

DYNA

ISSN: 0012-7353 ISSN: 2346-2183

Universidad Nacional de Colombia

Ávila-Lara, Diógenes Daniel; Rubio-Ríos, Anilú; Rosales-Marines, Lucero; Solanilla-Duque, José Fernando; Flores-Gallegos, Adriana Carolina; Rodríguez-Herrera, Raúl Optimization of parameters of a single effect evaporator for agave syrup production DYNA, vol. 88, no. 219, 2021, October-December, pp. 118-122 Universidad Nacional de Colombia

DOI: https://doi.org/10.15446/dyna.v88n219.97085

Available in: https://www.redalyc.org/articulo.oa?id=49671317014



Complete issue

More information about this article

Journal's webpage in redalyc.org



Scientific Information System Redalyc

Network of Scientific Journals from Latin America and the Caribbean, Spain and Portugal

Project academic non-profit, developed under the open access initiative







Optimization of parameters of a single effect evaporator for agave syrup production

Diógenes Daniel Ávila-Lara ^a, Anilú Rubio-Ríos ^{b*}, Lucero Rosales-Marines ^b, José Fernando Solanilla-Duque ^c, Adriana Carolina Flores-Gallegos ^a & Raúl Rodríguez-Herrera ^a

a Departamento de Investigación en Alimentos, Facultad de Ciencias Químicas, Universidad Autónoma de Coahuila, Saltillo Coahuila, México. aviladiogenes@uadec.edu.mx, carolinaflores@uadec.edu.mx, raul.rodriguez@uadec.edu.mx.

Received: July 4th, 2021. Received in revised form: September 1st, 2021. Accepted: September 22th, 2021.

Abstract

The objective of this study was the design and simulation of a single effect evaporator for an agave syrup production line, using as the main raw material, the aguamiel obtained from *Agave salmiana* produced in the ejido "Las Mangas" municipality of Saltillo Coahuila, Mexico. The design of the single effect evaporator was carried out by evaluating the equipment design variables (such as the raw material feed to the evaporator and its outlet streams) by the response surface methodology (RSM), with which it was possible to determine saturation temperature inside the evaporator (F1) and the feed temperature (F2), which allow optimizing the consumption and supply of the flow required for saturated steam entering the evaporator (S, kg/h) and the heat required for the operation of the evaporator (q, kJ/h) in yield to obtain agave syrup.

Keywords: Agave salmiana; simulation; aguamiel; RSM.

Optimización de parámetros de un evaporador de simple efecto para la producción de miel de agave

Resumen

El presente estudio tiene como objetivo el diseño y simulación de un evaporador de simple efecto para una línea de producción de miel de agave, usando como principal materia prima el aguamiel obtenido de Agave salmiana producido en el ejido "Las Mangas", municipio de Saltillo Coahuila, México. El diseño del evaporador de simple efecto se realizó mediante la evaluación de las variables de diseño del equipo (como la alimentación de materia prima al evaporador y sus corrientes de salida) por la metodología de superficie de respuesta RSM, con la cual fue posible determinar la temperatura de saturación en el interior del evaporador (F1) y la temperatura de alimentación (F2), que permiten optimizar el consumo y suministro del flujo requerido para el vapor saturado que entra al evaporador (S, kg/h) y el calor requerido para la operación del evaporador (q, kJ/h) en rendimiento para la obtención de miel de agave.

Palabras claves: Agave salmiana; simulación; aguamiel; RSM.

1. Introduction

In pproduction of fresh beverages, such aguamiel, or concentrated products from these drinks (syrup), the use of evaporator system is the main step of the process, which consists of the use of evaporators for the concentration of foods similar to those used to produce apple jam from fruit concentration [1]. Studies such as standardization of the process to obtain agave syrup [2] have opened the doors for the production of this syrup with good commercial and industrial potential.

The hypothesis, through a predictive model using the RSM methodology, is possible to establish the optimal conditions of the process, and that use of an evaporator of simple effect presents advantages for increasing the production of agave

How to cite: Ávila-Lara, D.D., Rubio-Ríos, A., Rosales-Marines, L., Solanilla-Duque, J.F., Flores-Gallegos, A.C. and Rodríguez-Herrera, R., Optimization of parameters of a single effect evaporator for agave syrup production.. DYNA, 88(219), pp. 118-122, October - December, 2021.

b Cuerpo Académico de Ingeniería y Simulación de Procesos Químicos, Facultad de Ciencias Químicas, Universidad Autónoma de Coahuila, Saltillo, Coahuila, México. a.rubio@uadec.edu.mx, lucero rosales@uadec.edu.mx

^c Departamento de Ingeniería Agroindustrial, Facultad de Ciencias Agrarias, Universidad de Cauca, Popayán, Colombia. jsolanilla@unicauca.edu.co

syrup and will allow us to reach the required concentration of solids. An evaporator is a piece of equipment in which, through the action of heat, there are phase changes and phenomena associated with the concentration of a solution, such as the formation of solids and decomposition of certain substances [3].

A strategy for the design of the simple-effect evaporator is the use of response surface methodology (RSM), in which the design variables are evaluated to determine the optimal operating values. RSM is widely used in chemical engineering, biology, and other fields of science and technology [4,5] and employs the design of experiments and the development of predictive models for simulation and optimization of processes by characterizing system responses. Based on this background, the objective of the present study was to determine, by a predictive model using RSM methodology, the optimal operating conditions of a simple effect evaporator with long vertical tubes to produce agave syrup.

2. Materials and methods

2.1 Simple effect evaporator

The simple effect evaporator for the agave syrup production process used in this study is shown in Fig. 1, where F is the flow feed of fresh aguamiel, T_F is the temperature at which fresh aguamiel is feed to the simple effect evaporator, W is the solid composition of aguamiel, H₂O is the moisture composition, V is the mass flow of water vapor leaving the simple effect evaporator, L is the mass flow of the concentrate, S is the mass flow of the saturated water vapor fed to the evaporator, T_L is the temperature at which the evaporator concentrate comes out, T_S is the temperature of saturated water vapor that is fed to the evaporator, H_S is the enthalpy of saturated water vapor that enters the evaporator of simple effect to the temperature of T_S. H_{SC} is the enthalpy of the steam that comes out condensed at the temperature of T_S. The evaporator shall operate at a pressure of 101,325 kPa on the outside and 75 kPa on the space of the evaporator, after being considered a filtration step to remove impurities from fresh aguamiel, as well as traces of maguey leaves (pencas) or insects.

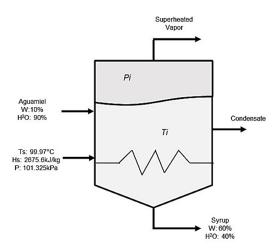


Figure 1 Simple effect evaporator designed for the artisanal production of agave syrup.

Source: Prepared by the authors.

Saturation and feeding temperatures for agave syrup production.

Factor	Level			
ractor	low	intermediate	high	
Saturation temperature inside	88.24	90	91.76	
the evaporator, F1, (°C)	00.21	70	71.70	
Feed temperature, F2, (°C)	20	25	30	

Source: Prepared by the authors.

2.2 Response Surface Methodology (RSM)

The optimization of evaporation process parameters was carried out using a prediction control model, which allows predicting the effect of the action of the control variables on the response of the process. RSM was used for establishing a series of experiments for predictions of a response y, fitting the model to determine the optimal conditions in the input variables. The factors that were considered in the RSM analysis are saturation temperature inside the evaporator (F1) and feed temperature (F2). Parametric optimization is done through RSM and Excel solver (Table 1).

The responses to be optimized were the flow required for the saturated steam entering the evaporator, S, (kg/h), and the heat required for the operation of the evaporator, q, (kJ/h). These data are analyzed by means of RSM and using a polynomial model associated with the experimental design:

$$Y = B_0 + \sum_{i=1}^{n} B_i x_i + \sum_{i=1}^{n} B_{ii} x_i^2 + \sum_{i=1}^{n-1} \sum_{i=1}^{n} B_{ij} x_i x_j$$

(Ec. predictive model)

The Excel Solver was used as a calculation tool to obtain the required optimal values and variables of the predictive model B0, B1, B2, B11, B22, B12, SSR, R² (Table 2).

2.3 Mass balance for production of agave syrup

The chemical composition and physicochemical characteristics of the fresh aguamiel were taken from the mean values reported by Castro et al [6] considering 10% solids (W) and 90% humidity ($\rm H_2O$) content, and for agave syrup, 60% of total soluble solids, data generated from aguamiel, and agave syrup obtained in the ejido "Las Mangas" Municipality of Saltillo Coahuila Mexico.

The general equation for the realization of global mass balances was:

Table 2. Performed treatments, factors, and codification.

Treat	Factors		Codification		An	swers
ment	Tsat (°C)	Tf(°C)	F1 (°C)	F2 (°C)	S (kg/h)	q(kJ/h)
1		20	-1.000	-1	24.08	54326.86
2	88.24	25	-1.000	0	23.86	53840.78
3		30	-1.000	1	23.64	53354.69
4		20	0.000	-1	24.11	54405.16
5	90	25	0.000	0	23.89	53919.08
6		30	0.000	1	23.68	53432.99
7		20	1.000	-1	24.15	54487.21
8	91.76	25	1.000	0	23.93	54001.12
9		30	1.000	1	23.72	53515.04

Source: Prepared by the authors.

2.3.1 Global Mass Balance

$$F = V + L \tag{1}$$

Where F is fresh aguamiel feed flow (25 kg/h), V is the mass flow of saturated steam leaving the evaporator for a simple effect, and L is the mass flow of the concentrate.

2.3.2 Balance sheets by component

$$W: F \cdot (0.1) = L \cdot (0.6)$$
 (2)

$$H20: F \cdot (0.9) = L \cdot (0.4) + V \tag{3}$$

Knowing the percentage of solids in the composition of fresh aguamiel (10%) and the final percentage of solids in agave syrup (60%), the mass flow value of agave syrup (L) was determined according to Eqs. 1, 2, and 3, to determine the massive flow of the input and output currents. The value of the mass flow of saturated steam (V) was calculated later, since in that the current has a percentage of 100% humidity that is to say that it only has water vapor.

Energy balance for production of agave syrup

The energy balance was used to determine the heat that is required within the simple effect evaporator, and the value of the mass flow of saturated water vapor that is needed, as well as the value of condensed water vapor this is due to the enthalpy of the saturated water vapor entering the equipment and enthalpy of the condensed steam coming out, according to input and output definitions in Fig. 1.

For calculation of enthalpy, H_F and H_L, enthalpy of fresh aguamiel feed and concentrate, respectively, a temperature of 0 °C was used for water, and temperature of fresh aguamiel was 20 °C. The equations for calculation of aguamiel enthalpy at the entrance, and concentrate at the process end were as follows:

Where: T_{ref} is the temperature at 0°C with respect to water, Cp_F is the heat capacity of the feed and Cp_L the heat capacity of the concentrate.

$$H_F = Cp_F \cdot (T_F - T_{ref}) \tag{4}$$

$$H_F = Cp_F \cdot (T_F - T_{ref})$$

$$H_L = Cp_L \cdot (T_I - T_{ref})$$
(5)

The saturated water vapor inputs required by the evaporator (S) were calculated in the energy balance, as well as the condensed steam output (S_{C).} It was required to calculate enthalpy values of the water vapor at the operating pressure and enthalpy of fresh aguamiel feed and the concentrate, HF, and H_L, respectively, using the equation of the heat capacity that is a function of the percentage of humidity (Eqs. 4 and 5).

Table 3 shows properties of water vapor at a pressure of 101,325 kPa on the outside of the evaporator and 75 kPa on the space of the evaporator [7], these data allow to determine the enthalpy of saturated water vapor at the entrance to the equipment (H_s), enthalpy of saturated steam leaving the equipment (H_V) and enthalpy of the condensed steam coming out (H_{SC}).

Table 3. Properties of water vapor in the evaporator

 System	P (kPa)	T _I (°C)	H _S (kJ/kg)	H _{SC} (kJ/kg)	H _V (kJ/kg)
Exterior of the evaporator	101.325	99.97	2675.6	419.06	
Evaporator spacing	75	91.76			2662.4

Source: Prepared by the authors.

Eqs. 6, 7, and 8 were applied for the calculation of the saturated water vapor required by the evaporator and the transfer heat in the equipment.

$$F \cdot H_F + S \cdot H_S = V \cdot H_V + L \cdot H_L + S_C \cdot H_{SC} \tag{6}$$

$$S = S_C \tag{7}$$

$$F \cdot H_F + S \cdot \lambda = V \cdot H_V + L \cdot H_L \tag{8}$$

Isolating the water vapor required by the evaporator, S, the following equation is obtained.

$$S = \frac{(V \cdot H_V + L \cdot H_L - F \cdot H_F)}{(H_S - H_{SC})} \tag{9}$$

$$q = A \cdot U \cdot (\Delta T) = S \cdot (H_S - H_{SC}) = S \cdot \lambda \tag{10}$$

Where A is the heat transfer area of the equipment, U is the heat transfer coefficient and λ corresponds to the latent heat condensation of the water vapor.

3. Results

Response Surface Methodology RSM

The analysis by RSM allowed for the observation of the interaction of the evaluated parameters, where all the factors were varied simultaneously over a set of experimental executions to establish the optimal operating conditions of the simple effect evaporator. The responses to optimize, either maximize or minimize the value, was the flow required for the saturated steam entering the evaporator (S, kg/h) and the heat required for the operation of the evaporator (q, kJ/h). It was determined the predictive model to obtain the optimal values for the model of saturated steam entering the evaporator (S, kg/h) shown in Table 4, Where B0, B1, B2, B11, B22, and B12 are the model variables according to their linear or quadratic contribution, respectively SSR is the sum of squares due to regression.

Table 4. Predictive model parameters for saturated water vapor required by the evaporator

I	Parameters
B0	23.8945
B1	0.0355
B2	-0.2154
B11	0.0009
B22	0.0001
B12	0.0002
SSR	0.000
R^2	1.00

Source: Prepared by the authors.

Table 5
Predictive model for the flow required for saturated water vapor entering the evaporator.

F1, (°C)	F2, (°C)	R1, (S, kg/h)	$S(R_{model})$	SQ- residual	residual
-1.000	-1	24.08	24.08	0.000	0.00
-1.000	0	23.86	23.86	0.000	0.00
-1.000	1	23.64	23.64	0.000	0.00
0	-1	24.11	24.11	0.000	0.00
0	0	23.89	23.89	0.000	0.00
0	1	23.68	23.68	0.000	0.00
1	-1	24.15	24.15	0.000	0.00
1	0	23.93	23.93	0.000	0.00
1	1	23.72	23.72	0.000	0.00

Source: Prepared by the authors.

Table 5 shows the behavior of the predictive model using, $R1 = B0 + B1 * F1 + B2 * F2 + B11 * F1^2 + B22 * F2^2 + B12 * F1 * F2$ such as the model equation for saturated steam entering the evaporator (S, kg/h) used for nonlinear regression of R1. F1 is temperature of the saturated water vapor inside the evaporator, F2 is feed temperature of the raw material, R1 is mass flow of saturated water vapor required by the evaporator and $S(R_{model})$ is the R1 calculated using the predictive model with which the adjustment of the data is 1, so that the parameters are reliable.

Table 6 shows the optimal values for maximum and minimum operating conditions in the saturated water vapor response entering the evaporator according to the selected factors. F1 is the temperature of the saturated water vapor inside the evaporator, F2 is the feed temperature of the raw material, R1 is the mass flow of saturated water vapor required by the evaporator.

These results allow us to establish what conditions must be considered to reducing the consumption of saturated water vapor required by the evaporator.

Solving the equation with RSM methodology yields the parameters for the heat equation required for the operation of the evaporator (q, kJ/h) (Table 7). Where B0, B1, B2, B11, B22, and B12 are the model variables according to their linear or quadratic contribution, respectively, SSR is the sum of squares due to regression.

Table 6.

Maximum and minimum operating conditions for saturated water vapour entering the evaporator.

F1 (°C)	F2 (°C)	R1 S(R _{model}) (kg/h)	
1	-1	24.15	
91.76	20	Maximum conditions	
-1	1	23.64	
88.24	30	Minimum conditions	

Source: Prepared by the authors.

Table 7. Predictive model parameters for the heat required for evaporator operation.

Parameters			
В0	53918.8622		
B1	80.1737		
B2	-486.0879		
B11	2.0447		
B22	0.0779		
B12	-0.0013		
SSR	0.090		
R^2	1.00		

Source: Prepared by the authors.

Table 8.

Predictive model for the flow required for saturated water vapor entering the evaporator.

F1 (°C)	F2 (°C)	R2(q, kJ/h)	$q(R_{\text{model}})$	SQ- residual	residual
-1.000	-1	54326.86	54326.90	0.001	-0.03
-1.000	0	53840.78	53840.73	0.002	0.04
-1.000	1	53354.69	53354.72	0.001	-0.04
0	-1	54405.16	54405.03	0.018	0.13
0	0	53919.08	53918.86	0.045	0.21
0	1	53432.99	53432.85	0.018	0.14
1	-1	54487.21	54487.25	0.001	-0.04
1	0	54001.12	54001.08	0.002	0.04
1	1	53515.04	53515.07	0.001	-0.03

Source: Prepared by the authors.

Table 8 shows the behavior of the predictive model using, $R2 = B0 + B1 * F1 + B2 * F2 + B11 * F1^2 + B22 * F2^2 + B12 * F1 * F2$ as a model equation for the heat required for evaporator operation (q, kJ/h) used for nonlinear regression of R2. Where: F1 is temperature of the saturated water vapor inside the evaporator, F2 is feed temperature of the raw material, R2 is transfer heat in the evaporator and q(Rmodel) is R2 calculated using the predictive model with which we can see that the adjustment of the data is 1, so that the parameters are reliable.

Table 9 presents the optimal values for maximum and minimum operating conditions in the heat response required for the operation of the evaporator according to the selected factors. According to this, it is possible to estimate the conditions that should be considered decreasing heat supply to the evaporator. Where: F1 is temperature of saturated water vapor inside the evaporator, F2 is feed temperature of the raw material, R2 is heat required by the evaporator.

3.2 Mass and energy balances

At this step, the maximum operating conditions for R1 and R2, presented in Tables 6 and 9, respectively, were used. Global mass balance was calculated by Eq. 1 where F = 25 kg/h is the fresh aguamiel feed.

From the individual balances, obtained with the equations 2 and 3, it was possible to determine values of concentrate flow, $L=4.17\ kg/h$ and water vapor, $V=20.83\ kg/h$, having a ratio of 1:6 with respect to the feed and the concentrated product.

The equations used for the calculation of aguamiel enthalpies at the beginning and concentrate at the end of the process equations 4 and 5 were developed using the equation of $Cp_i=0.0273^*(\%H)+1.4317$ to obtain $\,Cp_F=3.89$ kJ/kg °C and $\,H_F=77.77$ kJ/kg at the input and, $Cp_L=2.52$ kJ/kg °C and $\,H_L=231.57$ kJ/kg at the output, respectively, according to studies of the heat capacity (Cp) (kJ kg^-l °C^-l) for Colombian honeys from Boyacá, defining the function $\,Cp=0.0273^*(\%H)+1.4317\,$ for a humidity range between 20-45%, for honey crystals [7].

Table 9. Maximum and minimum operating conditions for the heat required by the evaporator.

F1 (°C)	F2 (°C)	R2 q(R _{model}) (kJ/h)
1	-1	54487.49
91.76	20	Maximum value
-1	1	53351.54
88.24	30	Minimum value

Source: Prepared by the authors.

3.3 Energy balance (Heat):

The equations necessary for the calculation of the water vapor required by the evaporator and transferred heat in the equipment were equations 6, 7, 8, 9 and 10. Considering S as the massive flow of water vapor required by the evaporator, S = 24.15 kg/h, and the transfer heat required by the process was q = 54487.21 kJ/h (15135.34 Watts).

4. Discussion

According to predictive models, the model equation used for the nonlinear regression of R1 and R2 is adequate with the parameters obtained, therefore using the RSM methodology of response surface the optimal values of the conditions for F1 and F2 with which the maximum and minimum results can be achieved were obtained. The inlet and outlet flow of the evaporator of simple effect at conditions of a pressure of 75 kPa inside and 101,325 kPa outside, with a temperature of 20 $^{\circ}$ C in the feeding and a heat of 54487.21 kJ/h, are $F=25\ kg/h,\ S=24.15\ kg/h$ and $V=20.83\ kg/h,\ L=4.17\ kg/h$, respectively.

The evaporation process is affected by different parameters such as saturation temperature at which the saturated vapor required by the evaporator is fed, the temperature inside the evaporator and the feed temperature.

In Tables 5 and 8, it is observed that to achieve a lower consumption of saturated water vapor and heat supplied to the evaporator, according to the results obtained, indicate that operating conditions to reduce consumption of saturated water vapor and heat supplied to the evaporator are 88.24 °C inside the evaporator and 30 °C the aguamiel, achieving an agave syrup production of 4.17 kg/h by feeding 25 kg/h of aguamiel. A evaluation of thee results with those reported by González Montemayor et al, (2020) [2], it can be seen the impact of temperature during the process of making agave honey also affects processing time, that using the maximum operating conditions would reduce the processing time but a greater consumption of saturated water vapor, supplied heat would be obtained.

5. Conclusion

The present study shows how the temperature inside the evaporator and the temperature at which the aguamiel is fed can affect the consumption of saturated water vapor and heat supplied to the evaporator. The response surface methodology (RSM) is one viable method for developing, improving, and optimizing the response variable. The methodology used allowed to identify the operational considerations to reduce consumption of saturated water vapor and heat supplied to the evaporator.

The mass and energy balances allow establishing that the theoretical expectations of founding, through a predictive model, the optimal operating conditions of a single-effect evaporator were met, since the predictive model proved to be an efficient tool in the systematic control of the operating parameters.

References

 Solis.Guzman, J., Fruit juice production line. Memory end-of-career project. Higher School of Industrial Engineers. University of Seville, Spain, 2005.

- [2] González-Montemayor, Á.M., Flores-Gallegos, A.C., Serrato-Villegas, L.E., Ruelas-Chacón, X., López, M.G. and Rodríguez-Herrera, R., Processing temperature effect on the chemical content of concentrated aguamiel syrups obtained from two different *Agave species*. Food Measure and Characterization, 14, pp. 1733-1743, 2020. DOI: 10.1007/s11694-020-00421-4.
- [3] Sánchez-Albores, R., Design of an evaporation system to concentrate pressing liquid from the pressing of the fishmeal process. Professional Residential Memory Degree. Technological Institute of Tuxtla Gutiérrez. Mexico, 2010.
- [4] Van Hoof, V., Van den Abeele, L., Buekenhoudt, A., Dotremont, C. and Lysen, R., Economic comparison between azeotropic distillation and different hybrid systems distillation combining with pervaporation for the dehydration of isopropanol Separation. and Purification Technology, 37, pp. 33-49, 2004.
- [5] Ravikumar, K., Pakshirajan, K., Swaminathan, T. and Balu, K., Optimization of batch process parameters using response surface methodology for dye removal by a novel adsorbent, Chemical Engineering Journal, 105, pp. 131-138, 2005.
- [6] Castro-Díaz, A.S. y Guerrero-Beltrán, J.A., El agave y sus productos. Temas Selectos de Ingeniería de Alimentos, 7(2), pp. 53-61, 2013.
- Cengel, Y.A., Introduction to Thermodynamics and Heat Transfer. McGraw-Hill, 2009.
- [8] Salamanca-Grosso, G., Osorio-Tangarife, M.P. and Reyes-Méndez, L.M., Physicochemical properties of monofloral encenillo honeys from the Altoandina zone in Boyacá, Colombia. Quimica Nova, 40(8), pp. 854-864, 2017. DOI: 10.21577/0100-4042.20170084
- **D.D.** Ávila-Lara, is a BSc. Eng. in Chemical Engineer by Autonomous University of Coahuila, Mexico. ORCID:0000-0002-6004-1010
- A. Rubio-Ríos, is PhD in Materials Science from the Centro de Investigación en Materiales Avanzados S.C. Research-professor in the Faculty of Chemical Sciences of the Autonomous University of Coahuila, Mexico. ORCID: 0000-0003-0222-1424
- L. Rosales-Marines, is PhD in Science in Chemical Engineering from the Instituto Tecnológico de Celaya. Research-professor in the Faculty of Chemical Sciences of the Autonomous University of Coahuila, Mexico. ORCID:0000-0001-7365-9859
- **J.F. Solanilla-Duque**, is PhD in Science and technology in colloids and interfaces from the Universidad Pablo de Olavide, Spain. Research-professor at Universidad del Cauca, Colombia. ORCID:0000-0001-6664-9134
- **A.C.Flores-Gallegos,** is PhD in Food Science and Technology from the Autonomous University of Coahuila, Mexico. Research-professor in the Faculty of Chemical Sciences of the Autonomous University of Coahuila. ORCID:0000-0001-5092-1404
- **R. Rodríguez-Herrera**, is PhD in Plant Breeding from the Texas A&M University, USA. Research-professor in the Faculty of Chemical Sciences of the Autonomous University of Coahuila. ORCID:0000-0002-6428-4925