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#### **MONOGRAPH**

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## BEYOND GLOBAL WARMING

Stressed oceans, globe-wide in the Anthropocene

## CARLES PELEJERO AND EVA CALVO

The footprint of human activities on the planet is so profound that many scientists are already suggesting that we have entered a new geological era, the Anthropocene. From among these activities, those that are accompanied by large emissions of carbon dioxide ( $CO_2$ ) affect our entire planet and, especially, the oceans. Besides becoming warmer, the oceans are also growing progressively more acidic and less oxygenated. In this article we discuss the extent of these global stresses on the oceans after contextualising the disproportionate anthropogenic increase in  $CO_2$  and examining how it is distributed. We conclude with a discussion of mitigation possibilities that use the oceans themselves, stressing the urgent need to tackle the problem, especially during this present decade.

Keywords: Anthropocene, warming, acidification, deoxygenation, global change.

# ■ THE DISPROPORTIONATE INCREASE IN ATMOSPHERIC CO<sub>2</sub>

One of the most graphic and palpable ways to help understand the magnitude of the problem of anthropogenic  $CO_2$  emissions is to compare the atmospheric concentrations of this greenhouse gas over recent decades with those of the past. We have had access to modern instrumental data since Charles David Keeling first began making measurements at an

elevation of 3,400 m at Mauna Loa, Hawaii; now over 60 years ago, he initiated what is now known as the Keeling curve. His first analysis, in March 1958, measured 313 ppmv (parts per million by volume). Since then, the CO<sub>2</sub> concentrations at this summit – which was chosen so that an average atmospheric value would be obtained without

being influenced by nearby industrial areas or large masses of vegetation – have increased continuously and progressively (following small annual cycles) and, in May 2013, they crossed the 400 ppm barrier. The 2019 average was 411 ppm, and in recent years atmospheric  $CO_2$  has been increasing by between 2 to 3 ppm per year.

«The CO<sub>2</sub> absorbed by the oceans affects a number of chemical equilibriums and results in gradual ocean

acidification»

For periods preceding the Keeling curve – earlier than 1958 – we have valuable evidence obtained from the ice accumulated over hundreds of thousands of years in Antarctica (Alley, 2014). As this ice compacts, air is trapped within it in the form of small bubbles, which preserve an extraordinary record of the composition of the atmosphere in the past. In some areas of Antarctica, we can find accumulations of more than three kilometres of ice. Through the study of

these ice cores at the beginning of the 1980s, it was discovered that, in the last ice age, now approximately 20,000 years ago, the concentration of CO<sub>2</sub> in the air had been significantly lower than it is today (Delmas et al., 1980). After almost thirty years studying new ice cores, a continuous record of atmospheric CO<sub>2</sub> was obtained

for the last 800,000 years, including a series of glacial/interglacial cycles (Lüthi et al., 2008). Recently, it became possible to determine  $CO_2$  concentrations through the analysis of Antarctic ice for earlier, periods reaching back up to two million years, although we still do not yet have a continuous record covering such a long period (Yan et al., 2019).

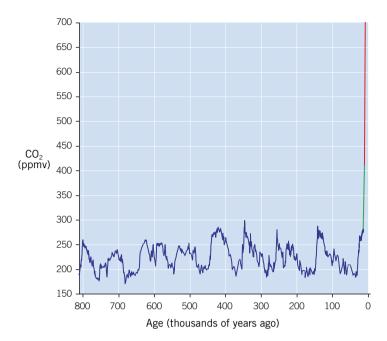


Figure 1.  $CO_2$  concentration in the atmosphere in parts per million by volume (ppmv) for the last 800,000 years and up to the pre-industrial era (data measured in Antarctic ice cores are shown in blue; data from Lüthi et al. [2008]), in the last few centuries and up to the present (shown in green, as measured in ice cores and instruments; data from *The Keeling Curve* project, Scripps Institution of Oceanography, University of San Diego, California), and in future projections up to the end of the twenty-first century, according to intermediate emissions scenarios (shown in red).

SOURCE: Created by the authors from referenced sources

As can be seen in Figure 1, the continuous record of atmospheric  $CO_2$  for the last 800,000 years shows values ranging from 180 ppm during the cold glacial periods to 280 ppm during the warm interglacial periods. In this context, the current values – already above 410 ppm – have clearly fallen outside the range of natural atmospheric  $CO_2$  variability for, at least, the last 800,000 years. Depending on future socioeconomic models, energy policies and population growth, the concentration of  $CO_2$  in the atmosphere will take different trajectories, which could exceed 1,000 ppm by the end of the century according to the most pessimistic scenarios, or reach values below 500 ppm in the most optimistic ones (Fuss et al., 2014).

#### ■ HOW IS THIS CO₂ DISTRIBUTED?

Fundamentally, the CO<sub>2</sub> that humans release into the atmosphere is distributed into three major compartments: atmosphere, continents, and oceans (Figure 2). During the 2009–2018 period, it was

estimated that 44 % of CO<sub>2</sub> emissions went into the atmosphere, 29 % went into the continents (fixed by plants), and 23 % was absorbed by the seas and oceans, with the remaining 4% representing the current mismatch between the calculation of global emissions and sinks (Friedlingstein et al., 2019). These calculations are independent and, because of their inherent uncertainties, emissions and sinks often do not match perfectly. A positive mismatch, as in this case, either means that emissions are being overestimated, or that sink estimates are lower than the actual values. We should be grateful, therefore, that not all the CO<sub>2</sub> that humans release by burning of fossil fuels remains in the atmosphere and, most importantly, that it can be absorbed by plants and ocean waters. Were it not so, the concentration of CO<sub>2</sub> in the atmosphere would be significantly higher, thereby aggravating the greenhouse effect and global warming. However, as we will discuss below, the CO<sub>2</sub> absorbed by the oceans affects a number of chemical balances and results in gradual ocean acidification, which conditions the development of many marine organisms.

## ■ GLOBAL STRESSES RELATED TO THE INCREASE IN CO<sub>2</sub>

Today, oceans and marine ecosystems are suffering from multiple stresses related to human activities, with impacts manifesting themselves both locally and globally. Some examples of these pressures include overfishing, pollution, destructive fishing techniques, eutrophication (increased nutrient availability), habitat destruction, species invasions, and maritime traffic, among others. In addition to these stresses, there are three more closely related to CO<sub>2</sub> emissions and climate change, and these generally manifest themselves globally in all the seas and oceans: warming, acidification, and deoxygenation (Gruber, 2011).

#### Warmer oceans

Without a doubt, the reports by the Intergovernmental Panel on Climate Change (IPCC) are becoming increasingly clear and conclusive – the enormous increase in atmospheric  $CO_2$  is causing global warming on our planet. It is estimated that, since the beginning of industrialisation, the Earth's global surface temperature has already increased by about 1 °C. Regarding the oceans, much of the excess heat experienced by the Earth due to the greenhouse effect (around 90 %) is retained in its waters. Surface water has warmed, on average, by about 0.6–0.8 °C from pre-industrial times to 2010. Depending on future emissions scenarios, sea surface temperatures could



Figure 2. Annual CO₂ emissions (average for the period 2009–2018) derived from the burning of fossil fuels and deforestation, and the three compartments that act as CO₂ sinks: the atmosphere, ground vegetation, and oceans.

SOURCE: Adapted from Global Carbon Project (2019)

rise by an average of 0.7 °C to 2.6 °C more by the end of the twenty-first century, according to the most optimistic and pessimistic scenarios, respectively, compiled in a recent IPCC Special Report on the Ocean and Cryosphere (IPCC, 2019). To a lesser extent - but already measurable instrumentally - deep oceanic waters are also getting warmer, even below 4,000 m, especially in the Southern Ocean.

ecosystems such as tropical coral reefs and Mediterranean «Much of the excess heat coralligenous communities, marine forests such as the experienced by the Earth Posidonia oceanica meadows or due to the greenhouse effect kelp forests, and phytoplankton is retained in oceanic waters» species, among others. For example, the increasingly frequent ocean heat waves cause coral bleaching episodes and mass mortality among sessile species. The 2016 marine heat wave on the Great

Barrier Reef off the north-eastern coast of Australia,

for instance, killed one-third of the shallowest corals

Such warming is having a major impact on

### More acidic oceans

(GBRMPA, 2017).

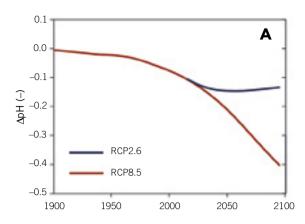
In addition to the global warming of marine waters, another planetary problem also caused by the exorbitant increase in atmospheric CO<sub>2</sub> has been intensively

studied for the last fifteen years or so. Because this greenhouse gas dissolves in seawater, it interferes with a series of chemical reactions – the equilibriums between carbonic acid and bicarbonate and carbonate ions resulting in an increase in proton concentration, i.e., increased water acidity or corrosiveness. It is estimated that the pH of ocean surface water has decreased, on average, by about 0.1 units from pre-industrial times to 2010 (Pelejero et al., 2010). Indeed, over the past few

decades, pH has been declining at speeds of between 0.017 and 0.027 units per decade and, depending on future emissions scenarios, the marine surface pH could fall by a further 0.04 to 0.29 units, on average, by the end of the twenty-first century, according to the most optimistic

or pessimistic scenarios, respectively (IPCC, 2019; see Figure 3).

This change in water chemistry has many implications for marine organisms, especially those that build calcium carbonate structures, such as corals, bivalves, gastropods, crustaceans and coccolithophores. Under higher acidity conditions these organisms often find it more difficult to build their shells, structures, and skeletons, which are also more likely to deteriorate and dissolve. Acidification stress adds to global warming



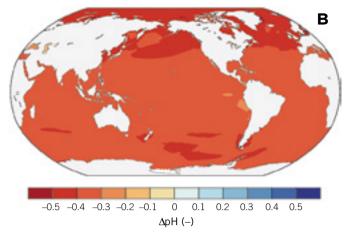


Figure 3. **A**) Simulation of past and future changes in global oceanic surface water pH according to optimistic (RCP2.6) and pessimistic (RCP8.5) scenarios. **B**) Changes in the surface pH of the oceans by the end of the twenty-first century in relation to the 1850–1900 period, according to a pessimistic scenario (RCP8.5). See Fuss et al. (2014) for more information on Representative Concentration Pathways (RCP) emission scenarios.

SOURCE: IPCC (2019, p. 470)

and often also interacts synergistically to further affect marine organisms. In the case of tropical corals, for example, which are already affected very significantly by global warming, when we add up the effect of acidification, which turns their structures more fragile and vulnerable, that means that they may have more difficulties to recover, for instance, after a bleaching event (Figure 4).

## Less oxygenated oceans

More recently – in the last five to ten years – a third global stressor has also become the focus of a lot of attention from the marine scientific community: the progressive water deoxygenation. This phenomenon is also closely related to the

increase in atmospheric CO<sub>2</sub> and climate change. In part, this phenomenon is caused by the warming of the oceans: the warmer the water, the less soluble the gases dissolved in it are. Under warmer conditions, oxygen has a greater tendency to escape from the water into the air. Unlike warming and acidification, which are more intense at the oceanic surface, in the open sea the problem of global deoxygenation is more pronounced in deeper depths, while the more superficial layers are usually highly oxygenated. The process also depends on the ocean basin. The Atlantic Ocean, for example, is better ventilated and is, therefore, better oxygenated than the Indian or Pacific Oceans, with the latter having the lowest levels of oxygen, particularly in its intermediate layers. Oxygen levels in the equatorial Pacific Ocean, at a depth of between 200 and 1,000 m, are particularly low.

«Some studies suggest that over the last five decades the oxygen content of the oceans has decreased by 2 %»

As for marine animals, only species adapted to living in such chemical conditions, like certain species of demersal fish or some cephalopods such as the giant squid can be found living in these oxygen-poor regions. The warming of the waters also results in greater stratification of the water column and less ventilation which, in turn, contributes to the progressive deoxygenation. In addition, warmer conditions boost

the metabolism of marine organisms and, as a result, increases their demand for oxygen for respiration. In some coastal areas, the excessive production of organic matter due to eutrophication processes can also lead to the formation of localised areas with low levels of oxygen.

Some studies suggest that, over the last five decades, the oxygen content of the oceans has decreased by 2% (Schmidtko et al., 2017). Depending on future emissions scenarios, dissolved oxygen concentrations in the ocean water column could fall by a further 1.8% to 3.5% on average by the end of the twenty-first century, according to the most optimistic or pessimistic scenarios, respectively (IPCC, 2019; see Figure 5).

## HOW TO TACKLE THE PROBLEM: INTERNATIONAL AGREEMENTS

At the 2015 United Nations Climate Change Conference in Paris, a major international agreement was reached to reduce CO<sub>2</sub> emissions, which entered into force at the end of 2016. This agreement required developed countries to commit to progressively

reducing emissions with the aim of keeping the increase in average global temperature below 2 °C compared to pre-industrial values, taking into account that the increase now stands at 1 °C. However, the commitments adopted so far by the different countries still leave us far from the objective set out in Paris. Later, a special report (IPCC, 2018) warned that this 2 °C warming, despite being an ambitious objective, would still be insufficient to avoid triggering irreversible changes. Specifically in relation to the oceans, it will be vital to limit warming to 1.5 °C to reduce the risks posed to marine biodiversity, fishing, and marine ecosystems, as well as to their functions and the ecosystem services they provide to humans. Among the effects of climate change on the most at-risk ecosystems, this report highlighted the recent



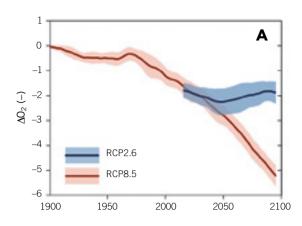
Figure 4. Acidification, caused by the dissolution of carbon dioxide in water, represents an additional stress on corals and other organisms that build calcium carbonate structures, because they are less able to regenerate after a bleaching process such as the one shown in the picture, taken in Australia's Great Barrier Reef.

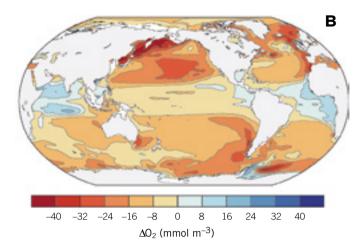
«This change in water chemistry has many implications for marine organisms, especially those that build calcium carbonate structures» alarming decline in Arctic marine ice as well as the changes in the ecosystems associated with tropical coral reefs. The Paris agreement, therefore, is a good international starting point for addressing the root cause of all these global changes (and specially, the disproportionate anthropogenic release of CO<sub>2</sub>) but ideally, it should be much more ambitious in terms of aiming to reduce emissions. In fact, as denounced in scientific reports and articles (see Höhne et al., 2020; United Nations Environment Programme, 2019), not much has been done in the last decade in terms of individual countries' policies to move in this direction.

## POTENTIAL SOLUTIONS IN THE OCEANS THEMSELVES

Some of the solutions to help reduce the emissions of CO<sub>2</sub> and other greenhouse gases can, in fact, be found in the oceans themselves. For example, a recent report suggested a number of ocean-based fields of action that could be used to mitigate emissions (Hoegh-Guldberg, 2019). The oceans offer, for instance, great potential for renewable energy through the use of the energy stored in marine currents, tides, and waves, or with the construction of offshore wind farms. However, it must also be considered that many of these are still in the initial research or pilot phases and, therefore, will require a great deal more research and effort before their implementation. Emissions from maritime transport, which currently represent approximately 2-3 % of all anthropogenic CO<sub>2</sub> emissions, must also be reduced. This can be achieved by increasing energy efficiency, i.e., by reducing the energy required to move ships, or by replacing fossil fuels with other fuels that do not produce carbon emissions (e.g., hydrogen, ammonia, or some biofuels). As for oceanic carbon sequestration, the so-called marine forests, which include coastal ecosystems such as kelp forests, Posidonia

meadows (Figure 6B), wetlands, or marshes, are very effective carbon sinks (collectively known as «blue carbon»), which can capture ten times more carbon per hectare than terrestrial ecosystems such as tropical forests (Mcleod et al., 2011). Thus, protecting these ecosystems, which are highly degraded by anthropogenic activities and global warming, will be key to reducing CO<sub>2</sub> because they play an important role in producing oxygen and protecting the coast from marine storms, cyclones, and tsunamis, and from the rise in sea levels resulting from climate change. Also related to marine organisms, macroalgae





 $Figure \, 5.\, \textbf{A})\, Simulation \, of past \, and \, future \, changes \, in \, the \, average \, dissolved \, oxygen \, concentration \, in \, the \, water \, column \, between \, 100 \, changes \, in \, the \, average \, dissolved \, oxygen \, concentration \, in \, the \, average \, dissolved \, oxygen \, concentration \, in \, the \, average \, dissolved \, oxygen \, concentration \, in \, the \, average \, dissolved \, oxygen \, concentration \, in \, the \, average \, dissolved \, oxygen \, concentration \, in \, the \, average \, dissolved \, oxygen \, concentration \, changes \, dissolved \, oxygen \, concentration \, dissolved \, oxygen \, concentration \, dissolved \, oxygen \, changes \, dissolved \, dissolved$ and 600 m depth, according to optimistic (RCP2.6) and pessimistic scenarios (RCP8.5). B) Spatial changes in the average dissolved oxygen concentration in the water column between 100 and 600 m depth by the end of the twenty-first century in relation to the 1850–1900, period according to a pessimistic scenario (RCP8.5). See Fuss et al. (2014) for more information on RCP emission scenarios.

«We are the last generation

capable of avoiding irreversible

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and the first that is already

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effects»

SOURCE: IPCC (2019, p. 470)

crops have a great potential as replacements for oil-derived compounds or even as a possible food supplement to reduce methane emissions from ruminant livestock (Machado et al., 2016).

## URGENT ACTION MUST BE TAKEN

This brand-new decade will be key in trying to reverse this problem or, at least, to minimise its effects as far as possible. As stated in the latest Emissions Gap Report from the **UN Environment Programme** (2019), we have very little room for manoeuvre; we cannot afford another decade without taking drastic action or it will be impossible to achieve the

maximum warming target of 1.5 °C or even 2.0 °C (Höhne et al., 2020). We require urgent and concerted action by every country and all our industrial sectors. Current policies on energy efficiency must be extended and strengthened, transport and mobility with minimal CO<sub>2</sub> emissions must be promoted, and an ambitious transition towards renewable energies must be promoted, while also bearing in mind that the planet's resources are limited, not only regarding fossil fuels. We are the last generation capable of avoiding irreversible and catastrophic changes, and the first that is already beginning to suffer their effects. In addition, we should not lose sight of the fact that the

problem of climate change and global change can only be solved by a major social transformation. We should all ask ourselves what kind of world we want to live in and leave to future generations. It is very important that we become aware of all these environmental problems, the effects they have at the planetary scale, and the magnitude of our footprint, which is very

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comparable to the geological processes of the past. 

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Figure 6. The oceans can play an important role agents in the mitigation and reduction of carbon dioxide emissions, either as potential generators of renewable energy or through carbon sequestration. A) An offshore wind farm in the United Kingdom. B) Offshore forests, such as the Posidonia oceanica meadows shown in the image, are effective carbon sinks.

«In order to reduce the release of CO<sub>2</sub> and other greenhouse gases by humans, some solutions can be found in the oceans themselves»

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