and reduce the limitations in the treatment of water for human consumption.

Acknowledgements

The authors thank the Universidad del Valle and to the Department of Science, Technology and Research (COLCIENCIAS) for financing the research project "Evaluation of enhanced coagulation for the removal of pesticides in water as a strategy for mitigating environmental factors with incidence in health - CI 2937" and the doctoral formation to Eng. Claudia Patricia Amézquita Marroquín (National Doctorate Convocation 567).

References

- [1] Perez-Vidal, A., Torres-Lozada, P. and Escobar-Rivera, J.C., Hazard identification in watersheds based on water safety plan approach: case study of Cali-Colombia. Environmental Engineering and Management Journal [Online]. 15(4), pp. 861-872, 2016. [date of reference June of 2016]. Available at: http://eemj.eu/index.php/EEMJ/article/view/2903
- [2] World Health Organization (WHO). Guidelines for drinking-water quality. Genéva: WHO, 2011.
- [3] World Health Organization (WHO). Guidelines for drinking water quality. Genéva: WHO, 2006.
- [4] Jimenez, B. and Rose, J., Urban water security: managing risks. Boca Raton: Taylor & Francis, UNESCO-IHP, 2009.
- [5] Di Bernardo, L., Di Bernardo, A. and Nogueira, P., Tratabilidade de agua e dos residuos gerados em estacoes de tratamiento de agua. Sao Carlos: LDiBe editora, 2011.
- [6] Crittenden, J.C., Trussell, R.R., Hand, D.W., Howe, K.J. and Tchobanoglous, G. Granular filtration. In Crittenden, et al. MWH's Water Treatment: Principles and Design, 3rd Edition, USA, 2012. pp. 727-818.
- [7] Minambiente (Ministerio de Ambiente, Vivienda y Desarrollo Territorial). Resolución 2115 del 22 de junio de 2007. Por medio de la cual se señalan características, instrumentos básicos y frecuencias del sistema de control y vigilancia para la calidad del agua para consumo humano. Minambiente, Bogotá. 2007.
- [8] Environmental Protection Agency (EPA). Estándares del reglamento nacional primario de agua potable [on line]. EPA, USA, 2000. [date of reference June of 2016]. Available at: http://water.epa.gov/drink/agua/estandares.cfm
- [9] American Water Works Asociation (AWWA). Water quality & treatment. A Handbook on Drinking Water. USA: McGraw-Hill, 2011.
- [10] Sandobal-Paz, L., Marques, E., Butti, F. and Akiko, J., Avaliação técnico-econômica da tecnologia de tratamento de água de dupla filtração. Eng. Sanit. Ambient., 20(4), pp. 525-532, 2015. DOI: 110.1590/S1413-41522015020040129909.
- [11] Gupta, V.K., and Ali, I., Water treatment for organic pollutants by adsorption technology. In. Gupta, V.K. and Ali, I., Environmental water: advances in treatment, remediation and recycling, Elsevier, 2013, pp. 29-91. DOI:10.1016/B978-0-444-59399-3.00002-7.
- [12] Karanfil, T., Activated carbon adsorption in drinking water treatment. In Bandosz, T.J., Activated Carbon Surfaces in Environmental Remediation, Elsevier, 2006, pp. 345-373. DOI: 10.1016/S1573-4285(06)80016-5
- [13] Przepiórski, J., Activated carbon filters and their industrial applications. In Bandosz, T.J., Activated Carbon Surfaces in Environmental Remediation, Elsevier, 2006, pp.421-474. DOI: 10.1016/S1573-4285(06)80018-9.
- [14] Ratnayaka, D.D., Brandt, M.J. and Johnson, K.M., Specialized and advanced water treatment processes. In Ratnayaka, D.D., Brandt, M.J., Johnson, K M., Water Supply, Elsevier, 2009, pp. 365-423. DOI: 10.1016/B978-0-7506-6843-9.00018-4.
- [15] Environmental Protection Agency (EPA). National primary drinking water regulations. EPA 816-F-09-004 [on line]. EPA, USA, 2009. [date of reference June of 2016]. Available at: http://pbadupws.nrc.gov/docs/ML1307/ML13078A040.pdf

- [16] CEPIS and OPS. Tratamiento de agua para consumo humano: plantas de filtración rápida [en líena]. CEPIS, OPS, Perú, 2004. [dato de consulta: June of 2016]. Disponible en: http://bibliotecavirtual.minam.gob.pe/biam/bitstream/id/5657/BIV00 012.pdf
- [17] Thiel, P., Zappia, L., Franzmann, P., Warton, B., Alessandrino, M., Heitz, A., Nolan, P., Scott, D., Hiller, B. and Masters, D., Activated carbon VS anthracite as primary dual media filters a pilot plant study. 69th Annual Water Industry Engineers and Operators' Conference, Bendigo, 2006. [online]. Available at: http://www.wioa.org.au/conference_papers/06_vic/documents/PetaThiel1.pdf
- [18] Bhatnagar, A., Hogland, W., Marques, M. and Sillanpääc, M., An overview of the modification methods of activated carbon for its water treatment applications. Chemical Engineering Journal, 219, pp. 499-511, 2013. DOI: 10.1016/j.cej.2012.12.038
- [19] Wiecheteck, G., Da silva, B. and Di bernardo, L., Remoção de substâncias húmicas utilizando dupla Filtração com filtro ascendente de areia grossa Ou de pedregulho. Engenharia Sanitaria e Ambiental, 9(2), pp. 156-203, 2004.
- [20] Silva, G.G., Naval, L.P., Di Bernardo, L. and Dantas, A.D.B., Tratamento de água de reservatórios por dupla filtração, oxidação e adsorção em carvão ativado granular. Engenharia Sanitaria e Ambiental, 17(1), pp. 71-80, 2012. DOI: 10.1590/S1413-41522012000100011.
- [21] Bundy, M., Doucette, W., McNeill, L. and Ericson, J.F., Removal of pharmaceuticals and related compounds by a bench-scale drinking water treatment system. Journal of Water Supply: Research and Technology—AQUA, 56(2), pp. 105-115, 2007. DOI: 10.2166/aqua.2007.091
- [22] Pham, T., Nguyen, V. and Van der Bruggen, B., Pilot-scale evaluation of GAC adsorption using low-cost, high-performance materials for removal of pesticides and organic matter in drinking water production. Journal of Environmental Engineering, [online]. 139(7), pp. 958-965, 2013. Available at: http://ascelibrary.org/doi/abs/10.1061/%28ASCE%29EE.1943-787 0.0000704#sthash.voYLqbDx.dpuf
- [23] Ho, L. and Newcombe, G., Granular activated carbon adsorption of 2-Methylisoborneol MIB Pilot- and laboratory-scale evaluations. Journal of environmental engineering, 136(9), pp. 965-974, 2010. DOI: 10.1061/(ASCE)EE.1943-7870.0000231#sthash.JdMlljtE.dpuf
- [24] Rigobello-Sloboda, E., Di Bernardo-Dantas, A., Di Bernardo, L. and Vieira, E.M., Removal of diclofenac by conventional drinking water treatment processes and granular activated carbon filtration. Chemosphere, 92(2), pp. 184-191, 2013. DOI: 10.1016/j.chemosphere.2013.03.010
- [25] Montoya, C., Loaiza, D., Torres, P. y Cruz, C., Efecto del incremento en la turbiedad del agua cruda sobre la eficiencia de procesos convencionales de potabilización. Revista EIA [online]. 8(16), pp. 137-148, 2011. [date of reference June of 2016]. Available at: http://repository.eia.edu.co/handle/11190/148
- [26] Perea, L.M., Torres, P., Cruz-Velez, C.H. y Escobar-Rivera, J.C., Influencia de la configuración del medio filtrnate sobre el proceso de filtración a tasa constante del agua clarificada del Rio Cauca. Revista de Ingeniería [Online]. 38, pp. 38-44, 2013. [date of reference June of 2016]. Available at: http://www.scielo.org.co/scielo.php?pid=S0121-49932013000100007&script=sci_abstract&tlng=es
- [27] APHA, AWWA, WEF. Standard methods for water and wastewater examination. USA: American Public Health Association, 2012.
- [28] Graese, S., Snoeyink, V. and Lee, R., Granular activated carbon filter-adsorber systems. Journal American Water Works Association [Online]. 79(12), pp. 64-74, 1987. [date of reference June of 2016]. Available at: https://www.awwa.org/publications/journal-awwa/abstract/articleid/11937.aspx. DOI: 10.1002/j.1551-8833.1987.tb02961.x
- [29] Zouboulisa, A., Traskasa, G. and Samarasb, P., Comparison of single and dual media filtration in a full-scale drinking water treatment plant, Desalination, 213(1–3), pp. 334-342, 2007. DOI: 10.1016/j.desal.2006.02.102
- [30] Lopes, M.P., Matos, C.T., Pereira, V.J., Benoliel, M.J., Valério, M.E., Bucha, L.B., Rodrigues, A., Penetra, A., Ferreira, E., Cardoso, V., Reis, M. and Crespo, J., Production of drinking water using a multi-

- barrier approach integrating nanofiltration: a pilot scale study, Separation and Purification Technology, 119, pp. 112-122, 2013. DOI: 10.1016/j.seppur.2013.09.002
- [31] Grace, M.A., Healy, M.G. and Clifford, E., Use of industrial byproducts and natural media to adsorb nutrients, metals and organic carbon from drinking water, The Science of the Total Environment, 518-519, pp. 491–497, 2015. doi:10.1016/j.scitotenv.2015.02.075
- [32] Rosero, M., Latorre, J., Torres, W. y Delgado, L. Presencia de materia orgánica y subproductos de la desinfección con cloro. Caso sistema de tratamiento de agua para consumo humano, Puerto Mallarino, Cali-Colombia, MSc. Thesis, Maestría en Ingeniería Sanitaria y Ambiental - Escuela de Ingeniería de Recursos Naturales y del Ambiente, Universidad del Valle, Colombia, 2005.
- [33] Kim, J. and Kang, B., DBPs removal in GAC filter-adsorber, Journal Water Research, 42(1-2), pp. 145-152, 2008. 10.1016/j.watres.2007.07.040
- P. Torres-Lozada, received the BSc. of Sanitary Engineer in 1982 from the Universidad del Valle, Cali, Colombia, MSc. (1993) and PhD (2001) in Civil Engineering: Hydraulics and Sanitation from the Universidade de São Paulo, Brazil. From 1995 to date, she is linked to the Universidad del Valle as full professor. She is categorized by Colciencias as a senior researcher and is a Leader of the Environmental Pollution Study and Control - ECCA Research Group with category A1. The research areas are water quality, water treatment for human consumption, wastewater treatment and solid waste. ORCID: 0000-0001-9323-6677
- C.P. Amezquita-Marroquín, received the BSc. of Sanitary Engineer in 2009, the MSc. degree in sanitary and environmental engineering in 2013 all of them from the Universidad del Valle, Cali, Colombia, She has worked on water treatment projects, currently a PhD student in engineering of the Universidad del Valle and she belongs to ECCA Research Group -Universidad del Valle, Engineering Faculty. ORCID: 0000-0002-0094-2204

K.D. Agudelo-Martínez, received the BSc. of Sanitary and Environmental Engineer in 2016 from the Universidad del Valle, Cali, Colombia. ORCID: 0000-0002-2728-2998

N. Ortiz-Benítez, received the BSc. of Sanitary and Environmental Engineer in 2016 from the Universidad del Valle, Cali, Colombia. ORCID: 0000-0003-1614-1396

D.S. Martínez-Ducuara, received the BSc. of Sanitary and Environmental Engineer in 2016 from the Universidad del Valle, Cali, Colombia. ORCID: 0000-0002-8803-3864



UNIVERSIDAD NACIONAL DE COLOMBIA

SEDE MEDELLÍN FACULTAD DE MINAS

Área Curricular de Medio Ambiente

Oferta de Posgrados

Especialización en Aprovechamiento de Recursos Hidráulicos Especialización en Gestión Ambiental Maestría en Ingeniería Recursos Hidráulicos Maestría en Medio Ambiente y Desarrollo Doctorado en Ingeniería - Recursos Hidráulicos Doctorado Interinstitucional en Ciencias del Mar

Mayor información:

E-mail: acma_med@unal.edu.co Teléfono: (57-4) 425 5105