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Original Research Article

In vitro study of solubilization ability of bovine pulp tissue using different irrigating solutions

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Abstract

Introduction and Objective: The aim of this study was to evaluate in vitro the capacity of solubilization of bovine pulp tissue, promoted by the following auxiliary chemical solutions: 2.5% sodium hypochlorite at pH 9, 5.25% sodium hypochlorite at pH 9, 2% chlorhexidine gel at pH 6, 17% EDTA at pH 7, and Smear-Clear®. Material and methods: A total of ten specimens of bovine pulp tissue, weighing 1.65 g each, were placed into flasks connected to a device developed for the study in order to reproduce irrigation dynamics. The flasks with the specimens received a volume of 80 ml of irrigating solution and the experimental times were 15, 30, 45 and 60 minutes. Results: The results were analyzed statistically by Kruskal-Wallis test to compare the different experiment times (15, 30, 45 and 60 minutes) of each irrigating solution. To compare the variation among the times of one solution, Kolmogorov-Smirnov test (Lilliefors) was used. Conclusion: Considering the results obtained and the limitations of the study, it can be concluded that 5.25% sodium hypochlorite exhibited the greatest solubilization ability, followed by 2.5% sodium hypochlorite, Smear-Clear® , 17% EDTA and 2% chlorhexidine gel.
Introduction

At the ending of the 19th century, the clinicians and researchers dedicated to root canal treatment clearly agreed on the necessity of cleaning, disinfection and shaping of the root canal system aiming to its further obturation. To reach this goal, instruments (hand and rotary files and burs) are employed to obtain the preparation of the root canal system together with auxiliary chemical solutions so that also the cleaning of variations within the root canal internal morphology is achieved [3].

Thus, biomechanical preparation comprises the association of the mechanical with the chemical stage. The mechanical action prepares, cleans, and shapes the root canal, while the chemical action acts on the components within the root canals in order to dissolve either the live or necrosed organic tissues, remove the debris, and eliminate the microorganisms. Currently, sodium hypochlorite at several concentrations has been the solution of choice for the treatment of root canal system; however, other solutions have been studied, always aiming to find that comprising all desirable requirements for an irrigant.

One of the main properties of the irrigant solutions is the ability of tissue dissolution. The aim of this study is to analyze the solubilization ability of different irrigants during the root canal treatment, especially Smear-Clear, an EDTA-T-based solution plus cetrimide, whose potentiality is little known.

Material and methods

To obtain the bovine pulp tissue, the bovine mandibles were collected immediately after slaughter. The teeth were extracted with the aid of a n. 17 forceps, placed into plastic bags containing saline solution (to maintain the physiologic features of the pulp), kept into Styrofoam box with a thin layer of ice (to maintain the teeth cooled) and sent to the laboratory. Then, the teeth were cut at their long axes and the root and crown portion were separated to expose the coronal pulp. With the aid of a size 10 K file, the pulp tissue was detached and removed from the dentinal walls. Each fragment was washed with saline solution. Thus, the study sample was obtained by placing the fragments onto a glass plate and with the aid of a millimetric ruler and scalpel, each fragment was standardized at 10 mm weighing 1.65 g each, totaling 8.27 g for all experimental groups.

The specimens were divided according to the following groups:
- Group I – 2.5% sodium hypochlorite at pH 9 (Fórmula e Ação Farmácia de Manipulação, São Paulo, SP), batch 141;
- Group II – 5.25% sodium hypochlorite at pH 9 (Fórmula e Ação Farmácia de Manipulação, São Paulo, SP), batch 122;
- Group III – 2% chlorhexidine gel at pH 6 (Fórmula e Ação Farmácia de Manipulação, São Paulo, SP), batch f110710011;
- Group IV – 17% EDTA at pH 7 (Fórmula e Ação Farmácia de Manipulação, São Paulo, SP), batch f060710011;
- Group V – Smear-Clear (Sybron-endo, Optimum, São Paulo, SP.), batch 2731300; EXPIRATION DATE: 2010-02;
- Group VI – distilled water, control group.

The solutions were kept in a dry, fresh place, under environment temperature to maintain their physical-chemical characteristics. To conduct the bovine pulp dissolution test, a 10-100 ml collector was adapted by previously perforating both sides. At one side, a urethane tube was adapted; at the other side, another tube was connected to link the collector to the circulation pump (SB 160; aquarium pump) (figure 1). The pump was placed into a plastic flask inside water. The bovine pulp fragments previously weighed with the aid of a precision scale were placed into the collectors with the aid of a net so that they became total immersed in the irrigants.

Figure 1 – Device constructed to perform the tests
A volume of 80 ml (the ideal amount to make the system work correctly) of the auxiliary chemical solution of each experimental group was used with a constant flow of 64 ml/min, and the following times: 15, 30, 45 and 60 minutes.

Elapsed the experimental times, the tissue remnants (when present) were washed in distilled water for 1 minute, dried with gauze, and weighed in the precision scale (Coleman®). The results were expressed as the percentage of the initial weigh of the specimens.

Results

The samples exhibited an abnormal and homogenous behavior. Kruskal-Wallis test was applied to compare the different experimental times (15, 30, 45 and 60 minutes) of each irrigant solution, and, the different irrigant solutions at each time period. Kolmogorov-Smirlov test (Lilliefors) was applied to compare the variations among the time periods of each solution (graph 1).

Discussion

The collection of the bovine teeth to remove the pulp immediately after the slaughter aimed to achieve specimens with good physiologic features (texture, color, and hydration). The fast and careful pulp tissue preparation was a primordial step of the study, avoiding damages such as the excessive heating of the tissue.

The sodium hypochlorite solutions at 2.5% and 5.25% concentrations, 17% EDTA, and 2% chlorhexidine gel were obtained from the same manufacturer (manipulation pharmacy Fórmula & Ação), aiming to standardization. The solutions were kept at environmental temperature, protected from light, to maintain their physical-chemical characteristics. The washing of the specimens between the periods tested for all solutions was carried out to avoid an additional dissolution – even minimum – during the turning off of the system and the transportation to the sink, especially for 2.5% and 5.25%, hypochlorite solutions. Thus, the scale and the device were strategically placed by the side of the sink to assure a fast weighing of the specimens. For each group, the procedures were repeated eight times to allow the statistical analysis.
For each evaluation period, the auxiliary chemical substance was changed to avoid contamination during the measurements by inserting 80 ml of saline solution. Then, the system was started and let to work for the aforementioned purpose.

The use of chemical substances during the root canal instrumentation is of fundamental importance. The cut dentine and the pulp remnants should be encompassed by the chemical solution to prevent the smear layer formation and its deposition onto the apical portion of the canal, therefore obstructing it. Additionally, these substances should exhibit the following characteristics: make the use of the instruments easy; remove the organic remnants, contaminated or not; act against the possible existing microorganisms.

Considered the results obtained, the sodium hypochlorite was the solution with the greatest solubilization ability of the bovine pulp tissues, at a shorter time period compared with the other solutions tested. These results corroborate the findings observed by Walker [32], Grossman and Meiman [11], Hand et al. [12], Rosenfeld et al. [28], Thé et al. [31], Koskinen et al. [16], Abou-Rass and Oglesby [1], Gordon et al. [10], Moorer and Wesselink [21], Nakamura et al. [24], Hasselgren et al. [13], Morgan et al. [22], Andersen et al. [2] and Johnson and Remeikis [14]. The tissue dissolution action promoted by sodium hydroxide, one of the byproducts of sodium hypochlorite is a powerful organic and fat solvent [21]. Another byproduct – the hypochlorous acid – is a powerful antimicrobial agent because it releases chlorine which combines with the amino group from proteins, forming the chloramines. The hypochlorous acid (HOCl) undergoes decomposition by light, air and heat action releasing free chlorine and secondarily oxygen; also depending on other factors such as: concentration, temperature, hydrogen potential, volume, contact time, contact surface and pulp state [1, 10, 12, 21, 23, 30].

Siqueira et al. [29] reported that the 2.0% chlorhexidine gel and 0.5% hypochlorite solutions at pH 7, ranging from 27ºC to 37ºC were not effective in dissolving organic tissues. It is important to emphasize the greater solubilization ability of Smear-Clear than that of 17% EDTA, despite the fact that they have similar formulations (same active principle). Studies previously conducted indicating the lower ability of this solution in dissolving pulp tissue [4-6, 9, 15, 25, 26, 34], primarily because of the pH of EDTA, around 7 [17]. In this present study, 17% EDTA exhibited small ability of tissue dissolution.

According to the manufacturer, Smear-Clear® is an adjunct solution for the final endodontic irrigation to promote the refining in the ending of the chemical-mechanical preparation, mainly as an auxiliary solution for smear layer solution, as verified by the study of Lui et al. [19]. Although its main function is to remove the smear layer, the results obtained by this present study suggested the presence of an important property: the ability of dissolving pulp tissue. As basic formulation, Smear-Clear (Sybron-endo) contains 17% EDTA plus a detergent agent (to improve the cleaning properties) and cetrimide (an antiseptic agent with bactericidal function). Chlorhexidine seems to exhibit some advantages in relation to sodium hypochlorite, such as substantivity [8, 18, 33] and its antimicrobial potential [8, 7, 18, 27]. One of its greatest disadvantages is the small ability to dissolve organic tissue, as observed by this present study corroborating the findings of D'Arcangelo et al. [7] and Marley et al. [20].

Conclusion

Within the limits of this study and based on the results obtained, it can be concluded that 5.25% sodium hypochlorite exhibited the greatest solubilization ability of bovine pulp tissue, followed by 2.5% sodium hypochlorite, Smear-Clear®, 17% EDTA, and 2% chlorhexidine gel.

References


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