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# Synthesis of Thulium-yttria Nanoparticles with EPR Response

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### **ABSTRACT**

Approaches to form new materials for radiation dosimetry are essential to enhance quality assurance and quality improvement practices based on radiation protection concept. The present work reports a hydrothermal synthesis based on a relative low temperature and pressure to form thulium-yttria nanoparticles with electron paramagnetic resonance response. Thulium-yttria nanoparticles were prepared and characterized by XRD, SEM, PCS, and EPR. According to results, the hydrothermal method provided thulium-yttria nanoparticles with cubic C-type structure, mean particle size (d50) less than 160nm, and EPR response. The EPR spectra of powders exhibited two resonance peaks p1 and p2 recorded at 350 and 160mT, respectively. The enhancement of the EPR response of yttria by the use of thulium as a dopant provide meaningful parameters to advance in the formation of new rare earth-based materials for radiation dosimetry.

Keywords: yttria, thulium oxide, EPR, radiation dosimetry, ceramic processing.

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#### 1. INTRODUCTION

The design and formation of new materials for radiation dosimetry are essential to improve practices where ionizing radiation is used [1-3]. The hydrothermal synthesis [4] offers the possibility to build up smart structures from colloidal suspensions, whereby the formation of particles is based on controlled chemical stoichiometry. Innovative contributions using this method have been reported elsewhere [5]–[10].

The rare earths (RE) exhibit unique chemical and physical properties, being addressed as critical materials by The European Union and United States of America.[11], [12] Belong to the RE group, yttria (Y<sub>2</sub>O<sub>3</sub>) has been used as a host material of other REs due to its intrinsic lattice characteristics that enable the insertion of other RE ions into its structure (doping), providing the development of advanced structures and useful applications in transparent ceramics[13], lasers[14], biomaterials[15], and energy[16]. Oliveira et al. [17] obtained multi-colour-emitting, high-crystalline and micro-sized Y<sub>2</sub>O<sub>3</sub>:Eu<sup>3+</sup>,Er<sup>3+</sup>,Yb<sup>3+</sup> particles using sorbitol as an alternative to enhance the luminescence of Pechini-derived phosphors. In our recent study [18], it was found out that the dosimetric characteristics of yttria were improved by using 2at.%Eu as dopant. A linear EPR doseresponse behaviour was recorded in a range of dose from 0.001 to 50kGy.

The effect of thulium (Tm) as dopant/activator of ceramic/glasses materials has been reported elsewhere [19]–[21]. The motivation of selecting Tm as dopant of yttria consists in the fact that the electronic levels of Tm ions can interact with the energy levels of yttria, and an improvement of yttria characteristics for radiation dosimetry may be achieved.

The present paper reports the formation of thulium-yttria nanoparticles with controlled particle characteristics and EPR response by an alternative hydrothermal synthesis. The results suggest that the thulium-yttria nanoparticles are promising materials for radiation dosimetry.

### 2. MATERIALS AND METHODS

Thulium-yttria powders (Y<sub>2</sub>O<sub>3</sub>:Tm) with controlled characteristics such as stoichiometry, shape, size, and density were synthesized using the following starting materials: yttria (Y<sub>2</sub>O<sub>3</sub>, 99.99%, Alfa

Aesar GmbH), thulium oxide (Tm<sub>2</sub>O<sub>3</sub>, 99.999%, Alfa Aesar GmbH), nitric acid (HNO<sub>3</sub>, Synth), ammonium hydroxide (NH<sub>4</sub>OH, Casa Americana).

A facile hydrothermal process [22] was performed to synthesize Y<sub>2</sub>O<sub>3</sub>:Tm powders, with thulium content varying from 0.1 up to 2.0at% (at.%, atomic percentage). All powder compositions were formed by stoichiometry calculations considering yttria and thulium oxide. The precursor powders were obtained from a stock solution processed at 60°C for 6h in a condenser system.

The as synthesized  $Y_2O_3$ :Tm powders were characterized as follows: mean particle size ( $d_{50}$ ) by Photon Correlation Spectroscopy (PCS, Litesizer500, Anton Paars), using hydrodynamic diameter model [23]X-ray diffraction (XRD, Rigaku Multiflex, Japan), with an angular range ( $2\theta$ ) from 15 to 70°, scanning of  $0.5^{\circ}$ .min<sup>-1</sup> and K $\alpha$  source, in which the crystallite size was calculated by Scherrer model[24], and based on the measurement of full-width at half-maximum (FWHM) values in the corresponding XRD pattern; helium pycnometry (Pycnometer Micrometrics 1330), and Scanning Electron Microscopy (SEM, INC Ax-act, Oxford Instruments).

The paramagnetic response of Y<sub>2</sub>O<sub>3</sub>:Tm powders as a function of dopant concentration (Tm<sup>3+</sup>) was evaluated by electron paramagnetic resonance using an X-band EPR spectrometer (Bruker EMX PLUS), under room temperature and atmosphere. EPR spectra of samples were recorded in field modulation frequency of 100kHz, microwave power of 2.5mW, center field of 300mT, sweep width of 300mT, modulation amplitude of 0.4mT, time constant of 0.01ms, 10 scans, temperature of 20°C, environmental atmosphere, at controlled humidity, and using DPPH (2,2-Diphenyl-1-picrylhydrazyl, Bruker) as EPR reference.

## 3. RESULTS AND DISCUSSION

XRD curves of thulium-yttria powders with up to 2at.%Tm are shown in Fig.1. Based on the results all samples exhibited cubic C-type structure, without secondary phases, corresponding to JCPDS 25-1200. The presence of thulium ions into yttria structure does not change considerably the lattice arrangement, because yttrium and thulium present similar ionic radius. Thus, the doping is characterized as substitutional, and no lattice distortion is induced. These results reveal that the hydrothermal synthesis successfully produced thulium-yttria powders with controlled crystalline structure.

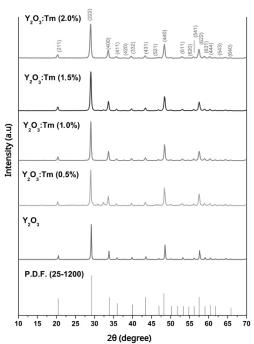


Figure 1. XRD curves of Y<sub>2</sub>O<sub>3</sub>:Tm powders synthesized by hydrothermal synthesis, followed by thermal treatment at 1100°C for 2h in room atmosphere.

Particle characteristics of thulium-yttria powders up to 2at.%Tm are illustrated in Fig.2. The variation of the mean particle size ( $d_{50}$ ) as a function of Tm content is illustrated in Fig.2a. According to results the increase of Tm content provided an increase of the mean particle from 50 to 150nm. Besides, SEM image illustrated in Fig. 2b revealed that thulium-yttria powders (0.5at% Tm) are constituted by fine particles, with rounded morphology and size smaller than 1 $\mu$ m. Moreover, crystallite size ( $d_c$ ) represented in Fig. 2c exhibited a small variation (between 5 and 6.8nm), which confirms the "substitution characteristic" of doping. The control of particle characteristics contributes directly with the whole ceramic processing as dispersion, shaping, and sintering. By this way, ceramic bodies with smart end-use can be formed. On the other hand, the pycnometric density (Fig.2d) increased significantly from 4.6 ("pure yttria", 0at.%Tm) to  $5.8g.cm^{-3}$  (2at.%Tm)

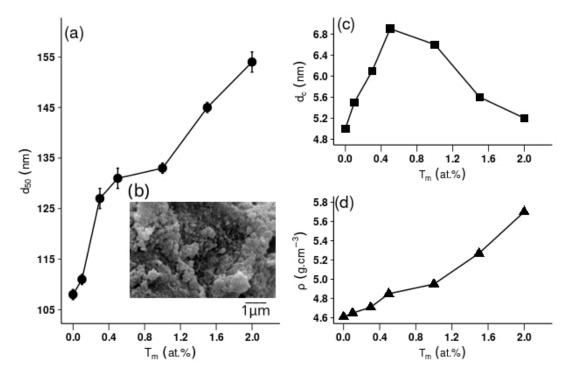


Figure 2. Thulium-yttria powder characteristics: (a) mean particle size ( $d_{50}$ ); (b) SEM image of composition prepared with 0.5at.%Tm; (c) crystallite size ( $d_{c}$ ); and (d) pycnometric density ( $\rho$ ).

Processing steps as synthesis, washing, grinding, as well as radiation provide the development of unpaired electrons in materials. The Electron Paramagnetic Resonance (EPR) is a non-destructive and non-intrusive technique that provides structural information from chemical and physical processes by detection of unpaired electrons [25]. Thus, EPR is a useful technique to advance in development of rare earth-based dosimetry materials.

The EPR spectra of thulium-yttria powders prepared with up to 2at.%Tm are illustrated in Fig.3. The major peak p1 is associated to interstitial ion O<sup>2-</sup> from adsorption of oxygen from atmosphere [26], and was recorded at 350mT, with a width of 3mT, and g-value 2.0040. Based on yttria spectra, a shift around of 0.0040 on g-value was observed. The addition of Tm<sup>3+</sup> led the formation of an additional resonance peak p2, which is associated to F<sup>+</sup> center i.e oxygen vacancies containing a remaining electron, recorded at 164.0mT and g-value 2.0065. Moreover, the p1 peak exhibited remarkable increase for composition 1.5at% Tm compared with "pure" yttria powders.

The insertion of low concentrations of Tm provided substantial formation of crystal defects, which in turn can trap more unpaired electrons formed during materials processing, including

irradiation. In addition, the best results observed for the samples prepared with 1.5%Tm can be explained considering the crystalline features of yttria. According to Zych et al. [27], the solid-state behavior of yttria is associated with the symmetry axis where the RE ions (dopant) are located. As RE ion are located at  $C_{3i}$ ,  $S_6$  symmetry axes, few electron transitions occur and the luminescence response is low. On the other hand, as RE ions fill  $C_2$  symmetry, transitions  ${}^5D_0$  -  ${}^7F_2$  occur and the luminescence response is high.

Based on the results, doping yttria with thulium enhances the EPR response of samples due to the formation of new defects, as well as energy transfer from Tm<sup>3+</sup> ion to yttria host.

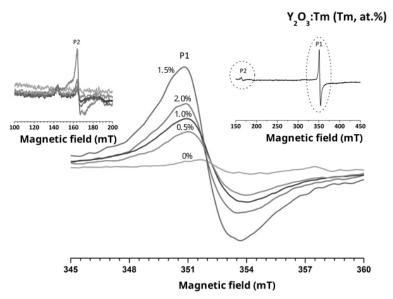


Figure 3. EPR curves of as synthesized powders of thulium-yttria recorded in environmental temperature.

### 4. CONCLUSIONS

Cubic C-type thulium-yttria nanoparticles with mean particle size less than 160nm, and electron paramagnetic resonance (EPR) response were obtained by hydrothermal synthesis, followed by thermal treatment at 1100°C for 2h in environmental atmosphere. Among all compositions evaluated, thulium-yttria nanoparticles prepared with 1.5at.%Tm presented the most intense EPR

response. The present synthesis was successful to produce rare-earth nanoparticles with promising characteristics for radiation dosimetry.

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