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Net ion fluxes and ammonia excretion during transport of *Rhamdia quelen* juveniles

Fluxo de íons e excreção de amônia durante o transporte juvenis de *Rhamdia quelen*

Luciano de Oliveira Garcia^I Leonardo José Gil Barcellos^{II} Bernardo Baldisserotto^{III}

ABSTRACT

The objective of this study was to verify net ion fluxes and ammonia excretion in silver catfish transported in plastic bags at three different loading densities: 221, 286 and 365g L⁻¹ for 5h. A water sample was collected at the beginning and at the end of the transport for analysis of water parameters. There was a significant positive relationship between net ion effluxes and negative relationship between ammonia excretion and loading density, demonstrated by the following equations: Na⁺: $y=24.5-0.27x$, $r^2=0.99$, Cl⁻: $y=40.2-0.61x$, $r^2=0.98$, K⁺: $y=8.0-27.6x$, $r^2=0.94$; ammonia excretion: $y=-11.43+0.017x$, $r^2=0.95$, where y: net ion flux (mmol kg⁻¹ h⁻¹) or ammonia excretion (mg kg⁻¹ h⁻¹) and x: loading density (g). Therefore, the increase of loading density increases net ion loss, but reduces ammonia excretion during the transport of silver catfish, indicating the possibility of ammonia accumulation.

Key words: fish transport, ionoregulation, load density.

RESUMO

O objetivo deste estudo foi verificar o fluxo de íons e a excreção de amônia em jundiás transportados em sacos plásticos em três diferentes densidades de carga: 221, 286 e 365g L⁻¹ por 5h. Uma amostra de água foi coletada no início e ao final do transporte para as análises dos parâmetros de qualidade da água. Houve uma relação positiva significativa entre o efluxo de íons e relação negativa entre excreção de amônia e densidade de carga, demonstrada pelas seguintes equações: Na⁺: $y=-24.5-0.27x$, $r^2=0.99$, Cl⁻: $y=40.2-0.61x$, $r^2=0.98$, K⁺: $y=8.0-27.6x$, $r^2=0.94$; excreção de amônia: $y=-11.43+0.017x$, $r^2=0.95$, em que y: fluxo de íons (mmol kg⁻¹ h⁻¹) ou excreção de amônia (mg kg⁻¹ h⁻¹) e x: densidade de carga (g). Portanto, o aumento da densidade de carga aumenta a

perda de íons, mas reduz a excreção de amônia durante o transporte de jundiá, indicando possibilidade de acúmulo de amônia no corpo.

Palavras-chave: transporte de peixes, ionorregulação, densidade de carga.

INTRODUCTION

Silver catfish *Rhamdia quelen* is the native species most frequently raised in southern Brazil (BALDISSEROTTO, 2009) because is well adapted for cultivation even in the winter (GARCIA et al., 2008). Commercial fish production requests an appropriate management during transportation. Water quality and physiological responses must be observed to avoid losses during this process and reduce posterior contamination with opportunist pathogens due to the stress occasioned by this procedure (CARNEIRO et al., 2009).

Transport in plastic bags is the most used system for fish transport in Brazil (GOMES et al., 1999; GOLOMBIESKI et al., 2003; CARNEIRO et al., 2009). The high transport cost request the use of a small amount of water and the highest load density possible, which cause accumulation of metabolites and deteriorate water quality (GOLOMBIESKI et al., 2003). Ammonia is the dominant end product of nitrogen metabolism in most teleosts, including

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silver catfish (GOLOMBIESKI et al., 2013), and is toxic at low concentrations (IP & CHEW, 2010). During transport the concentration of this metabolite increases and may be a potential risk that influences fish survival and duration of transport (GOMES et al., 1999; GOLOMBIESKI et al., 2003; CARNEIRO et al., 2009).

Studies regarding silver catfish transportation analyzed the effect of salt addition in the water of transport, load density, temperature, time of transport and load density (GOMES et al., 1999; GOLOMBIESKI et al., 2003; CARNEIRO et al., 2009). The stress of transport is associated with catecholamine liberation, which increases branchial perfusion and consequently increases ion loss to the water (RANDALL & PERRY, 1992). Measurements of net ion fluxes provide a good correlation with stress of transport in *Arapaima gigas* (pirarucu) (GOMES et al., 2006), *Carnegiella strigata* (marbled hatchetfish), *Paracheirodon axelrodi* (cardinal tetra) (GOMES et al., 2008; 2009) and silver catfish (PARODI et al., 2014). Therefore, the objective of this study was to verify net ion fluxes and ammonia excretion in silver catfish during the transport in plastic bags at different load densities.

MATERIALS AND METHODS

Silver catfish juveniles (76.6 ± 0.7 g) were obtained from the fish culture sector at the Universidade de Passo Fundo, Rio Grande do Sul state, South Brazil. These juveniles were collected from earth tanks and maintained in 1000L tanks in the laboratory for two days. Feeding was discontinued 24h prior transportation. After this period, fish were placed in plastic bags (15L) with 7.7L water and 2.3L of pure oxygen. Three loading densities were used: 221, 286 and 365g L⁻¹ (three replicates each). Juveniles were transported for 5h in paved roads (294km). The loading densities tested were similar to those tested by CARNEIRO et al. (2009) and the time of transport is the most commonly observed in the transport of this species in the Rio Grande do Sul state. To analyze the transport of silver catfish in a different size range, the size used in the present study were higher than those used by GOLOMBIESKI et al. (2003) and CARNEIRO et al. (2009).

Net ion fluxes and ammonia excretion were calculated according to the equation: $J_{net} = V([ion_1] - [ion_2]) \cdot (Mt)^{-1}$, where $[ion_1]$ and $[ion_2]$ are the ion or ammonia concentrations in the water of transport

at the beginning and end of the transport period, respectively, V is the water volume (in L), M is the mass of the fish (in kg) and t is the duration of the transport (in h). Water quality parameters (hardness, alkalinity, ammonia, nitrite, temperature, dissolved oxygen and pH) in the plastic bags were monitored at the start and at the end of the transport period. Water pH was monitored with a DMPH-2 (Digimed, São Paulo, Brazil) pHmeter. Total ammonia and water hardness levels were verified following EATON et al. (2005) and un-ionized ammonia levels were calculated according to the method of COLT (2002). Dissolved oxygen and temperature were measured with a YSI oxygen meter (model Y5512). Levels of total alkalinity and nitrite were determined as described by BOYD (1998). Chloride levels were determined according to ZALL et al. (1956), and Na⁺ and K⁺ levels were determined with a B262 flame spectrophotometer (Micronal, São Paulo, Brazil). Standard solutions were made with analytical-grade reagents (Vetec or Merck) dissolved in deionized water, and standard curves of each ion to be tested were made for five different concentrations.

Data were reported as mean \pm SEM. The homogeneity of variances among groups was tested with the Levene test. Water quality parameters in all treatment groups were compared by one-way ANOVA followed by Tukey's test. The analysis of the relationships between net ion fluxes and ammonia excretion and loading density were performed using the Sigma Plot 11.0 software. All tests were performed with the Software Statistica 5.1 (1997; StatSoft Inc., Tulsa, OK, USA). The minimum significance level was set at $P < 0.05$.

RESULTS

There was a significant positive relationship between net ion fluxes, total and unionized ammonia and loading density, but negative relationship between ammonia excretion and loading density. These relationships are demonstrated by the following equations: Na⁺: $y = -24.5 - 0.27x$, $r^2 = 0.99$ (Figure 1A), Cl⁻: $y = 40.2 - 0.61x$, $r^2 = 0.98$ (Figure 1B), K⁺: $y = 8.0 - 27.6x$, $r^2 = 0.94$ (Figure 1C); ammonia excretion: $y = -6.57 - 1.25x$, $r^2 = 0.91$ (Table 1); total ammonia $y = 6.68 + 0.009x$, $r^2 = 0.92$; unionized ammonia $y = -0.82 + 0.0046x$, $r^2 = 0.86$, where y: net ion flux (mmol kg⁻¹ h⁻¹) or ammonia excretion (mg kg⁻¹ h⁻¹), total and unionized ammonia (mg L⁻¹) and x: loading density (g L⁻¹). Dissolved oxygen levels, total ammonia and un-ionized ammonia were

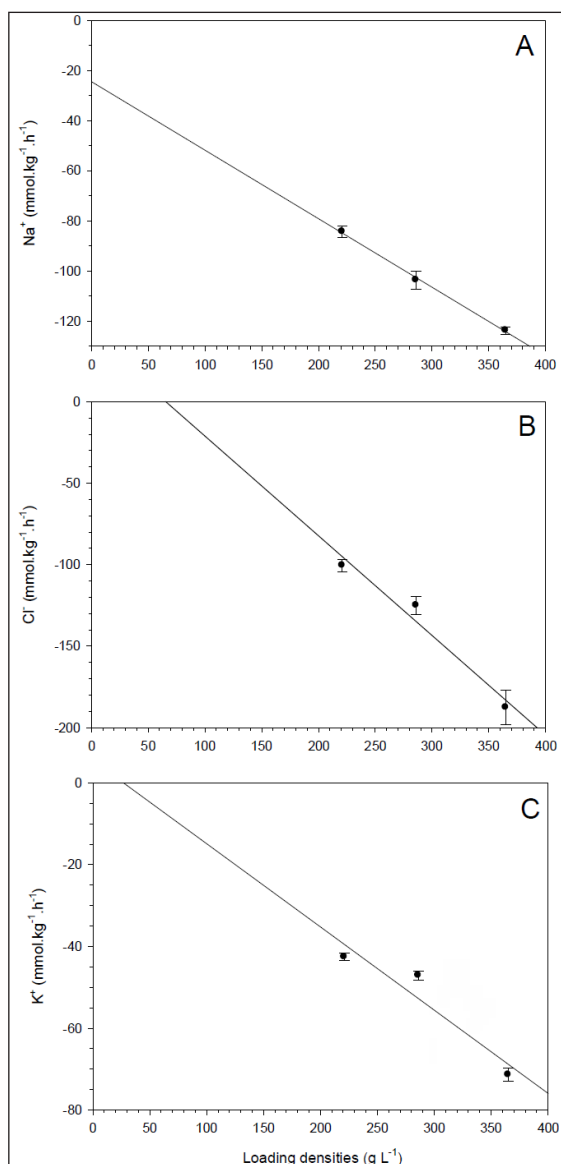


Figure 1 - Net Na⁺ (A), Cl⁻ (B), and K⁺ (C) fluxes in silver catfish juveniles transported for 5h in plastic bags at different loading densities. Values are expressed as means \pm SEM. Data are fitted to the following equations: A: $y = -24.5 - 0.27x$; $r^2 = 0.99$; B: $y = 40.2 - 0.61x$; $r^2 = 0.98$; C: $y = 8.0 - 0.27x$; $r^2 = 0.96$.

significantly higher and pH lower at the end of the transport. The other water parameters were not affected significantly by transportation (Table 2).

DISCUSSION

Transporting fish is very important for fish culture and monitoring water quality is essential to the success of this operation. This situation cause

stress, which increases cortisol levels in silver catfish (BARCELLOS et al., 2001; 2003; 2004) and stimulates glycogenesis, gluconeogenesis, protein catabolism and ammonia production (RANDALL & TSUI, 2002; IP & CHEW, 2010). Silver catfish net Na⁺, K⁺ and Cl⁻ effluxes through the transport increased with the increase of loading density, demonstrating that high loading density increases ionoregulatory stress in this species. Loss of these ions was also demonstrated in previous studies about silver catfish transportation (BECKER et al., 2012; 2013; PARODI et al., 2014; ZEPPENFELD et al., 2014) and is a good parameter to indicate stress in freshwater fish because stress increases gill ventilation rate, which increased ion branchial efflux (McDONALD et al., 1991).

In the present study the ammonia levels in the water of transport increased due to excretion via gills. Unionized ammonia values observed even at the highest load density at the end of experiment were very low because water pH reduced significantly and the percentage of unionized ammonia decreased at acidic pH (COLT, 2002). Ammonia excretion values through the 5h transport of silver catfish juveniles decreased with the increase of the loading density. Ammonia excretion values also decrease with the increase of loading density (75-350 g L⁻¹) in silver catfish juveniles (23-28g) transported for 5h (24.6°C) (CARNEIRO et al., 2009). This decrease on ammonia excretion may have occurred because the increase of water

Table 1 - Ammonia excretion in silver catfish as a function of load density and transport time. The relationships are described by the following equation: $y = -6.57 - 1.25x$, $r^2 = 0.91$ (present study); where y = ammonia excretion (mg kg⁻¹ h⁻¹) and x = load density (g L⁻¹).

Load density (g L ⁻¹)	Weight fish ⁻¹ (g)	Ammonia excretion (mg kg ⁻¹ h ⁻¹)
100 ¹	5.8	-34.72
75 ²	23.5	-17.25
150 ²	23.5	-15.75
250 ²	23.5	-16.39
350 ²	23.5	-14.56
221 ³	76.6	-7.99
286 ³	76.6	-6.24
365 ³	76.6	-5.55
162 ⁴	3.0	-17.85
169 ⁵	301.2	-42.60

¹GOMES et al., 1999; ² CARNEIRO et al., 2009; ³GARCIA et al. (present study); ⁴PARODI et al., in press; ⁵BECKER et al., 2012.

Table 2 - Physicochemical parameters of transport water in the experimental plastic bags before and after transport simulation at different load density. Data are reported as mean \pm SEM. Different letters indicate significant differences between different load density in total or unionized ammonia levels ($P < 0.05$).

Physicochemical Parameters	Initial values	-----The end of transport simulation (load density in g L ⁻¹) -----		
		221	286	365
pH	6.8 \pm 0.2 ^a	5.6 \pm 0.4 ^b	5.4 \pm 0.1 ^b	5.1 \pm 0.2 ^b
Dissolved oxygen (mg L ⁻¹)	6.1 \pm 0.3 ^a	11.4 \pm 0.9 ^b	10.0 \pm 0.7 ^b	8.6 \pm 0.5 ^c
Temperature (°C)	23.5 \pm 0.2	25.0 \pm 2.5	24.6 \pm 0.4	23.8 \pm 0.2
Hardness (mg CaCO ₃ L ⁻¹)	17.8 \pm 0.1	18.5 \pm 3.8	19.1 \pm 0.5	19.9 \pm 0.3
Alkalinity (mg CaCO ₃ L ⁻¹)	14.5 \pm 0.1	12.0 \pm 2.0	11.8 \pm 0.2	11.5 \pm 0.1
Nitrite (mg L ⁻¹)	0.001 \pm 0.001	0.005 \pm 0.002	0.005 \pm 0.004	0.005 \pm 0.002
Total ammonia (mg L ⁻¹)	0.07 \pm 0.01 ^a	8.9 \pm 1.7 ^b	9.0 \pm 1.1 ^b	10.2 \pm 0.8 ^b
Unionized ammonia (µg L ⁻¹)	0.22 \pm 0.03 ^a	2.01 \pm 0.38 ^b	1.25 \pm 0.15 ^c	0.67 \pm 0.05 ^c

ammonia levels decreases the plasma-water ammonia gradient next to the external surface of the gills, which impairs ammonia excretion and can increase ammonia accumulation (RANDALL & TSUI, 2002; WRIGHT & WOOD, 2012).

CONCLUSION

Silver catfish can be transported for 5h with a loading density up to 365g L⁻¹, but the use of lower densities reduces body ion loss and probably ammonia accumulation.

BIOETHICS AND BIOSSECURITY COMMITTEE

The methodology of this experiment was approved by the Ethical and Animal Welfare Committee of the Universidade Federal de Santa Maria (Process n. 046/2010).

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