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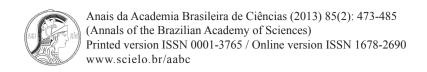
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Isolation and HPLC quantitation of kaurane-type diterpenes and cinnamic acid derivatives of long-term stored leaves of *Mikania laevigata* and *Mikania glomerata*

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ABSTRACT

The leaves of *Mikania laevigata* and *Mikania glomerata* are used in Brazil to treat respiratory affections, being kaurane-type diterpenes and coumarin considered as the bioactive compounds. The present study reports an investigation on the HPLC-DAD profiles and contents of coumarin (1), *trans-o*-coumaric (2), kaurenoic (3), benzoylgrandifloric (4) and cinnamoylgrandifloric (5) acids in dried leaves of *Mikania* species stored in dark room under controlled conditions. Excepting 2, the constituents were isolated and purified to be employed as reference compounds. The samples were analyzed at three monthly intervals up to 18 months for *M. laevigata* and 12 months for *M. glomerata. trans-o*-Coumaric was not detected in both, whereas 1 occurred only in *M. laevigata*. The concentrations of the assayed constituents did not vary significantly within the evaluated period (p < 0.05), for both species. In contrast, changes in the chromatographic profiles and spectral purity of peaks from 3, 4 and 5 were detected in samples of both *Mikania* stored for three months, while the coumarin profile in *M. laevigata* modified after six months of storage. The evaluation of chromatographic profiles based on spectral purity analyses of selected peaks was shown to be a more robust tool to access chemical stability of *Mikania* samples than the quantitation of chemical markers' contents.

Key words: chemical markers changes, HPLC-DAD analysis, *Mikania laevigata*, *Mikania glomerata*, storage, post-harvest.

INTRODUCTION

The validation of medicinal plants has only gained attention of the Brazilian scientific community and government in the last decade (Bertolucci et al. 2009); for that reason, the quality control of herbal drugs in the country is limited by the reduced number of chemical markers

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defined for native species, along with the lack of commercial sources of reference compounds (Braga et al. 2003).

Different chromatographic techniques are employed for the quality control of vegetal drugs and herbal products including qualitative and quantitative methods based on TLC, HPLC and GC analyses (Liang et al. 2009, Razmovski-Naumovski et al. 2010). Chromatographic methods

are extensively used for the quality control of plant raw materials due to their ability to detect variations in chemical composition originated from intraspecific differences, growing conditions, harvest time, processing methods and storage period (Liang et al. 2009). The development of HPLC methods for plant analysis is not straightforward and involves exhaustive optimization of operating conditions (e.g., mobile phase composition, pH, column and temperature) in order to obtain desirable outcomes such as higher plate number, shorter analysis times, peak purity and improved peak resolution (Dharmadi and Gonzalez 2005) - this last in the case of simultaneous analysis of multiple compounds.

Mikania laevigata Schultz Bip. ex Baker and Mikania glomerata Sprengel (Asteraceae) are medicinal species popularly known as guaco, widely used in Brazil to treat respiratory affections (Napimoga and Yatsuda 2010). Kaurane-type diterpenes from the ent-series [kaurenoic (KA), benzoylgrandifloric (BA) and cinnamoylgrandifloric (CA) acids] and the derivatives of cinnamic acid, coumarin (CO) and trans-o-coumaric acid (OC) (Fig. 1) have been identified as constituents of the species (Veneziani and Oliveira 1999, Vilegas et al. 1997, Oliveira et al. 1984) and may account for their alleged biological properties (Ambrosio et al. 2006, Moura et al. 2002). Previous reports have suggested similar chemical composition for the leaves of both Mikania (Veneziani and Oliveira 1999, Vilegas et al. 1997, Oliveira et al. 1984). However, we reported a marked difference in the constitution of M. laevigata and M. glomerata leaves, with the presence of KA, BA and CA in both and the lack of CO and OC in the last one (Bertolucci et al. 2009). Besides, we found higher contents of BA and CA in M. laevigata, while KA was disclosed as the most abundant compound in M. glomerata (Bertolucci et al. 2009).

Safety and efficacy of phytopharmaceutical products are directly affected by the chemical reliability at all stages of the manufacturing processes, including stability and shelf-life of vegetal drugs (Sahoo et al. 2010). Nevertheless, there are only a few reports addressing post-harvest effects, storage and shelf-life of medicinal plants (Guimarães et al. 2011, Madan et al. 2008, Stafford et al. 2005, Fennell et al. 2004, Griggs et al. 2001). As far as we know, storage-related changes in the constituents of *Mikania* species have never been investigated. Taking into account that cinnamic acid derivatives and kaurane-type diterpenes are considered the bioactive compounds of guaco, the goal of the present study was to investigate the HPLC profiles and contents of CO, OC, KA, BA and CA in stored leaves, thus requiring the isolation of some of them to be employed as reference compounds.

MATERIALS AND METHODS

GENERAL EXPERIMENTAL PROCEDURES

Fractionation of the extracts and purification of the isolated compounds were carried out by column chromatography on silica gel 60G (Merck 0.2-0.5 mm and Merck 0.063-0.200 mm) and by preparative TLC on self-coated plates with silica gel 60G (Merck, 70-230 µm). Liebermann-Burchard, *p*-anisaldehyde and NP/PEG solutions were employed as spray reagents for monitoring the fractionation, along with ammonium vapor exposure, the detection was carried out under UV light at 254 and 366 nm (Wagner et al. 1984).

A Shimadzu preparative HPLC system composed of LC-8A quaternary pump, SCL-8A controller, SPD-6AV UV-VIS detector and CR4A integrator was employed for the final purification of the compounds. Analyses were performed on Shimpack Prep Sil (250 × 10 mm d.i., Shimadzu) and Shim-pack Prep-ODS (250 × 10 mm d.i.) columns. Purity of the isolated compounds was checked by melting point data (without correction) determined on MQAPF-301 apparatus (Microquímica), HPLC-DAD and NMR analysis, whereas their identification was accomplished by spectroscopic analysis (UV,

IR, ¹H and ¹³C NMR). IR spectra were recorded on a Perkin-Elmer FT-IR spectrophotometer in ATR mode with internal reference (range 4.000-600 cm⁻¹). ¹H and ¹³C NMR spectra were obtained on Bruker Avance DRX-200 and DRX-400 equipments (Departamento de Ouímica, UFMG), operating respectively at 200 and 400 MHz for ¹H and at 50 and 100 MHz for ¹³C. TMS was employed as internal standard for both nuclei and CDCl₃ solutions were used in the analyses. Optical rotations were determined using a Bellingham Stanley ADP220 automatic recording spectropolarimeter. The analyses of chemical markers were carried out on a Waters 2695 HPLC system composed of a quaternary pump model L-6200A, autosampler, in-line degasser AF (Waters) and photo-diode array detector (Waters 2996). Waters Empower software was employed for data processing. The analyses were performed on a LiChrospher 100 RP-18 column (125 \times 4 mm i.d., 5 μ m; Merck), in combination with a LiChrospher 100 RP-18 guard column (4 × 4 mm i.d., 5µm; Merck).

ISOLATION AND SPECTROSCOPIC CHARACTERIZATION OF CHEMICAL MARKERS

Mikania glomerata and Mikania laevigata leaves were collected from specimens cultivated at the Departamento de Agricultura, Universidade Federal de Lavras, Minas Gerais, Brazil. The leaves were collected in summer (January, 2004), from the apical, intermediate and basal regions of 16-month old plants, cultivated under solar radiation.

The species were identified by Dr. Mara Rejane Ritter from the Instituto de Biociências, Universidade Federal do Rio Grande do Sul, Porto Alegre, Brazil, where vouchers are deposited under the codes ICN 141992 and ICN 141990, respectively. The leaves were dried in a ventilated oven at 40°C for 72 h. The dried materials (2 kg of *Mikania glomerata* and 4 kg of *Mikania laevigata*) were ground in a knife mill, following exhaustive

percolation with ethanol 96% at room temperature. Solvent was removed in a rotatory evaporator and the obtained extracts were kept in desiccators until the complete elimination of the residual solvent. The obtained crude extracts (331.6 g of *M. glomerata* and 564.9 g of *M. laevigata*) were employed for the isolation of chemical markers.

Mikania laevigata extract (500 g) was fractionated by silica gel column chromatography affording the *n*-hexane (6.76 g), CH₂Cl₂ (139.54 g), EtOAc (35.06 g) and MeOH (283.82 g) fractions. The *n*-hexane fraction was mainly composed by impure CO. The EtOAc fraction (35.06 g) was refractionated over a silica gel column to yield L2 (20.82 g) eluted with n-hexane/CH₂Cl₂ (7:3). In the sequence, L₂ was submitted to silica gel column chromatography to give pure CO (191.1 mg; eluent: n-hexane/CH₂Cl₂ 4:6), along with the sub-fractions L₃K (2.10 g; eluent: CH₂Cl₂/EtOAc 9:1) and L₃N (268.9 mg; eluent: CH₂Cl₂/EtOAc 8:2). Further fractionation of L₃K over a silica gel column afforded L₄F (499.7 mg; eluent: *n*-hexane/CH₂Cl₂ 4:6) and BA (404.7 mg; eluent: n-hexane/CH₂Cl₂ 2:8). The final purification of L₄F by preparative HPLC employing isocratic elution with *n*-hexane/ Et₂O (7:3) on a silica gel column afforded KA (7.8 mg). L₃N was chromatographed over a silica gel column eluted with n-hexane/Et₂O (9:1) to give CA (96.6 mg). OC was purchased from Sigma-Aldrich.

Mikania glomerata extract (5 × 32 g) was submitted to sonication with EtOAc (5 × 150 mL) for 15 min. After removing the residue by filtration, each EtOAc-soluble fraction was partitioned with saturated sodium bicarbonate solution (3 × 100 mL). The EtOAc fractions were combined and solvent was removed under reduced pressure to give a greenish residue (18.2 g). It was then dissolved in ethanol and treated several times with active charcoal, affording a whitish solid (300 mg), which was subjected to preparative HPLC on a silica gel column eluted with *n*-hexane/Et₂O (7:3) to give KA (34.8 mg).

Spectroscopic data obtained for the isolated compounds were compared with literature records (Batista et al. 2005, Kupriyanova 1997, Fabbri et al. 1997, Velandia et al. 1998, Oliveira et al. 1984, Vichnewski et al. 1977, Rowbotham and Schaefer 1972). Coumarin (CO, 191 mg): white solid, M.p. 70-72°C. Benzoylgrandifloric acid (BA, 404.7 mg): white solid, M.p. 183-186°C. $[\alpha]_D^{24} = -40^\circ$ (MeOH, c 2.0 mg/mL). Kaurenoic acid (KA, 7.8 mg from M. laevigata and 34.8 mg from M. glomerata): amorphous colorless solid. $[\alpha]_D^{24} = -100^\circ$ (MeOH, c 2.0 mg/mL). Cinnamoylgrandifloric acid (CA, 96.6 mg): white solid, M.p. 191-195°C. $[\alpha]_D^{24} = -40^\circ$ (MeOH, c 2.0 mg/mL).

DEVELOPMENT AND OPTIMIZATION OF HPLC CONDITIONS
FOR THE ANALYSIS OF CHEMICAL MARKERS

For the analysis, 1 g of dried leaves was submitted to ultrasound-assisted extraction with ethanol (2 × 30 mL), at room temperature (20 min each cycle). The extract was filtered through filter paper to a 100 mL volumetric flask and filled up with ethanol. An aliquot (10 mL) was taken and the solvent removed in a rotatory evaporator at 50°C under reduced pressure. The obtained residue was dissolved in MeOH (1 mL) and centrifuged at 8,400 g for 10 min, being the supernatant employed for analysis. During method development, fortified extracts of *Mikania laevigata* were prepared by adding OC to final concentration of 0.1 mg/mL, whereas *Mikania glomerata* extracts were spiked with OC and CO (final concentrations of 0.1 mg/mL each) before sonication.

Method development and optimization comprised the modification of parameters related to chromatographic resolution, including organic modifier, temperature, flow rate and slope of gradient elution. Phosphoric acid 0.1% was added to the eluents in all assayed conditions. Peak identification was based on UV data recorded on line by DAD, as well as on the co-injection of reference compounds. Chromatographic selectivity was

accessed by analyzing UV spectra recorded by DAD in the ascending, upper and descending regions of the peaks, being considered pure those peaks whose spectra matched exactly. The condition was considered selective when all five analyzed chemical markers showed adequate peak purity.

The efficiency of the established chromatographic condition was also assessed by determining system suitability parameters (resolution, retention factor, tailing and number of plates) of the chemical markers. The parameters were determined using Waters Empower software according to the equations recommended by the United States Pharmacopoeia (USP 2006) and the results were the mean value of 6 replicates. Determination of t_0 was accomplished by the injection of 10 μ L of sodium nitrate solution 0.01% (m/v) prepared in MeOH.

ANALYSES OF CHEMICAL MARKERS IN STORED LEAVES

Leaves of *Mikania laevigata* and *Mikania glomerata* were collected in June 2007 from the apical, intermediate and basal regions of cultivated 45-month old plants (n = 6 plants for each species). The fresh leaves were conditioned in kraft paper bags and dried in a ventilated oven at 40°C until constant weight. The paper bags were transferred into polypropylene bags and stored in a dark room with controlled temperature (25 ± 5 °C) and humidity ($54 \pm 20\%$). *M. glomerata* and *M. laevigata* samples were stored during 12 and 18 months, respectively.

Samples of the stored material were analyzed at three-month intervals. The leaves were ground in a knife mill, sieved in 0.85 mm tamis and freezer stored (-20°C) in safelock polypropylene bags until analysis. The developed chromatographic method was applied to assess changes in the chemical markers during storage of the plant material. Analysis was based on peak purity disclosed by UV spectral curves recorded on line for CO, OC, BA, CA

and KA. The contents of these compounds were quantified using a method previously described by us (Bertolucci et al. 2009).

STATISTICAL ANALYSIS

Statistical design was entirely casualized delineation, with 7 treatments for *Mikania laevigata* (T_0 , T_3 , T_6 , T_9 , T_{12} , T_{15} and T_{18} , respectively to experiment start and 3-18 months of storage) and 5 treatments for *Mikania glomerata* (T_0 , T_3 , T_6 , T_9 and T_{12}) with 6 repetitions each. The contents of constituents were submitted to ANOVA, followed by the Scott-Knott test, employing Sisvar software, version 5.0 (Ferreira 2007). Data were considered significantly different when p < 0.05.

RESULTS AND DISCUSSION

ISOLATION OF CHEMICAL MARKERS

The crude ethanolic extracts from the leaves of Mikania laevigata and Mikania glomerata were fractionated to isolate the major compounds in the species, to be employed as chemical markers. Fractionation of *M. laevigata* afforded coumarin (CO) and three ent-kaurane diterpenes, namely benzoylgrandifloric (BA), kaurenoic (KA) and cinnamoylgrandifloric (CA) acids. KA was also isolated from M. glomerata extract. Structure identification of the isolated compounds was achieved by usual spectroscopic analysis and comparison with previously reported data (Kupriyanova 1997, Oliveira et al. 1984, Rowbotham and Schaefer 1972, Batista et al. 2005, Velandia et al. 1998, Fabbri et al. 1997, Vichnewski et al. 1977), as well as by TLC and HPLC analysis employing authentic samples.

The occurrence of the isolated compounds in the leaves of *Mikania laevigata* and *Mikania glomerata* was investigated by HPLC analysis, along with *trans-o-*coumaric acid (OC), a biosynthetic precursor of coumarin. Identification of the peaks corresponding to the isolated compounds was

achieved by comparison with retention time, UV spectra and co-injection of reference compounds. Altogether, these data allowed us to propose OC, CO, KA, BA and CA as chemical markers for *M. laevigata*, whereas only kaurane-type diterpenes (KA, BA and CA) were found in *M. glomerata* leaves.

HPLC-DAD PROFILE OF *MIKANIA LAEVIGATA* ETHANOL EXTRACT

Development and optimization of the HPLC-DAD method comprised the evaluation of 14 different conditions. Mikania laevigata was selected for method development in view of its more complex profile than Mikania glomerata. A sample spiked with OC was employed for method development due to the low intensity of its peak found in the exploratory run. Acetonitrile and methanol were tested as organic modifiers; ACN exhibited better resolution for the kaurane diterpenes, whereas MeOH improved resolution for the cinnamic acid derivatives. Therefore, both solvents were employed for elution. The chromatographic exhaustive conditions were defined after adjustment in the elution strength, gradient slope, temperature and flow rate. Despite several attempts to improve resolution, elution of BA and CA was partially superimposed to other peaks (Fig 1a). Spectral data recorded by DAD for BA and CA indicated maximum wavelength absorptions at 230 and 270 nm, respectively, while the partially coeluted compounds showed maximum absorption at 210 nm. Given that all previous attempts to increase resolution for the kaurane derivatives had failed, we selected different wavelengths to register the chemical markers, respectively 210 nm for OC, CO and KA, 230 nm for BA and 270 nm for CA, thus resulting in spectral purity for each chemical marker peak. The established conditions allowed the unambiguous identification of the five constituents in the leaves of M. laevigata. Finally, we introduced a timed wavelength program for chemical markers detection, allowing registering them at the wavelength of maximum absorbance in one single chromatogram (Fig. 1b). The established chromatographic conditions are described in Table I. The reliability of the established conditions was checked by system suitability tests, comprising evaluation of resolution (Rs), tailing factor (T), number of plates (N), retention factor (t) and repeatability

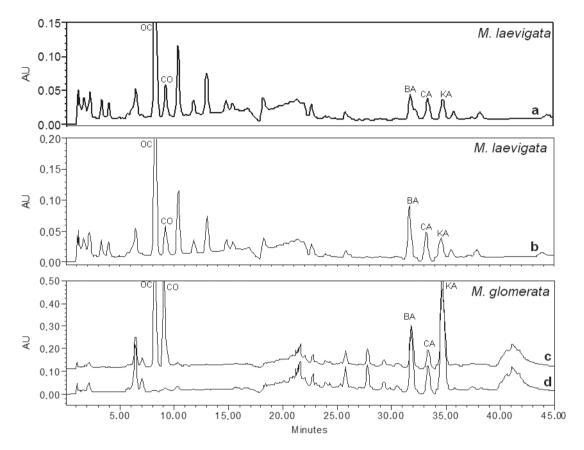


Figure 1 - HPLC profiles obtained for *Mikania laevigata* (a,b) and *Mikania glomerata* (c,d) employing the conditions described in Table I. The chromatograms were recorded using the timed wavelength program, except in "a", registered at 210 nm. The sample in "a", "b" and "c" were spiked with *o*-coumaric acid, along with coumarin in "c". Peaks: OC, *trans-o*-coumaric acid; CO, coumarin; BA, *ent*-benzoylgrandifloric acid; CA, *ent*-cinnamoylgrandifloric acid; KA, *ent*-kaurenoic acid.

of peak response [RSD of retention time (n = 6) for chemical markers' peaks]. The results are presented in Table II, along with the limits recommended by the U.S. Food and Drug Administration (FDA 2000). All values are in accordance with FDA recommendations, excepting the resolution between CO and OC. Such finding may not constitute a problem, since it is accepted that system suitability tests might be less rigorous for biological matrices and traces analysis

(Dong et al. 2001). Moreover, some authors consider Rs \sim 1.5 adequate for quantitative analysis (Meyer 1996, Dong et al. 2001). Precision is critical for analytical routine methods; variation in retention time between runs indicates low precision. In the established method, a maximum RSD for retention time was observed for CO (0.40%), attesting the high reproducibility of the chromatographic conditions, including a sufficient re-equilibration interval.

TABLE I
Chromatographic conditions established for the analysis of chemical markers in *Mikania laevigata* and *Mikania glomerata* leaves.

Time (min)	Water (%) ^a	Methanol (%) ^a	Acetonitrile (%) ^a
0	66	34	0
15	50	50	0
16	60	0	40
20	35	0	65
38	24	0	76
38.1	5	0	95
40	5	0	95
45	66	34	0
55	66	34	0

^aAll eluents were acidified with 0.1% phosphoric acid. Analyses were carried out at 25°C, at a flow rate of 1.0 mL/min, using a timed wavelength program (0-30 min, 210 nm; 31-32.4 min, 230 nm; 32.5-34 min, 270 nm and 34.1-45.0 min, 210 nm).

HPLC-DAD Profile of *Mikania Glomerata* Ethanol Extract

The HPLC conditions employed to record *Mikania laevigata* profile were applied to *Mikania glomerata* samples, disclosing kaurane-type diterpenes as major constituents (Fig. 1d). However, the occurrence of CO in the species has been previously reported (Santos et al. 2006, Oliveira et al. 1984) and therefore the chromatographic conditions were not modified, since coumarin might be present in other samples of *M. glomerata*.

Similarly to *Mikania laevigata*, system suitability tests were carried out for *Mikania glomerata* (Table II). All analyzed parameters fell within the limits established by FDA (FDA 2000), excepting the resolution for KA. In spite of that, the obtained value (Rs = 1.73) may be considered adequate for quantitative analysis (Dong et al.

TABLE II

System suitability parameters determined for the analysis of chemical markers in *Mikania laevigata* and *Mikania glomerata* leaves, employing the chromatographic conditions described in Table I.

Chemical markers	Parameters ^a						
	\mathbf{R}_{s}	Т	N	k	t _R (min)	t _R RSD	
·		Mil	kania laevigata				
OC_p	1.88	1.05	5,125	6.69	8.24	0.32	
CO	1.88	1.19	5,085	7.57	9.18	0.40	
BA	2.62	1.10	43,398	28.58	31.78	0.09	
CA	2.62	1.09	43,175	30.06	33.27	0.03	
KA	2.03	1.08	42,024	31.42	34.76	0.08	
		Mik	ania glomerata				
OC_p	1.70	1.08	5,614	6.66	8.22	0.24	
CO^a	1.70	1.39	5,055	7.43	9.05	0.31	
BA	2.51	1.12	44,944	28.68	31.82	0.05	
CA	2.51	1.09	43,920	30.06	33.27	0.03	
KA	1.73	1.19	40,122	31.36	34.69	0.05	
FDA (2000) recommended values	> 2	≤2	> 2,000	> 2.0	-	$\leq 1\%, n \geq 5$	

^aThe parameters were determined in the wavelength established for each chemical marker, according to the program describe in Table I. Rs: resolution; T: tailing factor; N: plate number; k: retention factor; t_R: retention time; RSD: relative standard deviation.
^bSamples of *Mikania laevigata* and *Mikania glomerata* spiked with 0.1 mg/mL o-coumaric acid were employed in the experiments. See text for details. CO, coumarin; OC, *trans-o*-coumaric acid; BA, *ent*-benzoylgrandifloric acid; CA, *ent*-cinnamoylgrandifloric acid; KA, *ent*-kaurenoic acid.

2001), as previously discussed for *M. laevigata*. Besides, spectral homogeneity attested peak purity for the three kaurane diterpenes, indicating method selectivity for *M. glomerata*.

Considering that the analyzed *Mikania* species present different matrices and that CO and OC may occur in other samples of *Mikania glomerata*, evaluation of method selectivity was also mandatory for this species. A sample of *M. glomerata* was spiked with OC and CO and the resulting chromatogram exhibited peak purity for both compounds (Fig. 1c). System suitability tests carried out for these chemical markers in the fortified sample of *M. glomerata* were also in accordance with FDA recommendations (FDA 2000), apart from the resolution of OC and CO, attesting method selectivity and precision (Table II).

The results obtained so far indicate a marked difference in the analyzed samples of *Mikania laevigata* and *Mikania glomerata*, with the presence of the kaurane-type derivatives (KA, BA and CA) in both and the lack of the cinnamic acid derivatives (CO and OC) in the second species. This finding corroborates the absence of coumarin previously reported for specimens of *M. glometara* collected in distinct locations of São Paulo state, Brazil (V.L.G. Rehder et al., unpublished data).

However, we cannot assure that cinnamic acid derivatives are not produced by Mikania glomerata, since both CO and OC have been already reported for the species (Santos et al. 2006, F. Bras. IV 2005, Veneziani and Oliveira 1999, Oliveira et al. 1984). The conflicting data may arise from misidentification of Mikania species based exclusively on foliar morphology, as described for the medicinal species Maytenus ilicifolia Mart. ex Reiss. and Maytenus aquifolium Mart. (Duarte and Debur 2005, Tiberti et al. 2007). The HPLC-DAD profiles here reported may contribute to overcome this problem and together with DNA fingerprints they will represent useful tools for the identification of Mikania laevigata and M. glomerata, allowing their unambiguous use for production of phytopharmaceuticals.

CHEMICAL MARKERS ANALYSIS DURING STORAGE

The HPLC profiles and quantitative composition of samples from *Mikania leavigata* and *Mikania glomerata* leaves, stored under controlled conditions, were analyzed three-monthly. CO and OC were not detected in *M. glomerata* within the evaluated period (Fig. 2, Table III), whereas CO was present in *M. laevigata* (Fig. 3, Table III). After six months of storage, peak purity analysis of CO revealed a coeluted compound, clearly detected in 12-month and 18-month stored plants (Fig. 4, CO), whose spectral data is compatible with a cinnamic acid derivative.

The CO content of 18-month stored plants $(0.124 \pm 0.026 \%)$ was statistically similar to time zero samples (0.113 \pm 0.033 %) (p < 0.05; Table III). Likewise, the concentration of the kaurane diterpenes did not vary significantly within the evaluated storage period (p < 0.05), for both species (Table III). However, spectral analysis of the peaks pointed out co-eluting compounds in samples of Mikania laevigata and Mikania glomerata stored for 6 and 3 months, respectively (data not shown). A marked decrease in resolution between CA and KA peaks ($R_S = 0.93$) was detected in *M. laevigata* after 9 months of storage in comparison to the experiment start ($R_S = 2.62$). This finding cannot be related to the loss of column efficiency, since the chromatograms obtained for non-stored plants showed similar resolution to T_0 samples (data not shown).

UV data recorded on line for BA peak showed co-eluting compounds in samples of *Mikania glomerata* stored for 3 months (Fig. 4, BA), similarly to the peak of KA in the same species (data not shown). Besides, a marked decrease in resolution was observed for both compounds. Altogether the results suggest the occurrence of chemical transformations in these derivatives. Considering that changes in the peaks of kaurane diterpenes were observed after 3 months of *M. glomerata* storage, the stability study of this species was discontinued after 12 months.

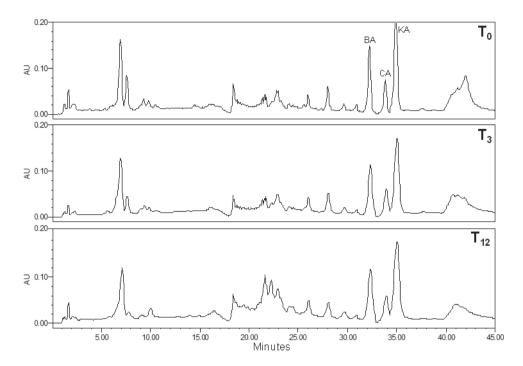


Figure 2 - HPLC profiles recorded for stored samples from *Mikania glomerata* leaves. Chromatographic condition: see Table I. Peaks: BA, *ent*-benzoylgrandifloric acid; CA, *ent*-cinnamoylgrandifloric acid; KA, *ent*-kaurenoic acid. Storage periods: T₀, study start; T₃, 3 months; T₁₂, 12 months.

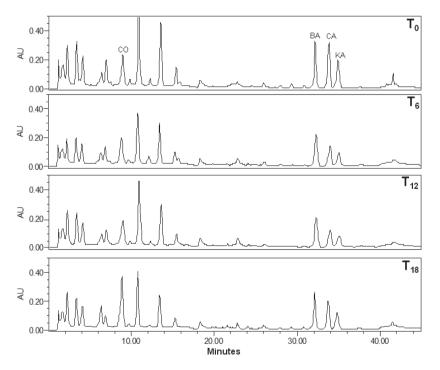


Figure 3 - HPLC profiles recorded for stored samples from *Mikania laevigata* leaves. Chromatographic condition: see Table I. Peaks: CO, coumarin; BA, *ent*-benzoylgrandifloric acid; CA, *ent*-cinnamoylgrandifloric acid; KA, *ent*-kaurenoic acid. Storage periods: T_0 , study start; T_6 , 6 months; T_{12} , 12 months; T_{18} , 18 months.

 T_{12}

54b	Content of chemical markers (% in dried leaves) ^a						
Storage period ^b	СО	OC	BA	CA	KA		
		Mikan	ia laevigata				
T ₀	0.113 ± 0.033	nd	0.276 ± 0.051	0.208 ± 0.060	0.377 ± 0.102		
T_3	0.104 ± 0.025	nd	0.288 ± 0.080	0.201 ± 0.064	0.273 ± 0.072		
T_6	0.105 ± 0.022	nd	0.311 ± 0.084	0.219 ± 0.080	0.404 ± 0.148		
Т9	0.102 ± 0.031	nd	0.292 ± 0.036	0.179 ± 0.047	0.346 ± 0.106		
T ₁₂	0.090 ± 0.031	nd	0.334 ± 0.025	0.144 ± 0.029	0.302 ± 0.054		
T ₁₅	0.108 ± 0.036	nd	0.297 ± 0.029	0.176 ± 0.044	0.322 ± 0.078		
T_{18}	0.124 ± 0.026	nd	0.313 ± 0.025	0.142 ± 0.025	0.279 ± 0.053		
		Mikani	ia glomerata				
T ₀	nd	nd	0.198 ± 0.027	0.079 ± 0.018	0.964 ± 0.160		
T_3	nd	nd	0.189 ± 0.012	0.087 ± 0.019	0.879 ± 0.140		
T_6	nd	nd	0.176 ± 0.017	0.073 ± 0.009	0.779 ± 0.109		
T ₉	nd	nd	0.174 ± 0.014	0.072 ± 0.006	0.847 ± 0.120		

TABLE III
Contents of chemical markers in stored samples from *Mikania laevigata* and *Mikania glomerata* leaves.

^aValues are mean \pm standard error obtained by triplicate analyses; nd = not detected. The mean values within each column do not differ significantly by the Scott-Knott test (p < 0.05). ^bStorage periods evaluated: T₀, study start; T₃, 3 months; T₆, 6 months; T₉, 9 months; T₁₂, 12 months; T₁₅, 15 months; T₁₈, 18 months. CO, coumarin; OC, *trans-o*-coumaric acid; BA, *ent*-benzoylgrandifloric acid; CA, *ent*-cinnamoylgrandifloric acid; KA, *ent*-kaurenoic acid.

nd

 0.180 ± 0.009

Spectral purity evaluation of selected peaks was shown to be a robust tool to access chemical stability of both *guaco* species, complementary to quantitative analysis. Excepting CA and KA peaks, whose chromatograms clearly indicated loss of resolution during the storage periods (Fig. 2 and 3), chemical changes in the other compounds were only detected by peak purity analysis, since the contents of the constituents showed no significant variation for both species (p < 0.05).

nd

In summary, our results demonstrate that the chemical integrity of the kaurane diterpenes (BA, CA and KA), constituents of *Mikania glomerata* and *Mikania laevigata*, is affected after three months of storage, whereas modifications in the coumarin peak, found only in *M. laevigata*, are detected after six months of storage. The Brazilian

Pharmacopoeia establishes a minimum content of 0.1% CO for M. laevigata (F. Bras. IV 2005), while the concentrations of kaurane diterpenes in Mikania species are not regulated by any official guide. The CO contents determined for stored samples of M. laevigata ranged from 0.10% to 0.12% and therefore fulfill the pharmacopoeical requirement, except for the 12-month sample (Table II). Hence, based strictly on the official guidelines, the analyzed samples would be considered adequate for human consumption. However, peak purity analyses of the constituents in stored samples indicate chemical changes, that may affect the biological effects of the vegetal drugs. Therefore, pharmacological analyses are required to investigate if the observed changes in chemical markers will affect the biological efficacy and safety of guaco.

 0.066 ± 0.004

 0.759 ± 0.060

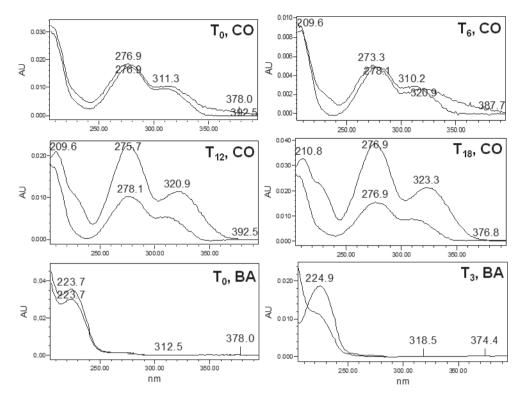


Figure 4 - UV spectra recorded on line by DAD in the ascending and descending regions of the peaks from coumarin (CO) and benzoylgrandifloric acid (BA) found in the chromatographic profiles of Figures 3 and 2, respectively. Storage periods: T_0 , study start; T_3 , 3 months; T_6 , 6 months; T_{12} , 12 months; T_{18} , 18 months.

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RESUMO

As folhas de Mikania laevigata e Mikania glomerata são usadas no Brasil para tratar afecções respiratórias, sendo diterpernos caurânicos e cumarina considerados os constituintes bioativos. O presente estudo relata a investigação dos perfis por CLAE-DAD e os teores de cumarina (1), ácidos trans-o-coumárico (2), caurenóico (3), benzoilgrandiflórico (4) e cinamoilgrandiflórico (5) em folhas secas de espécies de Mikania armazenadas em sala escura, sob condições controladas. Com exceção de 2, os constituintes foram isolados e purificados para serem empregados como substâncias de referência. As amostras foram analisadas trimestralmente, até um período de 18 meses para M. laevigata e 12 meses para M. glomerata. Ácido trans-o-coumárico não foi detectado em ambas as espécies, enquanto 1 foi encontrado somente em M. laevigata. As concentrações dos constituintes

quantificados não variaram significativamente no período avaliado (p < 0.05), para duas espécies. Por outro lado, alterações nos perfis cromatográficos e na pureza espectral dos picos relativos a $\bf 3$, $\bf 4$ e $\bf 5$ foram detectadas em amostras de ambas as espécies de *Mikania* armazenadas por três meses, enquanto o perfil do pico correspondente a cumarina mostrou-se alterado em $\bf M$. $\bf laevigata$ a partir do sexto mês de armazenamento. A análise dos perfis cromatográficos baseada na avaliação da pureza espectral dos picos selecionados mostrou-se uma ferramenta mais robusta para acessar a estabilidade química de amostras de $\bf M$ ikania submetidas a armazenamento do que a quantificação do teor dos marcadores químicos.

Palavras-chave: marcadores químicos, análises por CLAE-DAD, *Mikania laevigata*; *Mikania glomerata*, estocagem, pós-colheita.

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